

JEE Main 2024 Chemistry Question Paper Jan 30 Shift 2

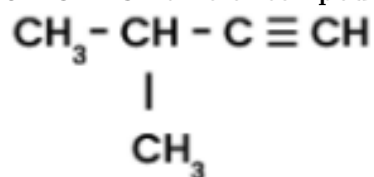
1. Why KMnO_4 shows colour?

- (1) Due to d-d transition
 - (2) Due to metal to ligand charge transfer
 - (3) Due to ligand to metal charge transfer
 - (4) Due to F-center
-

2. C is added to solution of A and B, find mole fraction of C.

- (1) $\frac{n_c}{(n_A+n_B+n_C)}$
 - (2) $\frac{n_c}{(n_A+n_B)}$
 - (3) $\frac{n_c}{(n_A+n_B+n_C)}$
 - (4) $\frac{n_c}{(n_A+n_B)}$
-

3. IUPAC name of compound:



- (1) 2-Methylbutyne
 - (2) 3-Methylbut-1-yne
 - (3) 2-Methylbutene
 - (4) 3-Methylbutane
-

4. Which reagent on reacting with phenol gives salicylaldehyde?

- (1) CO_2, NaOH
 - (2) $\text{CHCl}_3, \text{NaOH}$
 - (3) $\text{CCl}_4, \text{NaOH}$
 - (4) $\text{H}_2\text{O}, \text{H}^+$
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5. Which of the following has a square pyramidal shape?

- (1) ClF_3
- (2) BrF_5
- (3) XeF_4

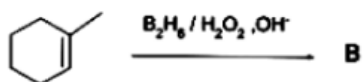
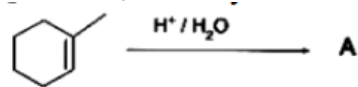
(4) NH_3

6. Statement I: NH has lower dipole moment than NF.

Statement II: In NF, the flow of electron is in the same direction.

- (1) Both statement I and statement II are false
 - (2) Statement I is true but statement II is false
 - (3) Statement I is false but statement II is true
 - (4) Both statement I and statement II are true
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7. Identify A and B.

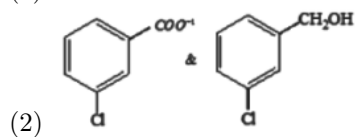
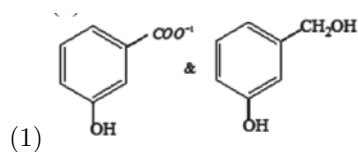


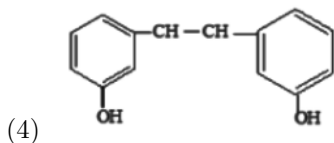
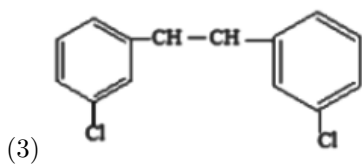
- (1) OH
 - (2) OH
 - (3) OH
 - (4) OH
-

8. What happens when phenol is treated with chloroform in the presence of NaOH at 343 K followed by hydrolysis?

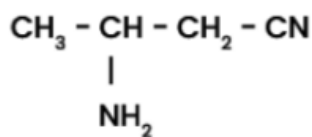
- (1) Salicylic Acid
 - (2) Salicylaldehyde
 - (3) Benzaldehyde
 - (4) Benzoic acid
-

9. When m-chlorobenzaldehyde is treated with 50% KOH solution, the product(s) obtained is:



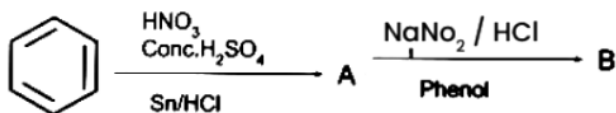


10. Correct IUPAC name the given compound is:



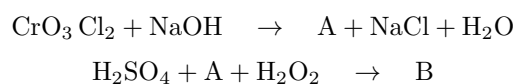
- (1) 3-aminobutanenitrile
- (2) 3-amino cyano butane
- (3) 2-aminobutanenitrile
- (4) 1-aminobutanenitrile

11. Find B:



- (1) p-hydroxy azobenzene
- (2) o-hydroxy azobenzene
- (3) m-hydroxy azobenzene
- (4) Azodye

12. In the given reaction A and B respectively are:



- (1) Na_2CrO_4 and CrO_5
- (2) CrO_3 and Na_2CrO_3
- (3) CrO_4 and $\text{Na}_2\text{Cr}_2\text{O}_7$
- (4) $\text{Na}_2\text{Cr}_2\text{O}_7$ and Na_2CrO_4

13. _____ is based on the difference in the solubility of different components of a mixture with a solvent.

- (1) Filtration
 - (2) Sublimation
 - (3) Crystallization
 - (4) Chromatography
-

14. What is the structure of $Mn_2(CO)_{10}$?

- (1) Two square pyramidal units joined by bridging CO ligands
 - (2) Two square pyramidal units joined by Mn-Mn bond
 - (3) Two tetrahedral units joined by Mn-Mn bond
 - (4) Two square planer units joined by Mn-Mn bond
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15. Statement I: H_2Te is more acidic than H_2S

Statement II: H_2Te has more BDE than H_2S

- (1) Statement I and II both are correct
 - (2) Statement I and II both are incorrect
 - (3) Statement I is incorrect and Statement II is correct
 - (4) Statement I is correct and Statement II is incorrect
-

16. Total optically active compound will be shown by these compound:

17. Number of spectral lines in the He^+ for transition from $n = 5$ to $n = 1$:
