

KEAM 2026 Engineering April 17

Question Paper (Memory-Based)

Conducted by CEE Kerala



General Instructions

- (i) **Duration:** The total duration of the examination is 3 hours (180 minutes).
- (ii) **Total Marks:** The complete paper carries a maximum of 600 marks.
- (iii) **Structure:** The paper has 3 Sections:
 - **Section A:** 45 Multiple Choice Questions (Physics).
 - **Section B:** 30 Multiple Choice Questions (Chemistry).
 - **Section B:** 75 Multiple Choice Questions (Mathematics).
- (iv) **Compulsory Questions:** All 150 questions are compulsory.
- (v) Each question has four options. Only **one** option is correct.
- (vi) **Correct Answer:** +4 marks.
- (vii) **Incorrect Answer:** -1 (Negative marking).
- (viii) **Unanswered/Marked for Review:** 0 marks.

Physics

1. In YDSE, when light of wavelength 700nm is used, a fringe width of 0.5mm is obtained. What happens when light of wavelength 500nm is used?

2. When light containing photon of energy $2h\nu_0$ falls on a metal of work function $h\nu_0$, electrons of velocity v_1 are ejected. When photons of energy $5h\nu_0$ is incident, velocity of electrons ejected

is v_2 . What is the ratio $\frac{v_1}{v_2}$?

3. Consider a convex lens made of material of refractive index $n = \frac{3}{2}$ and radius of curvature R . What is the relation between focal length and radius?

- (A) $f = \frac{R}{2}$
 - (B) $f = \frac{R}{4}$
 - (C) $f = \frac{R}{3}$
 - (D) $f = \frac{2R}{3}$
-

4. Kinetic energy of a particle of mass 1×10^{31} kg and wavelength 63 nm (where $h = 6.3 \times 10^{-34}$ Js)

- (A) 1.56×10^{-3} J
 - (B) 1.34×10^{-3} J
 - (C) 1.00×10^{-3} J
 - (D) 2.46×10^{-3} J
-

5. Dimension of Planck's constant is same as that of:

- (A) Energy
 - (B) Linear momentum
 - (C) Angular momentum
 - (D) Force
-

6. What is the number of significant figures in 420.00040×10^{-3} ?

7. Two metallic spheres of radii 1:2 are connected by a conducting wire. What is the ratio of electric field intensities at their surface?

8. What is the ratio of maximum height attained to the height attained at $t = 1$ s for a projectile of initial velocity u projected at an angle 30° with horizontal?

9. If $r_1 = \frac{c_p}{c_v}$ of a rigid diatomic gas and $r_2 = \frac{c_p}{c_v}$ of a non-rigid diatomic gas, find r_1 and r_2 .

10. If a body travels half of the total distance with velocity of 20 km/hr and other half with velocity of 30 km/hr, find average velocity.

11. If an open pipe suddenly closed, the frequency of the third harmonic of the closed pipe is 50Hz more than the fundamental frequency of the open pipe. Find the fundamental frequency.

12. Given mass of neutron $1.0087u$, mass of proton $1.0073u$, mass of ${}^4\text{He} = 4.0018u$, find the binding energy of He.

- (A) 27.8 MeV
 - (B) 28.1 MeV
 - (C) 29.5 MeV
 - (D) 30.2 MeV
-

13. If I , E and L are the moment of inertia, rotational kinetic energy, and angular momentum respectively, then:

- (A) $I = \frac{E}{L}$
 - (B) $2E = \frac{L}{I}$
 - (C) $L = \sqrt{2EI}$
 - (D) $E = L = \frac{L}{I}$
-

14. Kirchhoff's first and second laws are consequence of conservation of — and — respectively:

- (A) Energy and Charge
 - (B) Charge and Energy
 - (C) Angular momentum and energy of capacitance C
-

15. Two identical capacitors of capacitance C are connected in series. If the space between the plates of one of the capacitors is filled with a medium of dielectric constant k , what is the effective capacitance?

16. If the ratio of escape velocities is 3:2 from two different planets A and B of radii in the ratio 2:3, find the ratio of acceleration due to gravity at the surface of A to that at the surface of B.

17. Find the velocity of wave given by $y = 0.05 \sin\left(\frac{2\pi}{\lambda}(x - 200t)\right)$.

18. If work function of a metal is 6.6 eV, find the threshold wavelength. Given $h = 6.6 \times 10^{-34} \text{ J}\cdot\text{s}$.

19. A circular loop is made from a wire of length 6m. If 2A current passes through the circular loop, what is the magnetic moment of the loop?

20. A galvanometer of 500 resistance is shunted such that only 4% of the current passes through the galvanometer. Find the shunt resistance.

- (A) 12.5
- (B) 15
- (C) 18

(D) 20

21. If a body of mass 5 kg has a linear momentum of 4 kgm/s, find the kinetic energy.

- (A) 4J
 - (B) 8J
 - (C) 16J
 - (D) 32J
-

22. Find the relation between the wavelength of proton and electron, if both particles have the same kinetic energy.

23. A particle of charge equal to 10 times the charge of electrons revolves in a circle with frequency equal to 10 revolutions per second. Find the magnetic field at the centre of the circular path.

24. If $I = 2\text{A}$, $\phi = 10^{-2}\text{Weber}$, $N = 1000$, calculate the self-inductance.

25. What should be connected in the circuit to remove ripples in AC?

26. What should be connected in the circuit to remove ripples in AC:

- (A) Capacitor in series with load resistance
 - (B) Capacitor in parallel with load resistance
 - (C) Inductor connected in parallel with load resistance
 - (D) Inductor connected in series with load resistance
-

27. Bernoulli's principle is applicable for:

- (A) Non-compressible non-viscous fluid having stream line flow
 - (B) Non-compressible non-viscous fluid having turbulent flow
 - (C) Compressible viscous fluid having stream line flow
 - (D) Compressible viscous fluid having turbulent flow
-

28. If power = 150 kW, torque = 100 Nm, find the angular velocity ω .

29. What is the ratio of the longest wavelength in Lyman and Balmer series?

30. A Carnot engine is working between 400K and 500K. If the output work is 1 kJ, what is the heat absorbed?

- (A) 2 kJ
 - (B) 3 kJ
 - (C) 4 kJ
 - (D) 5 kJ
-

31. At a certain height h from the surface of Earth, the value of acceleration due to gravity is $\frac{g}{9}$, where g is the acceleration due to gravity at the surface. What is the value of h in terms of the radius of Earth R ?

- (A) $\frac{R}{2}$
 - (B) $\frac{R}{3}$
 - (C) $2R$
 - (D) $3R$
-

32. If a proton is displaced by 5m in an electric field of 50 N/C, what is the work done by the electric field?

33. A particle of charge equal to 10 times the charge of an electron revolves in a circle with frequency equal to 10 revolutions per second. Find the magnetic field at the centre of the circular path.

34. Given $I = 2A$, $\phi = 10^{-2}$ weber, $N = 1000$, calculate the self-inductance.

- (A) 0.5 H
 - (B) 1.0 H
 - (C) 1.5 H
 - (D) 2.0 H
-

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-

37. If $I = 16A$, electron density $n = 5 \times 10^{23} \text{ m}^{-3}$, and $A = 1 \times 10^{-7} \text{ m}^2$, find the drift velocity.

38. Transformer core is laminated because:

- (A) To reduce eddy current loss
 - (B) To reduce hysteresis loss
 - (C) To reduce copper loss
 - (D) To reduce core loss
-

39. For an elastic collision

- (A) both momentum and K.E is conserved
 - (B) only K.E is conserved
 - (C) only momentum is conserved
 - (D) neither momentum nor K.E conserved
-

40. Current in a circuit is 0.6 A when an external resistance of 3Ω is connected. When the external resistance is changed to 6Ω , current in the circuit becomes 0.4 A. Find the internal resistance of the cell.

41. Which of the following statement is correct for EM wave:

- (A) Velocity in vacuum is 3×10^6 cm/s
 - (B) They can travel in vacuum
 - (C) Energy density of electric field and magnetic field are different
 - (D) It contains electric field vibration only
 - (E) It contains magnetic field vibration only
-

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Chemistry

1. What is the IUPAC name of Mesityl oxide?

- (A) 4-methylpent-3-en-2-one
 - (B) 2,4,6-Trimethylphenylacetone
 - (C) 2,4,6-Trimethyl-3-penten-2-one
 - (D) 2,4,6-Trimethyl-3-hexen-2-one
-

2. IUPAC name of Element number 105?

3. What is 1-chlorocyclohexene?

- (A) Vinylic Halide
 - (B) Benzylic Halide
 - (C) Allylic Halide
-

4. Decreasing order basic strength

- (A) $\text{NH}_3 > \text{CH}_3\text{NH}_2 > (\text{CH}_3)_2\text{NH} > (\text{CH}_3)_3\text{N}$
 - (B) $(\text{CH}_3)\text{NH}_2 > \text{CH}_3\text{NH}_2 > (\text{CH}_3)_2\text{NH} > \text{NH}_3$
-

(C) $\text{CH}_3\text{NH}_2 > (\text{CH}_3)_2\text{NH} > (\text{CH}_3)\text{NH}_2 > \text{NH}_3$

5. The KE of particle of mass 1×10^{-31} Kg and the de Broglie wavelength 63 nm ($h = 6.3 \times 10^{-34}$)

6. Which of the following have minimum and maximum threshold energy K, Na, Mg, Li?

7. Number of CC, CH, C=C in But-2-ene-1-yne respectively

8. There is 40 % C, 67% H2, find the empirical formula

9. Energy of $2h\nu_0$ fall on a metal of work function $h\nu_0$ cause velocity of v_1 , when $5h\nu_0$ fall velocity ratio of v_1/v_2

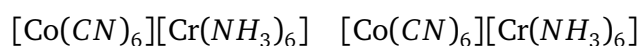
(A) $v_1/v_2 = 1/5$

(B) $v_1/v_2 = 5/1$

(C) $v_1/v_2 = 1/25$

(D) $v_1/v_2 = 25/1$

10. The isomerism shown by the following compound:



(A) Ionisation isomerism

(B) Coordinate isomerism

(C) Linkage isomerism

(D) Hydrate isomerism



Oxidation state of phosphorus change from — to — respectively.

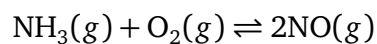
12. Which 3d series of element has least enthalpy of atomization?

- (A) Sc
 - (B) Mn
 - (C) V
 - (D) Cu
 - (E) Zn
-

13. Ratio of between the maximum wavelength of Lyman and Balmer series.

- (A) $\frac{5}{27}$
 - (B) $\frac{27}{5}$
 - (C) $\frac{4}{3}$
 - (D) $\frac{36}{5}$
 - (E) $\frac{5}{36}$
-

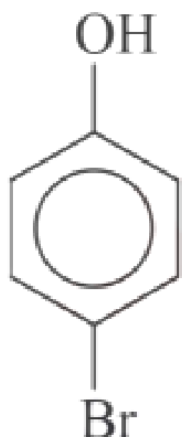
14. The K_c of the reaction



at 1500 K is 0.1. What is the concentration of NO, when the initial concentration of N_2 and O_2 is 0.04 mol?

- (A) 1.09×10^{-2} M
 - (B) 10.9×10^{-2} M
 - (C) 2.18×10^{-2} M
 - (D) 1.09×10^{-4} M
 - (E) 2.18×10^{-4} M
-

15. Which reagent gives as major product from phenol



- (A) $\text{Br}_2 + \text{CS}_2$, at 273 K
- (B) $\text{Br}_2 + \text{heat}$
- (C) Bromine water
- (D) $\text{Br}_2 + \text{CCl}_4$ at 273 K
- (E) $\text{Br}_2 + \text{acetone}$ at 273 K

16. Hinsberg reagent is

- (A) p-toluene sulphonyl chloride
- (B) Benzene sulphonyl chloride
- (C) Phthalimide and KOH
- (D) Anhydrous ZnCl_2 and conc HCl
- (E) Benzoyl chloride and NaOH

17. If 'm' is the molality, 'M' is the molarity, 'd' is the density in g/cm^3 and 'M₂' is the molarity of solute. What is the relation between them?

18. Which of the following has highest pK_a value?

- (A) CH_3COOH
- (B) $\text{F-CH}_2\text{-COOH}$

- (C) $\text{CN-CH}_2\text{-COOH}$
 - (D) $\text{Cl-CH}_2\text{-COOH}$
 - (E) $\text{NO}_2\text{-CH}_2\text{-COOH}$
-

19. Which of the following statements regarding the structure of CO_2 is correct?

- (A) CO_2 contains 1 C=O and 1 C=O and one lone pair in each oxygen
 - (B) CO_2 contains 2 C=O , and 2 lone pairs in each oxygen
 - (C) CO_2 contains 2 C=O , and 2 lone pairs in each oxygen
 - (D) CO_2 contains 1 C=O , and 1 C=O and two lone pairs in each oxygen atom
 - (E) None of these
-

20. Which of the following has a planar structure with two lone pairs?

- (A) XeF_4
 - (B) NiF_4
 - (C) SF_4
 - (D) SF_6
 - (E) XeF_4
-

21. Lactose is composed of:

- (A) α -D glucose and β -D fructose
 - (B) β -D glucose and β -D galactose
 - (C) 2 units of α -D glucose
 - (D) 2 units of β -D glucose
 - (E) α -D glucose and β -D galactose
-

22. Which of the following is allylic alcohol?

- (A) $\text{C}_6\text{H}_5\text{CH}_2\text{OH}$
(B) $\text{CH}_2 = \text{CH} - \text{C}(\text{CH}_3)_2\text{OH}$
(C) $\text{CH}_2 - \text{CH}_2 - \text{CH}_2\text{OH}$
(D) $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{OH}$
-

23. Relationship between t_{90} and t_{99} for a first order reaction:

- (A) $t_{99} = 3t_{90}$
(B) $t_{99} = 2t_{90}$
(C) $t_{99} = 2.303t_{90}$
(D) $t_{99} = 20.693t_{90}$
(E) $t_{99} = 6.93t_{90}$
-

24. Enthalpy of formation of C_6H_6 , $\text{CO}_2(g)$ and $\text{H}_2\text{O}(l)$ are -393.5, -285.8 and +48.5 KJ/mol respectively. Find the enthalpy of combustion of C_6H_6 :

- (A) 3267.4 KJ/mol
(B) 3218.49 KJ/mol
(C) 857.5 KJ/mol
(D) 2361 KJ/mol
(E) 2361 KJ/mol
-

25. Increasing order of metallic character

- (A) $\text{Na} > \text{Mg} > \text{Be} > \text{Si} > \text{P}$
(B) $\text{Na} > \text{Be} > \text{P} > \text{Si} > \text{Mg}$
(C) $\text{Mg} > \text{Be} > \text{P} > \text{Si} > \text{Na}$
(D) $\text{P} > \text{Si} > \text{Be} > \text{Mg} > \text{Na}$
(E) $\text{Mg} > \text{Si} > \text{Be} > \text{Na} > \text{P}$
-

1. (3, 4) and (4, a) lie on line. Find a?

2. If $y = 4\sqrt{x}$ then $\frac{d^2y}{dx^2} =$

3. The range of the function $f(x) = \frac{1}{7+4\sin x+3\cos x}$

(A) $[\frac{1}{14}, \frac{1}{4}]$

(B) $[\frac{1}{7}, \frac{1}{3}]$

(C) $[\frac{1}{3}, \frac{1}{7}]$

(D) $[\frac{1}{7}, 1]$

4. Number of words that can be formed starting and ending with the same letter from the word BANANA.

(A) 60

(B) 72

(C) 48

(D) 36

5. Number of ways 3 boys and 4 girls can be arranged such that there is one girl between any 2 boys and one boy between any 2 girls.

6. If $y = \frac{1+\tan^2 x}{1-\tan^2 x}$, find $y'(\frac{\pi}{8})$, where $0 < x < \frac{\pi}{4}$.

7. If $\alpha = \frac{\pi}{4}$, find $(\sin \alpha + \sin \beta)^2 + (\cos \alpha + \cos \beta)^2$.

8. If $y = \log_e(x^3 + 24)$, find $\frac{dy}{dx}$ at $y = \log_e 2$.

9. Find $\int_{-1}^1 \frac{\log(1+|x|)}{1+|x|} dx$

10. Find unit vector parallel to $-(s + 4s)\hat{i} + (7 - 2s)\hat{j} + (3 + 4s)\hat{k}$

11. Compute $\int (\cot 2x + \cos 2x) dx$

(A) $\frac{1}{2} \ln(\sin 2x) + \frac{1}{2} \sin 2x + C$

(B) $\frac{1}{2} \ln(\sin 2x) + \frac{1}{2} \cos 2x + C$

(C) $\ln(\sin 2x) + \sin 2x + C$

(D) $\ln(\sin 2x) + \cos 2x + C$

12. Compute $\int \frac{\sqrt{x+1}}{\sqrt{x}} dx$

(A) $2\sqrt{x+1} - \sqrt{x} + C$

(B) $2\sqrt{x+1} + \sqrt{x} + C$

(C) $\sqrt{x+1} - \sqrt{x} + C$

(D) $\sqrt{x+1} + \sqrt{x} + C$

13. Solve $6(2x + 3) + x > 53 - 2x$.

14. Evaluate $\int \frac{x^2+6x+1}{(x+3)^2} dx$.

15. Find α if

$$\lim_{x \rightarrow 0} \frac{1 - \sec^2(\alpha x)}{\alpha x^2} = -3$$

16. The distance of the point (10, 10, 10) from the Z-axis.

17. Find $\int_0^{\frac{\pi}{2}} \sqrt{\cos x \sin 2x} dx$

18. Find minimum value of $\sin x \sin\left(x + \frac{\pi}{3}\right)$

- (A) $\frac{1}{2}$
- (B) 0
- (C) 1
- (D) $\frac{\sqrt{3}}{2}$

19. Find the integral $\int 2 dy = (y + \cos x) dx$

- (A) $y = \sin x + C$
- (B) $y = \cos x + C$
- (C) $y = x + C$
- (D) $y = \sin x + \cos x + C$

20. If the area of the circle $x^2 + y^2 + 8x - 6y + c = 0$.

21. If $n(B) = 61$, $n(A \cup B) = 99$, $n(A \cap B) = 28$, find $n(A')$.

22. Given that $\vec{a} = 2\hat{i} - \lambda\hat{j} + 5\hat{k}$, $\vec{b} = \mu\hat{i} + 7\hat{j} + 3\hat{k}$, and the midpoint of $\overline{AB} = 3\hat{i} + 2\hat{j} + 4\hat{k}$, find $\lambda + \mu$.

23. If $\alpha^2 - \frac{1}{\alpha^2} = 2$, find $(\alpha + \frac{1}{\alpha})^{16}$.

24. Maximum of $f(x) = \alpha - 4x - x^2$, find $\alpha = ?$

25. If $(2 - x)^9 = a_0 + a_1x + \dots + a_9x^9$, find $a_1 + a_2 + \dots + a_8$

26. Given $y = 4e^{-x} - 2e^{-2x} - e^{-3x}$, find y'' .

(A) $4e^{-x} - 4e^{-2x} - 3e^{-3x}$

(B) $4e^{-x} - 2e^{-2x} - 6e^{-3x}$

(C) $4e^{-x} - 2e^{-2x} - 3e^{-3x}$

(D) $4e^{-x} - 2e^{-2x} - 5e^{-3x}$

27. Given $a_1 + a_2 + a_3 + a_4 = 960$ and $a_4 - 8a = a_1$, find a_1 .

(A) 320

(B) 240

(C) 160

(D) 120

28. Evaluate $\lim_{x \rightarrow 0} \frac{x - \tan(3x)}{\sin(2x)}$

29. Given $y = \frac{1}{1 + \tan x}$, find $f^{-1}(x)$, where $0 < x < \frac{\pi}{2}$.

30. Arithmetic mean Geometric mean of 2 numbers a b in the ratio 5:3. Find $\frac{a^2 + b^2}{ab}$.

31. Find $\int \frac{x \cos 2x}{\cos x - \sin x} dx$

32. If 2 vectors $4\hat{i} + \ell\hat{j} - 6\hat{k}$ and $-6\hat{i} + 12\hat{j} + 9\hat{k}$ are collinear, find ℓ .

33. Given the numbers 4, 7, x , 13, 16, with a mean of 10, find the mean deviation about the mean.

- (A) 2.8
 - (B) 3.2
 - (C) 2.5
 - (D) 3.0
-

34. Given $(3 \cos x - 2 \sec x)^2 = 9 \cos^2 x + 4 \tan^2 x + k$, find k .

- (A) 4
- (B) 5
- (C) 3

(D) 6

35. Evaluate $\lim_{x \rightarrow 1} \frac{\sqrt{x+3} \cdot \sqrt{x-1}}{x-1}$

36. The coefficient of x^3 in $(2+x)^n$ is 160. Find the coefficient of x^6 in $(2-x^2)^n$.

37. If $f(x) = x^2 - 10x$, $g(x) = e^x + 5$, find $g(2x) - f(g(x))$.

38. Max of $Z = 7x + 10y$ subject to $x + y \geq 3$, $x + 2y \geq 4$, $x, y \geq 0$.

39. If $|a - b| = \frac{\sqrt{3}}{2}$ where a and b are unit vectors, find the angle between a and b .

40. Find the value of $\cot 10^\circ \times \cot 30^\circ \times \cot 45^\circ \times \cot 60^\circ \times \cot 80^\circ$

41. If $2 \cot^{-1} \left(\frac{4}{3} \right) = \cos^{-1} \left(\frac{x}{5} \right)$, find x .

(A) 3

(B) 4

(C) 5

(D) 6

42. Find the locus of $Z = x + iy$ satisfying

$$\frac{\operatorname{Re}(z)}{2+i} + \frac{\operatorname{Im}(z)}{1+2i} = \frac{3}{1-2i}$$

- (A) $x^2 + y^2 = 1$
(B) $x^2 + y^2 = 9$
(C) $x^2 + y^2 = 4$
(D) $x^2 + y^2 = 25$
-

43. Find the equation of the curve (x, y) if $\cos^{-1}(x - 2) = \sin^{-1}(y + 1)$.

44. If the sum of the first two terms of a G.P is 12 and the third term is 16, find the common ratio r .

45. If $y = \frac{1}{3\sqrt{x}} \left(\frac{2}{x} - 3 \right)$, find the interval in which y is strictly decreasing.

46. If $Z = 1 + i$ and $Z - 24\bar{Z} = \lambda Z^2$, find λ .
