## DLVED PAPER

2010

## PART - I (PHYSICS)

- A straight wire carrying current i is turned into a circular loop. If the magnitude of magnetic moment associated with it in MKS unit is M, the length of wire will be
- (b)  $\sqrt{\frac{4\pi M}{i}}$

- The ratio of the amounts of heat developed in the four arms of a balance Wheatstone bridge, when the arms have resistances  $P=100 \Omega$ ,

Q=10  $\Omega$ , R=300  $\Omega$  and S=30  $\Omega$  respectively is

- (a) 3:30:1:10
- (b) 30:3:10:1
- (c) 30:10:1:3
- (d) 30:1:3:10
- An electric kettle takes 4 A current at 220V. How much time will it take to boil 1 kg of water from temperature 20 °C? The temperature of boiling water is 100°C.
  - (a) 12.6 min
- (b) 4.2 min
- 6.3 min (c)
- (d) 8.4 min
- Magnetic field at the centre of a circular loop of 4. area A is B. The magnetic moment of the loop will
  - (a)
- (b)  $\frac{BA^{3/2}}{\mu_0\pi}$

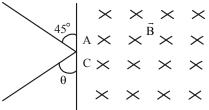
- In Young's double slit experiment, the spacing between the slits is d and wavelength of light used is 6000 Å. If the angular width of a fringe formed on a distance screen is  $1^{\circ}$ , then value of d is
  - (a) 1mm
- (b) 0.05 mm
- $0.03\,\mathrm{mm}$ (c)
- (d) 0.01 mm

- An electric dipole consists of two opposite charges of magnitude  $q = 1 \times 10^{-6}$  C separated by 2.0 cm. The dipole is placed in an external field of  $1\times10^5$  NC<sup>-1</sup>. What maximum torque does the field exert on the dipole? How much work must an external agent do to turn the dipole end to end, starting from position of alignment ( $\theta = 0^{\circ}$ )?
  - (a)  $4.4 \times 10^6 \text{N-m}, 3.2 \times 10^{-4} \text{J}$
  - (b)  $-2 \times 10^{-3} \text{N-m}, -4 \times 10^{3} \text{J}$
  - (c)  $4 \times 10^3$  N-m,  $2 \times 10^{-3}$  J
  - (d)  $2 \times 10^{-3}$  N-m.  $4 \times 10^{-3}$  J
- The electron of hydrogen atom is considered to be revolving round a proton in circular orbit of radius  $h^2/me^2$  with velocity  $e^2/h$ , where  $h = \hbar / 2\pi$ . The current *i* is

  - (a)  $\frac{4\pi^2 \text{me}^5}{\hbar^2}$  (b)  $\frac{4\pi^2 \text{me}^5}{\hbar^3}$
  - (c)  $\frac{4\pi^2 m^2 e^2}{\hbar^3}$  (d)  $\frac{4\pi^2 m^2 e^5}{\hbar^3}$
- In a double slit experiment, 5th dark fringe is formed opposite to one of the slits, the wavelength of light is

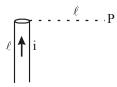
- Which of the following rays is emitted by a human body?
  - (a) X-rays
- (b) UV rays
- (c) Visible rays
- (d) IR rays

**10.** A proton of mass  $1.67 \times 10^{-27}$ kg enters a uniform magnetic field 1T of at point A shown in figure with a speed of  $10^7 \,\mathrm{ms}^{-1}$ .



The magnetic field is directed normal to the plane of paper downwards. The proton emerges out of the magnetic field at point C, then the distance AC and the value of angle  $\theta$  will respectively be

- (a)  $0.7 \text{ m} \cdot 45 \circ$
- (b)  $0.7 \,\mathrm{m}.\,90 \,\circ$
- (c)  $0.14 \text{ m}, 90 \circ$
- (d)  $0.14 \,\mathrm{m}, 45 \,\mathrm{\circ}$
- 11. A neutral water molecule  $(H_2O)$  in its vapour state has an electric dipole moment of magnitude  $6.4 \times 10^{-30}$ C-m. How far apart are the molecules centres of positive and negative charges?
  - (a) 4m
- (b) 4mm
- (c)  $4\mu m$
- (d) 4 pm
- **12.** Figure shows a straight wire length *l* carrying current i. The magnitude of magnetic field produced by the current at point P is



- 13. Zener diode is used for
  - producing oscillations in an oscillator
  - amplification (b)
  - (c) stabilisation
  - (d) rectification
- **14.** Two light sources are said to be coherent if they are obtained from
  - (a) two independent point sources emitting light of the same wavelength
  - a single point source
  - (c) a wide source
  - two ordinary bulbs emitting light of different wavelengths

- 15. A small coil is introduced between the poles of an electromagnet so that its axis coincides with the magnetic field direction. The number of turns is n and the cross-sectional area of the coil is A. When the coil turns through 180° about its diameter, the charge flowing through the coil is Q. The total resistance of the circuit is R. What is the magnitude of the magnetic induction?
  - (a)
- (c) 2RA

(c)

- **16.** The attenuation in optical fibre is mainly due to
  - absorption
    - (b) scattering neither absorption nor scattering
  - (d) Both (a) and (b)
- 17. An arc of radius r carries charge. The linear density of charge is  $\lambda$  and the arc subtends an

angle  $\frac{\pi}{3}$  at the centre. What is electric potential at the centre?

- 18. Sinusoidal carrier voltage of frequency 1.5 MHz and amplitude 50 V is amplitude modulated by sinusoidal voltage of frequency 10 kHz producing 50% modulation. The lower and upper side-band frequencies in kHz are
  - (a) 1490, 1510
- (b) 1510, 1490
- $\frac{1}{1490}$ ,  $\frac{1}{1510}$  (d)  $\frac{1}{1510}$ ,  $\frac{1}{1490}$
- 19.  $50 \Omega$  and  $100 \Omega$  resistors are connected in series. This connection is connected with a battery of 2.4 V. When a voltmeter of 100  $\Omega$  resistance is connected across  $100 \Omega$  resistor, then the reading of the voltmeter will be
  - (a) 1.6V
- (b) 1.0V
- 1.2V (c)
- (d) 2.0V
- In space charge limited region, the plate current in a diode is 10 mA for plate voltage 150V. If the plate voltagte is increased to 600V, then the plate current will be
  - $10\,\mathrm{mA}$ (a)
- (b) 40 mA
- 80 mA
- (d) 160 mA

- **21.** Light of wavelength  $\lambda$  strikes a photo-sensitive surface and electrons are ejected with kinetic energy E. If the kinetic energy is to be increased to 2E, the wavelength must be changed to  $\lambda'$ wher
  - (a)  $\lambda' = \frac{\lambda}{2}$  (b)  $\lambda' = 2\lambda$
  - (c)  $\frac{\lambda}{2} < \lambda' < \lambda$  (d)  $\lambda' > \lambda$
- 22. The maximum velocity of electrons emitted from a metal surface is v, when frequency of light falling on it is f. The maximum velocity when frequency becomes 4f is
  - (a) 2v
- (b) > 2v
- (c) < 2v
- (d) between 2v and 4v
- 23. The collector plate in an experiment on photoelectric effect is kept vertically above the emitter plate. Light source is put on and a saturation photo-current is recorded. An electric field is switched on which has a vertically downward direction, then
  - the photo-current will increase
  - the kinetic energy of the electrons will increase
  - the stopping potential will decrease
  - (d) the threshold wavelength will increase
- 24. A cylindrical conductor of radius R carries a current i. The value of magnetic field at a point

which is  $\frac{R}{4}$  distance inside from the surface is

10 T. The value of magnetic field at point which is 4R distance outside from the surface

- (b)  $\frac{8}{3}$  T
- (c)  $\frac{40}{3}$ T (d)  $\frac{80}{3}$ T
- **25.** The power of a thin convex lens  $\binom{a}{n_g} = 1.5$  is 5.0 D. When it is placed in a liquid of refractive index and, then it behaves as a concave lens of focal length 100cm. The refractive index of the liquid an, will be
  - (a) 5/3
- (b) 4/3
- (c)  $\sqrt{3}$
- (d) 5/4
- 26. Find the value of magnetic field between plates of capacitor at a distance 1m from centre, where electric field varies by 10<sup>10</sup> V/m per second.
  - (a)  $5.56 \times 10^{-8}$ T
- (b)  $5.56 \times 10^{-3}$ T
- (c)  $5.56 \mu T$
- (d) 5.55T

- 27. Using an AC voltmeter the potential difference in the electrical line in a house is read to be 234V. If line frequency is known to be 50 cycles/s, the equation for the line voltage is
  - (a)  $V=165\sin(100 \pi t)$
  - (b)  $V=331\sin(100 \pi t)$
  - (c)  $V=220\sin(100 \pi t)$
  - (d)  $V=440\sin(100 \pi t)$
- **28.** There are a 25W 220 V bulb and a 100W 220Vline. Which eletric bulb will glow more brightly?
  - (a) 25W bulb
  - (b) 100W bulb
  - Both will have equal incadescene
  - (d) Neither 25 W nor 100 W bulb will give light
- 29. Silver has a work function of 4.7 eV. When ultraviolet light of wavelength 100 nm is incident upon it, potential of 7.7 V is required to stop photoelectrons from reaching the collector plate. The potential required to stop electrons when light of wavelength 200 nm is incident upon silver
  - (a) 1.5V
- (b) 1.85V
- (c) 1.95V
- (d) 2.37V
- 30. Two particles X and Y having equal charges, after being accelerated through the same potential difference, enter a region of uniform magnetic field and describe circular paths of radii R<sub>1</sub> and R<sub>2</sub>, Respectively. The ratio of masses of X and Y
  - (a)  $(R_1/R_2)^{-2}$
- (b)  $(R_2/R_1)$
- (c)  $(R_1/R_2)^2$
- (d)  $(R_1/R_2)$
- 31. According to the Bohr's theory of hydrogen atom, the speed of the electron, energy and the radius of its orbit vary with the principal quantum number n, respectively, as
  - (a)  $\frac{1}{n}, \frac{1}{n^2}, n^2$  (b)  $\frac{1}{n}, n^2, \frac{1}{n^2}$
  - (c)  $n^2, \frac{1}{n^2}, n^2$  (d)  $n, \frac{1}{n^2}, \frac{1}{n^2}$
- 32. In the hydrogen atom, the electron is making  $6.6 \times 10^{15}$  rps. If the radius of orbit is  $0.53 \times 10^{-10}$  m, then magnetic field produced at the centre of the orbit is
  - 140T
- (b) 12.5T
- (c) 1.4T
- (d) 0.14T

- **33.** Two identical light sources  $S_1$  and  $S_2$  emit light of same wavelength  $\,\lambda$  . These light rays will exhibit interference if
  - (a) their phase differences remain constant
  - their phases are distributed randomly (b)
  - (c) their light intensities remain constant
  - (d) their light intensities change randomly
- 34. In Meter bridge or Wheatstone bridge for measurement of resistance, the known and the unknown resistances are interchanged. The error so removed is
  - (a) end correction
  - index error (b)
  - due to temperature effect (c)
  - (d) random error
- **35.** A fish, looking up through the water, sees the outside world contained in a circular horizon. If the refractive index of water is 4/3 and the fish is 12cm below the surface of water, the radius of the circle in centimetre is

(a) 
$$\frac{12 \times 3}{\sqrt{5}}$$

(b) 
$$12 \times 3 \times \sqrt{5}$$

(c) 
$$\frac{12\times3}{\sqrt{7}}$$

(d) 
$$12 \times 3 \times \sqrt{7}$$

- **36.** Radio waves diffract around building althrough light waves do not. The reason is that radio waves
  - (a) travel with speed larger than c
  - (b) have much larger wavelength then light
  - (c) carry news
  - (d) are not electromagnetic waves
- 37. In the Bohr model of a hydrogen atom, the centripetal force is furnished by the coulomb attraction between the proton and the electron. If a<sub>0</sub> is the radius of the ground state orbit, m is the mass and e is charge on the electron and  $\varepsilon_0$  is the vacuum permittivity, the speed of the electron

(b) 
$$\frac{e}{\sqrt{\epsilon_0 a_0 m}}$$

(c) 
$$\frac{e}{\sqrt{4\pi\epsilon_0 a_0 m}}$$

$$\text{(c)} \quad \frac{e}{\sqrt{4\pi\epsilon_0 a_0 m}} \qquad \text{(d)} \quad \frac{\sqrt{4\pi\epsilon_0 a_0 m}}{e}$$

- **38.** A potential difference of 2V is applied between the opposite faces of a Ge crystal plate of area 1 cm<sup>2</sup> and thickness 0.5 mm. If the concentration of electrons in Ge is  $2 \times 10^{19}$ /m<sup>2</sup> and mobilities of electrons and holes are 0.36 m<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup> and 0.14 m<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup> respectively, then the current flowing through the plate will be
  - (a)  $0.25 \,\mathrm{A}$
- (b) 0.45 A
- (c) 0.56A
- (d) 0.64A

- **39.** An AM wave has 1800 W of total power content. For 100% modulation the carrier should have power content equal to
  - (a) 1000 W
- (b) 1200 W
- (c) 1500 W
- (d) 1600 W
- **40.** Two light rays having the same wavelength  $\lambda$  in vacuum are in phase initially. Then the first ray travels a path  $l_1$  through a medium of refractive index  $n_1$  while the second ray travels a path of length  $l_2$  through a medium of refractive index  $n_2$ . The two waves are then combined to observe interference. The phase difference between the

(a) 
$$\frac{2\pi}{\lambda}(l_2-l_1)$$

(a) 
$$\frac{2\pi}{\lambda}(l_2 - l_1)$$
 (b)  $\frac{2\pi}{\lambda}(n_1 l_2 - n_2 l_1)$ 

(c) 
$$\frac{2\pi}{\lambda} (n_2 l_2 - n_1 l_1)$$
 (d)  $\frac{2\pi}{\lambda} (\frac{l_1}{n_1} - \frac{l_2}{n_2})$ 

(d) 
$$\frac{2\pi}{\lambda} \left( \frac{l_1}{n_1} - \frac{l_2}{n_2} \right)$$

## PART - II (CHEMISTRY)

- 41. The correct formula of the complex tetraammineaquachlorocobalt (III) chloride is
  - (a) [Cl(H<sub>2</sub>O)(NH<sub>3</sub>)<sub>4</sub>Co]Cl
  - (b) [CoCl(H<sub>2</sub>O)(NH<sub>3</sub>)<sub>4</sub>]Cl
  - (c)  $[Co(NH_3)_4(H_2O)Cl]Cl$
  - (d) [CoCl(H<sub>2</sub>O)(NH<sub>3</sub>)<sub>4</sub>]Cl<sub>2</sub>
- **42.** The equivalent conductance at infinite dilution of a weak acid such as HF
  - can be determined by extrapolation of measurements on dilute solutions of HCl, HBr and HI
  - can be determined by measurement on very dilute HF solutions
  - can best be determined from measurements on dilute solutions of NaF, NaCl and HCl
  - (d) is an undefined quantity

43. 
$$C_2H_5I \xrightarrow{\text{Alcoholic}} X \xrightarrow{\text{Br}_2} Y \xrightarrow{\text{KCN}} Z$$

$$A \xleftarrow{H_3O^+}$$

The product 'A' is

- (a) succinic acid
- (b) melonic acid
- (c) oxalic acid
- (d) maleic acid
- **44.** For a reaction of type  $A + B \rightarrow products$ , it is observed that doubling concentration of A causes the reaction rate to be four times as great, but doubling amount of B does not affect the rate. The unit of rate constant is
  - (a)  $s^{-1}$
- (c)  $s^{-1} \text{ mol}^{-1} L$
- $\begin{array}{ccc} & & \text{(b)} & \text{s}^{-1} \; \text{mol} \; L^{-1} \\ L & & \text{(d)} & \text{s} \; \text{s}^{-1} \; \text{mol}^{-2} \; L^2 \end{array}$

|            | (a) $\bar{k}_2 = 4k_1$ (b) $k_2 = 2k_1$   |  |  |  |
|------------|---|--|--|--|
|            | (c) $k_2 = 0.25 k_1$ (d) $k_2 = 0.5 k_1$  |  |  |  |
| 46.        | The formula of ethyl carbinol is  |  |  |  |
|            | (a) CH <sub>3</sub> OH (b) CH <sub>3</sub> CH <sub>2</sub> OH<br>(c) CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> OH (d) (CH <sub>3</sub> ) <sub>3</sub> COH |  |  |  |
|            | (c) $CH_3CH_2CH_2OH$ (d) $(CH_3)_3COH$  |  |  |  |
| 47.        |   |  |  |  |
|            | Meyer's test?   |  |  |  |
|            | (a) n-propyl alcohol (b) Isopropyl alcohol  |  |  |  |
| 4.0        | (c) tert-butyl alcohol (d) sec-butyl alcohol  |  |  |  |
| 48.        | Enthalpy of a compound is equal to its  |  |  |  |
|            | (a) heat of combustion(b) heat of formation   |  |  |  |
| 40         | (c) heat of reaction (d) heat of solution   |  |  |  |
| 49.        | For which one of the following reactions will there   |  |  |  |
|            | be a positive $\Delta S$ ?  |  |  |  |
|            | (a) $H_2O(g) \longrightarrow H_2O(l)$   |  |  |  |
|            | (b) $H_2 + I_2 \longrightarrow 2HI$   |  |  |  |
|            | (c) $CaCO_3(s) \longrightarrow CaO(s) + CO_2(g)$  |  |  |  |
|            | (d) $N_2(g) + 3H_2(g) \longrightarrow 2NH_3(g)$   |  |  |  |
| <b>50.</b> | Across the lanthanide series, the basicity of the   |  |  |  |
|            | lanthanide hydroxides   |  |  |  |
|            | (a) increases   |  |  |  |
|            | (b) decreases   |  |  |  |
|            | (c) first increases and then decreases  |  |  |  |
| <b>-1</b>  | (d) first decreases and then increases  |  |  |  |
| 51.        | 1   |  |  |  |
|            | ethoxide, the product obtained is (a) p-nitroanisole (b) ethyl phenyl ether   |  |  |  |
|            | (c) p-nitrophenetole (d) no reaction occurs   |  |  |  |
| 52.        |   |  |  |  |
| 34.        | narticles and forms V <sup>235</sup> Element V is   |  |  |  |
|            | particles and forms $\frac{76}{76}$ 1. Element X is   |  |  |  |
|            | particles and forms $_{76}Y^{235}$ . Element X is  (a) $_{81}X^{247}$ (b) $_{80}X^{247}$ (c) $_{81}X^{246}$ (d) $_{80}X^{246}$                                  |  |  |  |
| 53         | For the reaction, $(u)_{80}X$   |  |  |  |
| 33.        | · · · · · · · · · · · · · · · · · · ·   |  |  |  |
|            | $2A(g) + B_2(g) \Longrightarrow 2AB_2(g)$   |  |  |  |
|            | the equilibrium constant, $K_p$ at 300 K is 16.0. The   |  |  |  |
|            | value of $K_p$ for $AB_2(g) \rightleftharpoons A(g) + 1/2 B_2(g)$   |  |  |  |
|            | is  |  |  |  |
|            | (a) 8 (b) 0.25  |  |  |  |
|            | (c) 0.125 (d) 32  |  |  |  |
| 54.        | Frenkel defect is generally observed in   |  |  |  |
|            | (a) AgBr (b) AgI  |  |  |  |
|            |   |  |  |  |
|            | (c) ZnS (d) All of the above  |  |  |  |

atoms, ions or molecules are

(a) weakly bonded together

(b) strongly bonded together

45. A chemical reaction was carried out at 320 K and

and k2 respectively. Then

300 K. The rate constants were found to be k<sub>1</sub>

(c) spherically symmetrical (d) arranged in planes **56.**  $[Co (NH_3)_4Cl_2]NO_2$  and  $[Co (NH_3)_4ClNO_2]Cl$ exhibit which type of isomerism? (a) Geometrical (b) Optical (c) Linkage (d) Ionisation 57. Which of the following compounds is not coloured? (a)  $Na_2[Cu(Cl_4]]$ (b) Na[Cd(Cl)<sub>4</sub>] (c)  $K_4[Fe(CN)_6]$ (d)  $K_3[Fe(CN)_6]$ Which of the following is a Gattermann aldehyde synthesis? H + CO + HC1+ HCl + HCN CHO **59.** Aldol is β-hydroxybutyraldehyde (a)  $\alpha$ -hydroxybutanal (b) β-hydroxypropanal (d) None of the above **60.** Nitrobenzene can be converted into azobenzene by reduction with (a)  $Zn, NH_4Cl, \Delta$ (b) Zn/NaOH, CH<sub>3</sub>OH (c) Zn/NaOH (d) LiAlH<sub>4</sub>, ether **61.** The one which is least basic is  $NH_3$ (b)  $C_6H_5NH_2$ (a)

 $(C_6H_5)_3N$ 

(a) Haemoglobin

Florigen

(a) 3

(c) 4

world?

(c)

**62.** Coordination number of Ni in  $[Ni(C_2O_4)_3]^{4-}$  is

(d)  $(\tilde{C}_{6}\tilde{H}_{5})_{2}\tilde{N}H$ 

(b) Chlorophyll

(d) ATP

(b) 6

(d) 5 Mg is an important component of which biomolecule occurring extensively in living

| •          | () A 3.50   | ,          | The field of the product in the reaction                                  |
|------------|---|------------|---|
|            | (a) $AgNO_3$  |            | $A_2(g) + 2B(g) \rightleftharpoons C(g) + Q kJ$                           |
|            | (b) $Ag_2S$   |            | would be higher at  |
|            | (c) Alloy of $80\%$ Ag $+20\%$ Cu   |            | (a) high temperature and high pressure                                    |
|            | (d) AgCl  |            | (b) high temperature and low pressure                                     |
| <b>65.</b> | Identify the statement which is not correct   |            | (c) low temperature and high pressure                                     |
|            | regarding CuSO₄   |            | (d) low temperature and low pressure                                      |
|            | (a) It reacts with KI to give iodine  | 74.        | In which of the following case, does the reaction                         |
|            | (b) It reacts with KCl to give Cu <sub>2</sub> Cl <sub>2</sub>                                |            | go farthest to completion?  |
|            | (c) It reacts with NaOH and glucose to give   |            | (a) $K = 10^2$ (b) $K = 10$   |
|            | Cu <sub>2</sub> O   |            | (c) $K = 10^{-2}$ (d) $K = 1$   |
|            | (d) It gives CuO on strong heating in air   | 75         | Formation of cyanohydrin from a ketone is an                              |
| 66.        | Transition metals usually exhibit highest oxidation   | 13.        | example of  |
|            | states in their   |            |   |
|            | (a) chlorides (b) fluorides   |            |   |
|            | (c) bromides (d) iodides  |            | (b) nucleophilic addition   |
| 67         | The number of Faradays needed to reduce 4 g   |            | (c) nucleophilic substitution   |
| 07.        | equivalents of Cu <sup>2+</sup> to Cu metal will be   | 70         | (d) electrophilic substitution  |
|            | (a) 1 (b) 2   | /0.        | Glycerol on treatment with oxalic acid at 110°C                           |
|            | (a) 1 (b) 2   |            | forms   |
|            | (c) $\frac{1}{2}$ (d) 4   |            | (a) formic acid (b) allyl alcohol   |
|            | (c) $\frac{1}{2}$ (d) 4   |            | (c) CO <sub>2</sub> and CO (d) acrolein                                   |
| 68.        | Which one of the following cells can convert  | 77.        | The activity of an old piece of wood is just 25%                          |
|            | chemical energy of H <sub>2</sub> and O <sub>2</sub> directly into                            |            | of the fresh piece of wood. If $t_{1/2}$ of C-14 is 6000                  |
|            | electrical energy?  |            | yr, the age of piece of wood is   |
|            | (a) Mercury cell (b) Daniel cell  |            | (a) 6000 yr (b) 3000 yr   |
|            | (c) Fuel cell (d) Lead storage cell   |            | (a) 6000 yr (b) 5000 yr (c) 9000 yr (d) 12000 yr                          |
| 69         | On treatment of propanone with dilute $Ba(OH)_2$ ,  | <b>78.</b> | The radius of Na <sup>+</sup> is 95 pm and that of Cl <sup>-</sup> ion is |
| 0).        | the product formed is   |            | 181 pm. Hence, the coordination number of Na <sup>+</sup>                 |
|            | (a) aldol   |            | will be   |
|            | (b) phorone   |            | (a) 4 (b) 6   |
|            | (c) propionaldehyde   |            | (c) 8 (d) unpredictable   |
|            |   | 79.        | The reaction, $ROH + H_2CN_2$ in the presence of                          |
| 70         | (d) 4-hydroxy-4-methyl-2-pentanone Which of the following converts CH, CONH, to               |            | HBF <sub>4</sub> , gives the following product                            |
| 70.        | Which of the following converts CH <sub>3</sub> CONH <sub>2</sub> to                          |            | (a) ROCH <sub>3</sub> (b) RCH <sub>2</sub> OH                             |
|            | CH <sub>3</sub> NH <sub>2</sub> ?<br>(a) NaBr (b) NaOBr                                       |            | (c) $ROHCN_2N_2$ (b) $RCH_2CH_3$  |
|            |   | 80         | The fatty acid which shows reducing property is                           |
| 71         | (c) Br <sub>2</sub> (d) None of the above   | 00.        | (a) acetic acid (b) ethanoic acid   |
| /1.        | Which metal aprons are worn by radiographer to  |            | (c) oxalic acid (d) formic acid   |
|            | protect him from radiation?   |            | (e) Oxane acid (d) for the acid   |
|            | (a) Mercury coated apron  |            | PART - III (MATHEMATICS)  |
|            | (b) Lead apron  |            |   |
|            | (c) Copper apron  | 81.        | If F is function such that $F(0) = 2$ , $F(1) = 3$ ,                      |
|            | (d) Aluminimised apron  |            | $F(x+2)=2F(x) - F(x+1)$ for $x \ge 0$ , then $F(5)$ is                    |
| 12.        | The standard Gibb's free energy change, $\Delta G^{o}$ is                                     |            | equal to  |
|            | related to equilibrium constant, K <sub>p</sub> as  |            | (a) -7 (b) -3<br>(c) 17 (d) 13  |
|            | $\Gamma$ , $ abla \Delta G^{\circ}$   | 00         |   |
|            | (a) $K_p = -RT \ln \Delta G^{\circ}$ (b) $K_p = \left[\frac{e}{RT}\right]^{\Delta G^{\circ}}$ | 82.        | Let $S$ be a set containing $n$ elements. Then,                           |
|            | (a) p [RT]  |            | number of binary operations on S is                                       |
|            |   |            | (a) $n^n$ (b) $2^{n^2}$   |
|            | (c) $K_p = -\frac{\Delta G}{RT}$ (d) $K_p = e^{-\Delta G^{\circ}/RT}$                         |            | · · · · · · · · · · · · · · · · · · ·                                     |
|            | P RT  |            | (c) $n^{n^2}$ (d) $n^2$   |
|            |   |            |   |
|            |   |            |   |
|            |   |            |   |

**73.** The yield of the product in the reaction

**64.** Sterling silver is

**83.** The numerically greatest term in the expansion

of 
$$(3-5x)^{11}$$
 when  $x = \frac{1}{5}$ , is

- (a)  $55 \times 3^9$
- (b)  $55 \times 3^6$
- (c)  $45 \times 3^9$
- (d)  $45 \times 3^6$
- 84. The number of solutions of the equation  $\sin(e^x) = 5^x + 5^{-x}$ , is
  - (a) 0
- (b) 1
- (c) 2
- (d) infinitely many
- **85.** If  $a^x = b^y = c^z = d^u$  and a, b, c, d are in GP, then x, y, z, u are in
  - (a) AP
- (b) GP
- (c) HP
- (d) None of these
- **86.** If z satisfies the equation |z|-z=1+2i, then z is
  - (a)  $\frac{3}{2} + 2i$  (b)  $\frac{3}{2} 2i$
  - (c)  $2 \frac{3}{2}i$  (d)  $2 + \frac{3}{2}i$
- **87.** If  $z = \frac{1 i\sqrt{3}}{1 + i\sqrt{3}}$ , then arg (z) is
  - (a) 60°
- (b) 120°
- (c) 240°
- (d) 300°
- **88.** If  $f(x) = \sqrt{\log_{10} x^2}$ . The set of all values of x for which f(x) is real, is
  - (a) [-1, 1]
- (b)  $[1, \infty]$
- (c)  $(-\infty, -1]$
- (d)  $(-\infty, -1] \cup [1, \infty)$
- 89. For what values of m can the expression  $2x^2 + mxy + 3y^2 - 5y - 2$

be expressed as the product of two linear factors?

- (a) 0
- (b)  $\pm 1$
- (c)  $\pm 7$
- (d) 49
- 90. If B is a non-singular matrix and A is a square matrix, then  $\det(B^{-1}AB)$  is equal to
  - (a)  $\det(A^{-1})$
- (b)  $\det(B^{-1})$
- (c) det (A)
- (d) det (B)
- **91.** If f(x), g(x) and h(x) are three polynomials of degree 2 and

$$\Delta(x) = \begin{vmatrix} f(x) & g(x) & h(x) \\ f'(x) & g'(x) & h'(x) \\ f''(x) & g''(x) & h''(x) \end{vmatrix}$$

then  $\Delta(x)$  is a polynomial of degree

- (a) 2
- (b) 3
- (c) 0
- (d) atmost 3

- **92.** The chances of defective screws in three boxes
  - A, B, C are  $\frac{1}{5}$ ,  $\frac{1}{6}$ ,  $\frac{1}{7}$  respectively. A box is selected

at random and a screw drawn from it at random is found to be defective. Then, the probability that it came from box A, is

- (b)  $\frac{1}{15}$
- (d)  $\frac{42}{107}$
- 93. The value of  $\frac{\cos \theta}{1+\sin \theta}$  is equal to
  - (a)  $\tan\left(\frac{\theta}{2} \frac{\pi}{4}\right)$  (b)  $\tan\left(-\frac{\pi}{4} \frac{\theta}{2}\right)$
  - (c)  $\tan\left(\frac{\pi}{4} \frac{\theta}{2}\right)$  (d)  $\tan\left(\frac{\pi}{4} + \frac{\theta}{2}\right)$
- 94. If  $3\sin\theta + 5\cos\theta = 5$ , then the value of  $5 \sin \theta - 3 \cos \theta$  is equal to
  - (a) 5
- (c) 4
- (d) None of these
- 95. The principal value of  $\sin -1 \left\{ \sin \frac{5\pi}{6} \right\}$  is

- (d) None of these
- **96.** A rod of length *l* slides with its ends on two perpendicular lines. Then, the locus of its mid
  - (a)  $x^2 + y^2 = \frac{l^2}{4}$  (b)  $x^2 + y^2 = \frac{l^2}{2}$
  - (c)  $x^2 y^2 = \frac{1^2}{4}$  (d) None of these
- 97. The equation of straight line through the intersection of line 2x + y=1 and 3x + 2y=5 and passing through the origin is
  - (a) 7x + 3y = 0
- (b) 7x y = 0
- (c) 3x + 2y = 0
- (d) x + y = 0
- **98.** The line joining (5,0) to  $(10\cos\theta, 10\sin\theta)$  is divided internally in the ratio 2:3 at P. If  $\theta$  varies, then the locus of P is

- (a) a straight line
- (b) a pair of straight lines
- (c) a circle
- (d) None of the above
- **99.** If 2x + y + k = 0 is a normal to the parabola  $y^2 = -8x$ , then the value of k, is
- (b) 16
- (c) 24
- **100.**  $\lim_{n\to\infty} \left[ \frac{1}{1.2} + \frac{1}{2.3} + \frac{1}{3.4} + \dots + \frac{1}{n(n+1)} \right]$  is equal

  - (a)
- (b) -1
- 1 0 (c)
- (d) None of these
- **101.** The condition that the line lx + my = 1 may be normal to the curve  $y^2 = 4ax$ , is
  - (a)  $al^3 2alm^2 = m^2$  (b)  $al^2 + 2alm^3 = m^2$
  - (c)  $al^3 + 2alm^2 = m^3$  (d)  $al^3 + 2alm^2 = m^2$
- **102.** If  $\int f(x)dx = f(x)$ , then  $\int \{f(x)\}^2 dx$  is equal to
  - (a)  $\frac{1}{2} \{f(x)\}^2$  (b)  $\{f(x)\}^3$
  - (c)  $\frac{\{f(x)\}^3}{2}$  (d)  $\{f(x)\}^2$
- 103.  $\int \sin^{-1} \left\{ \frac{(2x+2)}{\sqrt{4x^2+8x+13}} \right\} dx$  is equal to
  - (a)  $(x+1) \tan^{-1} \left( \frac{2x+2}{2} \right) \frac{3}{4} \log \left( \frac{4x^2 + 8x + 13}{9} \right) + c$
  - (b)  $\frac{3}{2} \tan^{-1} \left( \frac{2x+2}{3} \right) \frac{3}{4} \log \left( \frac{4x^2 + 8x + 13}{9} \right) + c$
  - (c)  $(x+1)\tan^{-1}\left(\frac{2x+2}{3}\right) \frac{3}{2}\log\left(4x^2 + 8x + 13\right) + c$
  - (d)  $\frac{3}{2}(x+1)\tan^{-1}\left(\frac{2x+2}{3}\right) \frac{3}{4}\log(4x^2 + 8x + 13) + c$
- 104. If the equation of an elipse is  $3x^2 + 2y^2 + 6x - 8y + 5 = 0$ , then which of the following are true?
  - (a)  $e = \frac{1}{\sqrt{3}}$
  - (b) centre is (-1, 2)
  - (c) foci are (-1, 1) and (-1, 3)
  - (d) All of the above

- **105.** The equation of the common tangents to the two
  - hyperbolas  $\frac{x^2}{h^2} \frac{y^2}{h^2} = 1$  and  $\frac{y^2}{h^2} \frac{x^2}{h^2} = 1$ , are
  - (a)  $y = \pm x \pm \sqrt{b^2 a^2}$
  - (b)  $v = \pm x \pm \sqrt{a^2 h^2}$
  - (c)  $y = \pm x \pm \sqrt{a^2 + b^2}$
  - (d)  $y = \pm x \pm \sqrt{a^2 b^2}$
- **106.** Domain of the function  $f(x) = \log_x \cos x$ , is
  - (a)  $\left(-\frac{\pi}{2}, \frac{\pi}{2}\right) \{1\}$  (b)  $\left[-\frac{\pi}{2}, \frac{\pi}{2}\right] \{1\}$
  - (c)  $\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$
- (d) None of these
- **107.** Range of the function  $y = \sin^{-1}\left(\frac{x^2}{1+x^2}\right)$ , is
  - (a)  $\left(0,\frac{\pi}{2}\right)$
- (b)  $\left[0, \frac{\pi}{2}\right]$
- (c)  $\left(0, \frac{\pi}{2}\right]$  (d)  $\left[0, \frac{\pi}{2}\right]$
- **108.** If  $x = \sec \theta \cos \theta$ ,  $y = \sec^n \theta \cos^n \theta$ , then
  - $(x^2+4)\left(\frac{dy}{dx}\right)^2$  is equal to
  - (a)  $n^2(y^2-4)$  (b)  $n^2(4-y^2)$
  - (c)  $n^2(y^2+4)$  (d) None of these
- 109. If  $y = \sqrt{x + \sqrt{y + \sqrt{x + \sqrt{y + \dots \infty}}}}$ , then  $\frac{dy}{dx}$  is

  - (a)  $\frac{y+x}{y^2-2x}$  (b)  $\frac{y^3-x}{2v^2-2xv-1}$
  - (c)  $\frac{y^3 + x}{2v^2 x}$
- (d) None of these
- 110. If  $\int_{1/\sqrt{t^2-1}}^{\infty} \frac{dt}{t} = \frac{\pi}{6}$ , then x can be equal to
  - (a)  $\frac{2}{\sqrt{3}}$
- (b)  $\sqrt{3}$
- (d) None of these

- 111. The area bounded by the curve  $y=|\sin x|$ , x-axis and the lines  $|x| = \pi$ , is
  - (a) 2 sq unit
- (b) 1 sq unit
- (c) 4 sq unit
- (d) None of these
- 112. The degree of the differential equation of all curves having normal of constant length c is
- (b) 3
- (c) 4
- (d) None of these
- **113.** If  $\vec{a} = 2\hat{i} + 2\hat{j} + 3\hat{k}$ ,  $\vec{b} = -\hat{i} + 2\hat{j} + \hat{k}$  and  $\vec{c} = 3\hat{i} + \hat{j}$ ,

then  $\vec{a} + t\vec{b}$  is perpendicular to  $\vec{c}$ , if t is equal to

- (c) 6
- (d) 8
- 114. The distance between the line

 $\vec{r} = 2\hat{i} - 2\hat{j} + 3\hat{k} + \lambda(\hat{i} - \hat{j} + 4\hat{k})$  and the plane

- $\vec{r} \cdot (\hat{i} + 5\hat{j} + \hat{k}) = 5$ , is
- (a)  $\frac{10}{3}$  (b)  $\frac{10}{\sqrt{3}}$

- 115. The equation of sphere concentric with the sphere  $x^2 + y^2 + z^2 - 4x - 6y - 8z - 5 = 0$  and which passes through the origin, is
  - (a)  $x^2 + y^2 + z^2 4x 6y 8z = 0$
  - (b)  $x^2 + y^2 + z^2 6y 8z = 0$
  - (c)  $x^2 + y^2 + z^2 = 0$
  - (d)  $x^2 + y^2 + z^2 4x 6y 8z 6 = 0$
- 116. If the lines  $\frac{x-1}{2} = \frac{y+1}{3} = \frac{z-1}{4}$  and

 $\frac{x-3}{1} = \frac{y-k}{2} = \frac{z}{1}$  intersect, then the value of k,

- (b)  $\frac{9}{2}$
- (d)  $-\frac{3}{2}$
- 117. The two curves  $y = 3^x$  and  $y = 5^x$  intersect at an
  - (a)  $\tan^{-1} \left( \frac{\log 3 \log 5}{1 + \log 3 \log 5} \right)$
  - (b)  $\tan^{-1} \left( \frac{\log 3 + \log 5}{1 \log 3 \log 5} \right)$
  - (c)  $\tan^{-1} \left( \frac{\log 3 + \log 5}{1 + \log 3 \log 5} \right)$
  - (d)  $\tan^{-1} \left( \frac{\log 3 \log 5}{1 \log 3 \log 5} \right)$
- **118.** The equation  $\lambda x^2 + 4xy + y^2 + \lambda x + 3y + 2 = 0$ represents a parabola, if  $\lambda$  is
  - (a) 0
- (c) 2
- (d) 4
- **119.** If two circles  $2x^2 + 2y^2 3x + 6y + k = 0$  and  $x^{2} + y^{2} - 4x + 10y + 16 = 0$  cut orthogonally, then the value of k is
  - (a) 41
- (b) 14
- (c) 4
- (d) 1
- **120.** If A (-2, 1), B (2, 3) and C (-2, -4) are three points. Then, the angle between BA and BC is
  - (a)  $\tan^{-1}\left(\frac{2}{3}\right)$  (b)  $\tan^{-1}\left(\frac{3}{2}\right)$
  - (c)  $\tan^{-1} \left( \frac{1}{3} \right)$  (d)  $\tan^{-1} \left( \frac{1}{2} \right)$