

JEE Main 2021 Paper 1 (B.E./B.Tech) 26 February 2021 Shift 1 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2021 - Session 1**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 26 February 2021, Shift 1. The total duration is **3 hours**.
- (ii) The paper has **90 questions**, of which **75 are to be attempted**, and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 30 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **10 numerical value questions**; attempt **any 5** of them.
- (iv) **All Section A questions are compulsory**. Attempt only the specified number of Section B questions. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme**: MCQs carry **+4** for a correct answer and **-1** for an incorrect answer. Numerical questions carry **+4** for a correct answer and **0** for an incorrect answer. Unanswered questions carry **0**.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

1. The maximum value of the term independent of 't' in the expansion of

$$\left(tx^{\frac{1}{5}} + \frac{(1-x)^{\frac{1}{10}}}{t} \right)^{10}$$

where $x \in (0, 1)$ is :

- (A) $\frac{10!}{\sqrt{3} (5!)^2}$
 (B) $\frac{2 \cdot 10!}{3\sqrt{3} (5!)^2}$
 (C) $\frac{10!}{3(5!)^2}$
 (D) $\frac{2 \cdot 10!}{3(5!)^2}$

2. The value of

$$\lim_{h \rightarrow 0} 2 \left\{ \frac{\sqrt{3} \sin\left(\frac{\pi}{6} + h\right) - \cos\left(\frac{\pi}{6} + h\right)}{\sqrt{3} h (\sqrt{3} \cosh - \sinh)} \right\}$$

is :

- (A) $\frac{4}{3}$
 (B) $\frac{2}{3}$
 (C) $\frac{3}{4}$
 (D) $\frac{2}{\sqrt{3}}$

3. Let A be a symmetric matrix of order 2 with integer entries. If the sum of the diagonal elements of A^2 is 1, then the possible number of such matrices is :

- (A) 6
 (C) 1
 (B) 4
 (D) 12

4. The value of

$$\int_{-\pi/2}^{\pi/2} \frac{\cos^2 x}{1 + 3^x} dx$$

is :

- (A) 2π
 (B) $\frac{\pi}{2}$
 (C) 4π
 (D) $\frac{\pi}{4}$

5. The number of seven digit integers with sum of the digits equal to 10 and formed by using the digits 1, 2 and 3 only is :

- (A) 35
 (C) 82
 (B) 42
 (D) 77

6. Let $R = \{(P, Q) \mid P \text{ and } Q \text{ are at the same distance from the origin}\}$ be a relation, then the equivalence class of $(1, -1)$ is the set :

- (A) $S = \{(x, y) \mid x^2 + y^2 = \sqrt{2}\}$
 (B) $S = \{(x, y) \mid x^2 + y^2 = 2\}$
 (C) $S = \{(x, y) \mid x^2 + y^2 = 1\}$
 (D) $S = \{(x, y) \mid x^2 + y^2 = 4\}$

7. The value of

$$\sum_{n=1}^{100} \int_{n-1}^n e^{x-[x]} dx$$

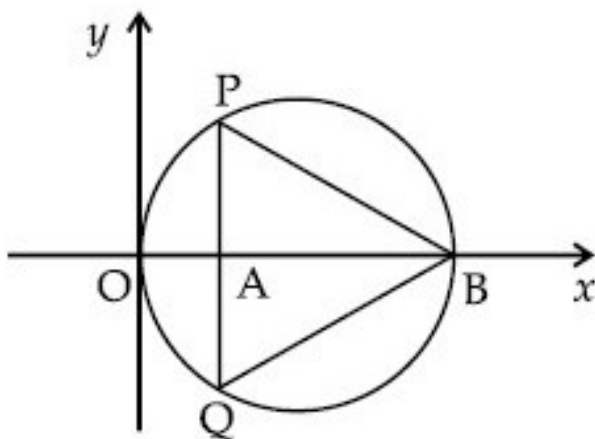
, where $[x]$ is the greatest integer $\leq x$, is :

- (A) $100e$
 (C) $100(1 + e)$
 (B) $100(e - 1)$
 (D) $100(1 - e)$

8. The intersection of three lines $x - y = 0$, $x + 2y = 3$ and $2x + y = 6$ is a :

- (A) Right angled triangle
(B) Equilateral triangle
(C) None of the above
(D) Isosceles triangle

9. In the circle given below, let $OA = 1$ unit, $OB = 13$ unit and $PQ \perp OB$. Then, the area of the triangle PQB (in square units) is :



- (A) $24\sqrt{2}$
(B) $24\sqrt{3}$
(C) $26\sqrt{2}$
(D) $26\sqrt{3}$

10. The value of

$$\begin{vmatrix} (a+1)(a+2) & a+2 & 1 \\ (a+2)(a+3) & a+3 & 1 \\ (a+3)(a+4) & a+4 & 1 \end{vmatrix}$$

is :

- (A) -2
(B) 0
(C) $(a+2)(a+3)(a+4)$
(D) $(a+1)(a+2)(a+3)$

11. If

$$\frac{\sin^{-1}x}{a} = \frac{\cos^{-1}x}{b} = \frac{\tan^{-1}y}{c}$$

;

$$0 < x < 1$$

,

then the value of

$$\cos\left(\frac{\pi c}{a+b}\right)$$

is :

- (A) $\frac{1-y^2}{2y}$
(B) $\frac{1-y^2}{y\sqrt{y}}$
(C) $1-y^2$
(D) $\frac{1-y^2}{1+y^2}$

12. The maximum slope of the curve

$$y = \frac{1}{2}x^4 - 5x^3 + 18x^2 - 19x$$

occurs at the point :

- (A) $\left(3, \frac{21}{2}\right)$
(B) $(0, 0)$
(C) $(2, 9)$
(D) $(2, 2)$

13. The rate of growth of bacteria in a culture is proportional to the number of bacteria present and the bacteria count is 1000 at initial time $t = 0$. The number of bacteria is increased by 20% in 2 hours. If the population of bacteria is 2000 after

$$\frac{k}{\log_e\left(\frac{6}{5}\right)}$$

hours, then

$$\left(\frac{k}{\log_e 2}\right)^2$$

is equal to :

- (A) 16
(B) 8
(C) 2
(D) 4

14. In an increasing geometric series, the sum of the second and the sixth term is $\frac{25}{2}$ and the product of the third and fifth term is 25. Then, the sum of 4th, 6th and 8th terms is equal to :

- (A) 30 (B) 32
(C) 26 (D) 35

15. Let f be any function defined on R and let it satisfy the condition :

$$|f(x) - f(y)| \leq |x - y|^2, \forall (x, y) \in R$$

If $f(0) = 1$, then :

- (A) $f(x)$ can take any value in R
(B) $f(x) < 0, \forall x \in R$
(C) $f(x) > 0, \forall x \in R$
(D) $f(x) = 0, \forall x \in R$

16. The sum of the infinite series

$$1 + \frac{2}{3} + \frac{7}{3^2} + \frac{12}{3^3} + \frac{17}{3^4} + \frac{22}{3^5} + \dots$$

is equal to :

- (A) $\frac{9}{4}$
(B) $\frac{13}{4}$
(C) $\frac{15}{4}$
(D) $\frac{11}{4}$

17. Consider the three planes

$$P_1 : 3x + 15y + 21z = 9,$$

$$P_2 : x - 3y - z = 5, \text{ and}$$

$$P_3 : 2x + 10y + 14z = 5$$

Then, which one of the following is true?

- (A) P_1 and P_2 are parallel. (B) P_1, P_2 and P_3 all are parallel.
(C) P_1 and P_3 are parallel. (D) P_2 and P_3 are parallel.

18. If $(1, 5, 35), (7, 5, 5), (1, \lambda, 7)$ and $(2\lambda, 1, 2)$ are coplanar, then the sum of all possible values of λ is :

- (A) $-\frac{44}{5}$
(B) $-\frac{39}{5}$
(C) $\frac{44}{5}$
(D) $\frac{39}{5}$

19. A fair coin is tossed a fixed number of times. If the probability of getting 7 heads is equal to probability of getting 9 heads, then the probability of getting 2 heads is :

- (A) $\frac{15}{2^8}$
(B) $\frac{15}{2^{12}}$
(C) $\frac{15}{2^{13}}$
(D) $\frac{15}{2^{14}}$

20. If $\vec{a}$ and $\vec{b}$ are perpendicular, then

$$\vec{a} \times \left(\vec{a} \times \left(\vec{a} \times \left(\vec{a} \times \vec{b} \right) \right) \right)$$

is equal to :

- (A) $\frac{1}{2} |\vec{a}|^4 \vec{b}$
(B) $\vec{0}$
(C) $\vec{a} \times \vec{b}$
(D) $|\vec{a}|^4 \vec{b}$

21. The value of the integral

$$\int_0^{\pi} |\sin 2x| dx$$

is _____.

22. The number of solutions of the equation $\log_4(x-1) = \log_2(x-3)$ is _____.

23. The sum of 162^{th} power of the roots of the equation $x^3 - 2x^2 + 2x - 1 = 0$ is _____.

24. The area bounded by the lines $y = ||x - 1| - 2|$ is _____.

25. The difference between degree and order of a differential equation that represents the family of curves given by

$$y^2 = a \left(x + \frac{\sqrt{a}}{2} \right)$$

, $a > 0$ is _____.

26. If $y = y(x)$ is the solution of the equation

$$e^{\sin y} \cos y \frac{dy}{dx} + e^{\sin y} \cos x = \cos x$$

, $y(0) = 0$; then

$$1 + y\left(\frac{\pi}{6}\right) + \frac{\sqrt{3}}{2}y\left(\frac{\pi}{3}\right) + \frac{1}{\sqrt{2}}y\left(\frac{\pi}{4}\right)$$

is equal to _____.

27. The number of integral values of 'k' for which the equation

$$3 \sin x + 4 \cos x = k + 1$$

has a solution, $k \in \mathbb{R}$ is _____.

28. If

$$\sqrt{3} (\cos^2 x) = (\sqrt{3} - 1) \cos x + 1$$

, the number of solutions of the given equation when

$$x \in \left[0, \frac{\pi}{2} \right]$$

is _____.

29. Let $(\lambda, 2, 1)$ be a point on the plane which passes through the point $(4, -2, 2)$. If the plane is perpendicular to the line joining the points $(-2, -21, 29)$ and $(-1, -16, 23)$, then

$$\left(\frac{\lambda}{11} \right)^2 - \frac{4\lambda}{11} - 4$$

is equal to _____.

30. Let $m, n \in \mathbb{N}$ and $\gcd(2, n) = 1$. If

$$30 \binom{30}{0} + 29 \binom{30}{1} + \dots + 2 \binom{30}{28} + 1 \binom{30}{29} = n \cdot 2^m$$

, then $n + m$ is equal to _____.

(Here

$$\binom{n}{k} = {}^nC_k$$

)

31. Consider the combination of 2 capacitors C_1 and C_2 with $C_2 > C_1$, when connected in parallel, the equivalent capacitance is $\frac{15}{4}$ times the equivalent capacitance of the same connected in series. Calculate the ratio of capacitors, $\frac{C_2}{C_1}$.

- (A) $\frac{15}{11}$
 (B) No Solutions
 (C) $\frac{29}{15}$
 (D) $\frac{15}{4}$

32. An alternating current is given by the equation $i = i_1 \sin \omega t + i_2 \cos \omega t$. The rms current will be :

- (A) $\frac{1}{\sqrt{2}}(i_1^2 + i_2^2)^{\frac{1}{2}}$
 (B) $\frac{1}{\sqrt{2}}(i_1 + i_2)$
 (C) $\frac{1}{\sqrt{2}}(i_1 + i_2)^2$
 (D) $\frac{1}{2}(i_1^2 + i_2^2)^{\frac{1}{2}}$

33. Four identical solid spheres each of mass 'm' and radius 'a' are placed with their centres on the four corners of a square of side 'b'. The moment of inertia of the system about one side of square where the axis of rotation is parallel to the plane of the square is :

- (A) $\frac{4}{5}ma^2$
 (B) $\frac{8}{5}ma^2 + mb^2$
 (C) $\frac{4}{5}ma^2 + 2mb^2$
 (D) $\frac{8}{5}ma^2 + 2mb^2$

34. A large number of water drops, each of radius r , combine to have a drop of radius R . If the surface tension is T and mechanical equivalent of heat is J , the rise in heat energy per unit volume will be :

- (A) $\frac{2T}{J}\left(\frac{1}{r} - \frac{1}{R}\right)$
 (B) $\frac{3T}{J}\left(\frac{1}{r} - \frac{1}{R}\right)$
 (C) $\frac{3T}{rJ}$
 (D) $\frac{2T}{rJ}$

35. LED is constructed from Ga-As-P semiconducting material. The energy gap of this LED is 1.9 eV. Calculate the wavelength of light emitted and its colour.

[$h = 6.63 \times 10^{-34}$ Js and $c = 3 \times 10^8$ ms $^{-1}$]

- (A) 654 nm and orange colour
 (B) 654 nm and red colour
 (C) 1046 nm and red colour
 (D) 1046 nm and blue colour

36. A short straight object of height 100 cm lies before the central axis of a spherical mirror whose focal length has absolute value $|f| = 40$ cm. The image of object produced by the mirror is of height 25 cm and has the same orientation of the object. One may conclude from the information :

- (A) Image is real, same side of convex mirror.
 (B) Image is virtual, opposite side of convex mirror.
 (C) Image is virtual, opposite side of concave mirror.
 (D) Image is real, same side of concave mirror.

37. In a Young's double slit experiment two slits are separated by 2 mm and the screen is placed one meter away. When a light of wavelength 500 nm is used, the fringe separation will be :

- (A) 0.50 mm (B) 0.25 mm
(C) 1 mm (D) 0.75 mm

38. Assume that a tunnel is dug along a chord of the earth, at a perpendicular distance ($R/2$) from the earth's centre, where 'R' is the radius of the Earth. The wall of the tunnel is frictionless. If a particle is released in this tunnel, it will execute a simple harmonic motion with a time period :

- (A) $2\pi\sqrt{\frac{R}{g}}$
(B) $\frac{g}{2\pi R}$
(C) $\frac{2\pi R}{g}$
(D) $\frac{1}{2\pi}\sqrt{\frac{g}{R}}$

39. In a typical combustion engine the workdone by a gas molecule is given by

$$W = \alpha^2 \beta e^{\frac{-\beta x^2}{kT}}$$

, where x is the displacement, k is the Boltzmann constant and T is the temperature. If α and β are constants, dimensions of α will be :

- (A) $[M^0 L T^0]$
(B) $[MLT^{-1}]$
(C) $[MLT^{-2}]$
(D) $[M^2 L T^{-2}]$

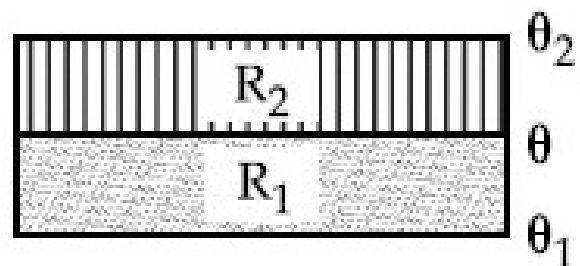
40. If two similar springs each of spring constant K_1 are joined in series, the new spring constant and time period would be changed by a factor :

- (A) $\frac{1}{2}, 2\sqrt{2}$
(B) $\frac{1}{4}, 2\sqrt{2}$
(C) $\frac{1}{2}, \sqrt{2}$
(D) $\frac{1}{4}, \sqrt{2}$

41. A particle is moving with uniform speed along the circumference of a circle of radius R under the action of a central fictitious force F which is inversely proportional to R^3 . Its time period of revolution will be given by :

- (A) $T \propto R^{\frac{4}{3}}$
(B) $T \propto R^{\frac{5}{2}}$
(C) $T \propto R^{\frac{3}{2}}$
(D) $T \propto R^2$

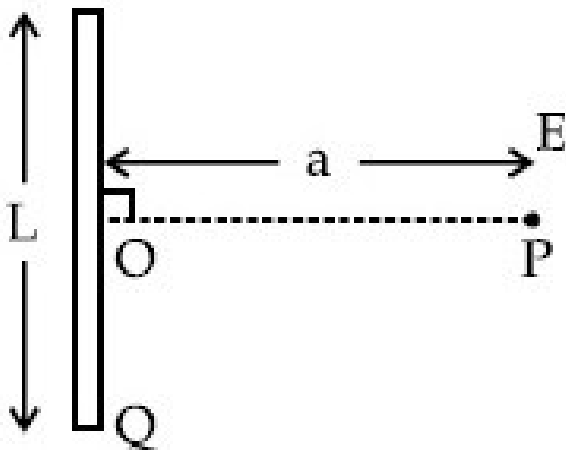
42. The temperature θ at the junction of two insulating sheets, having thermal resistances R_1 and R_2 as well as top and bottom temperatures θ_1 and θ_2 (as shown in figure) is given by :



- (A) $\frac{\theta_1 R_2 + \theta_2 R_1}{R_1 + R_2}$
(B) $\frac{\theta_1 R_1 + \theta_2 R_2}{R_1 + R_2}$
(C) $\frac{\theta_1 R_2 - \theta_2 R_1}{R_2 - R_1}$
(D) $\frac{\theta_2 R_2 - \theta_1 R_1}{R_2 - R_1}$

43. Find the electric field at point P (as shown in figure) on the perpendicular bisector of a uniformly charged thin wire of length L carrying a charge Q. The distance of the point P from the centre of the rod is $a =$

$$\frac{\sqrt{3}}{2}L$$



- (A) $\frac{Q}{4\pi\epsilon_0 L^2}$
 (B) $\frac{Q}{3\pi\epsilon_0 L^2}$
 (C) $\frac{Q}{2\sqrt{3}\pi\epsilon_0 L^2}$
 (D) $\frac{\sqrt{3}Q}{4\pi\epsilon_0 L^2}$

44. A planet revolving in elliptical orbit has :
 A. a constant velocity of revolution.
 B. has the least velocity when it is nearest to the sun.
 C. its areal velocity is directly proportional to its velocity.
 D. areal velocity is inversely proportional to its velocity.
 E. to follow a trajectory such that the areal velocity is constant.
 Choose the correct answer from the options given below :

- (A) D only (B) E only
 (C) C only (D) A only

45. Given below are two statements : one is labeled as Assertion A and the other is labelled as Reason R.

Assertion A : An electron microscope can achieve better resolving power than an optical microscope.

Reason R : The de Broglie's wavelength of the electrons emitted from an electron gun is much less than wavelength of visible light.

In the light of the above statements, choose the correct answer from the options given below :

- (A) A is false but R is true. (B) Both A and R are true but R is NOT the correct explanation of A.
 (C) Both A and R are true and R is the correct explanation of A. (D) A is true but R is false.

46. Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : Body 'P' having mass M moving with speed 'u' has head-on collision elastically with another body 'Q' having mass 'm' initially at rest. If $m \ll M$, body 'Q' will have a maximum speed equal to '2u' after collision.

Reason R : During elastic collision, the momentum and kinetic energy are both conserved.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (A) A is correct but R is not correct. (B) A is not correct but R is correct.
 (C) Both A and R are correct and R is the correct explanation of A. (D) Both A and R are correct but R is NOT the correct explanation of A.

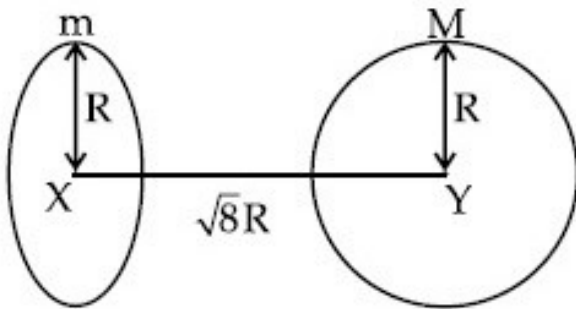
47. The normal density of a material is ρ and its bulk modulus of elasticity is K . The magnitude of increase in density of material, when a pressure P is applied uniformly on all sides, will be :

- (A) $\frac{\rho K}{P}$
 (B) $\frac{PK}{\rho}$
 (C) $\frac{\rho P}{K}$
 (D) $\frac{K}{\rho P}$

48. If λ_1 and λ_2 are the wavelengths of the third member of Lyman and first member of the Paschen series respectively, then the value of $\lambda_1 : \lambda_2$ is :

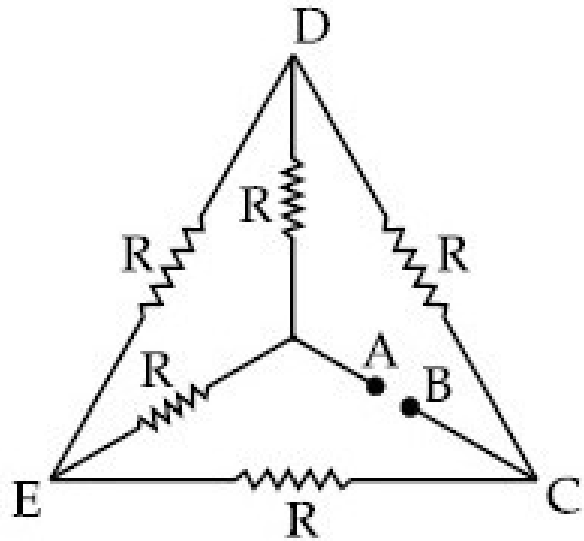
- (A) 7 : 135
 (B) 7 : 108
 (C) 1 : 9
 (D) 1 : 3

49. Find the gravitational force of attraction between the ring and sphere as shown in the diagram, where the plane of the ring is perpendicular to the line joining the centres. If $\sqrt{8}R$ is the distance between the centres of a ring (of mass 'm') and a sphere (mass 'M') where both have equal radius 'R'.



- (A) $\frac{2\sqrt{2}}{3} \cdot \frac{GMm}{R^2}$
 (B) $\frac{\sqrt{8}}{9} \cdot \frac{GmM}{R}$
 (C) $\frac{\sqrt{8}}{27} \cdot \frac{GmM}{R^2}$
 (D) $\frac{1}{3\sqrt{8}} \cdot \frac{GMm}{R^2}$

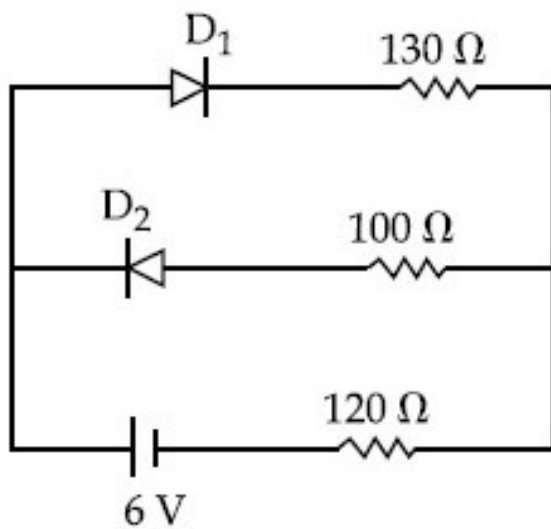
50. Five equal resistances are connected in a network as shown in figure. The net resistance between the points A and B is :



- (A) $\frac{3R}{2}$
 (B) $\frac{R}{2}$
 (C) $2R$
 (D) R

51. A person standing on a spring balance inside a stationary lift measures 60 kg. The weight of that person if the lift descends with uniform downward acceleration of 1.8 m/s^2 will be _____ N. [$g = 10 \text{ m/s}^2$]

52. The circuit contains two diodes each with a forward resistance of 50Ω and with infinite reverse resistance. If the battery voltage is 6V, the current through the 120Ω resistance is _____ mA.



53. In a series LCR resonant circuit, the quality factor is measured as 100. If the inductance is increased by two fold and resistance is decreased by two fold, then the quality factor after this change will be _____.

54. The mass per unit length of a uniform wire is 0.135 g/cm . A transverse wave of the form $y = -0.21 \sin(x + 30t)$ is produced in it, where x is in meter and t is in second. Then, the expected value of tension in the wire is $x \times 10^{-2} \text{ N}$. Value of x is _____. (Round off to the nearest integer)

55. In an electrical circuit, a battery is connected to pass 20C of charge through it in a certain given time. The potential difference between two plates of the battery is maintained at 15V. The workdone by the battery is _____ J.

56. A boy pushes a box of mass 2 kg with a force $\vec{F} = (20\hat{i} + 10\hat{j})N$ on a frictionless surface. If the box was initially at rest, then _____ m is displacement along the x-axis after 10s.

57. A radiation is emitted by 1000W bulb and it generates an electric field and magnetic field at P placed at a distance of 2m. The efficiency of the bulb is 1.25%. The value of peak electric field at P is $x \times 10^{-1}$ V/m. Value of x is _____. (Rounded off to the nearest integer) [Take

$$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2\text{N}^{-1} \text{ m}^{-2}, c = 3 \times 10^8 \text{ ms}^{-1}]$$

58. A container is divided into two chambers by a partition. The volume of first chamber is 4.5 litre and second chamber is 5.5 litre. The first chamber contain 3.0 moles of gas at pressure 2.0 atm and second chamber contain 4.0 moles of gas at pressure 3.0 atm. After the partition is removed and the mixture attains equilibrium, then, the common equilibrium pressure existing in the mixture is $x \times 10^{-1}$ atm. Value of x is _____.

59. As shown in the figure, a block of mass $\sqrt{3}$ kg is kept on a horizontal rough surface of coefficient of friction

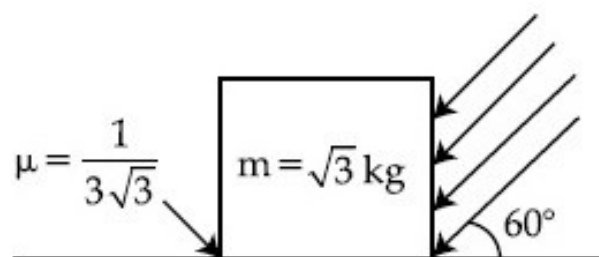
$$\frac{1}{3\sqrt{3}}$$

. The critical force to be applied on the vertical surface as shown at an angle 60° with horizontal such that it does not move, will be $3x$. The value of x will be _____.

$$[g = 10 \text{ m/s}^2; \sin 60^\circ =$$

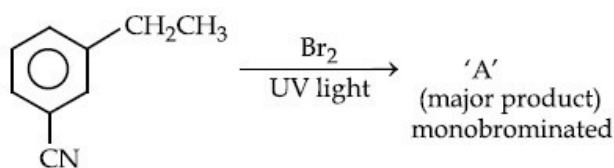
$$\frac{\sqrt{3}}{2}$$

$$; \cos 60^\circ = \frac{1}{2}]$$

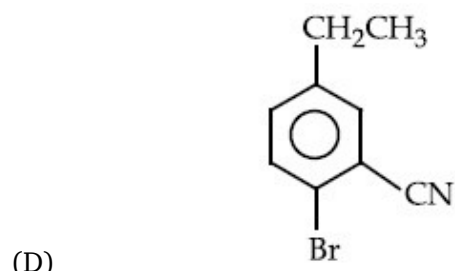
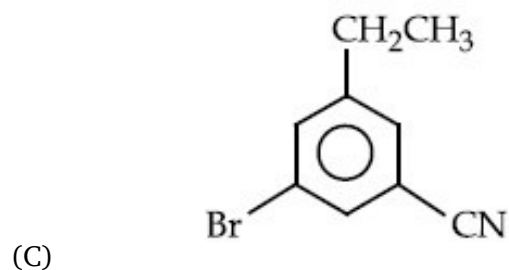
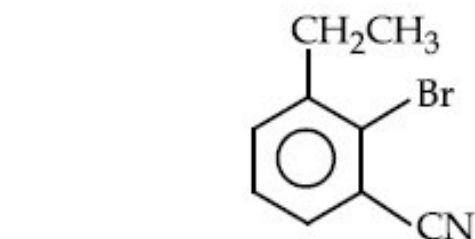
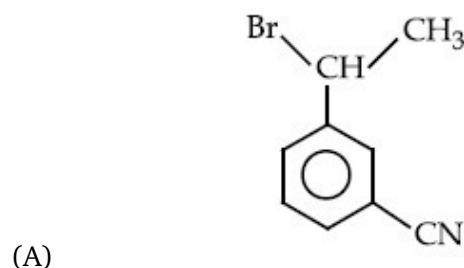


60. The maximum and minimum amplitude of an amplitude modulated wave is 16V and 8V respectively. The modulation index for this amplitude modulated wave is $x \times 10^{-2}$. The value of x is _____.

61. For the given reaction



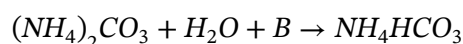
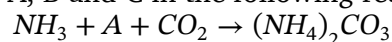
What is 'A'?



62. Which of the following is 'a' FALSE statement ?

- (A) Carius method is used for the estimation of nitrogen in an organic compound. (B) Kjeldahl's method is used for the estimation of nitrogen in an organic compound.
- (C) Carius tube is used in the estimation of sulphur in an organic compound. (D) Phosphoric acid produced on oxidation of phosphorus present in an organic compound is precipitated as $\text{Mg}_2\text{P}_2\text{O}_7$ by adding magnesia mixture.

63. Find A, B and C in the following reactions :



- (A) $\text{A} - \text{O}_2; \text{B} - \text{CO}_2; \text{C} - \text{Na}_2\text{CO}_3$
- (B) $\text{A} - \text{H}_2\text{O}; \text{B} - \text{O}_2; \text{C} - \text{NaHCO}_3$
- (C) $\text{A} - \text{H}_2\text{O}; \text{B} - \text{O}_2; \text{C} - \text{Na}_2\text{CO}_3$
- (D) $\text{A} - \text{H}_2\text{O}; \text{B} - \text{CO}_2; \text{C} - \text{NaHCO}_3$

64. Match List - I with List - II.

	List I Electronic configuration		List II $\Delta_i H$ in kJ mol^{-1}
(a)	$1s^2 2s^2$	(i)	801
(b)	$1s^2 2s^2 2p^4$	(ii)	899
(c)	$1s^2 2s^2 2p^3$	(iii)	1314
(d)	$1s^2 2s^2 2p^1$	(iv)	1402

Choose the most appropriate answer from the options given below :

- (A) (a) $\rightarrow$ (i), (b) $\rightarrow$ (iii), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (iii) (B) (a) $\rightarrow$ (iv), (b) $\rightarrow$ (i), (c) $\rightarrow$ (ii), (d) $\rightarrow$ (iii)
- (C) (a) $\rightarrow$ (ii), (b) $\rightarrow$ (iii), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (i) (D) (a) $\rightarrow$ (i), (b) $\rightarrow$ (iv), (c) $\rightarrow$ (iii), (d) $\rightarrow$ (ii)

65. Given below are two statements :

Statement I : A mixture of chloroform and aniline can be separated by simple distillation.

Statement II : When separating aniline from a mixture of aniline and water by steam distillation aniline boils below its boiling point.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|---|---|
| (A) Both Statement I and Statement II are true | (B) Both Statement I and Statement II are false |
| (C) Statement I is true but Statement II is false | (D) Statement I is false but Statement II is true |

66. Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : Dipole-dipole interactions are the only non-covalent interactions, resulting in hydrogen bond formation.

Reason R : Fluorine is the most electronegative element and hydrogen bonds in HF are symmetrical.

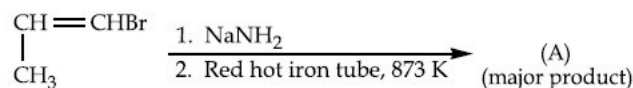
In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|---|---|
| (A) Both A and R are true and R is the correct explanation of A | (B) A is true but R is false |
| (C) A is false but R is true | (D) Both A and R are true but R is NOT the correct explanation of A |

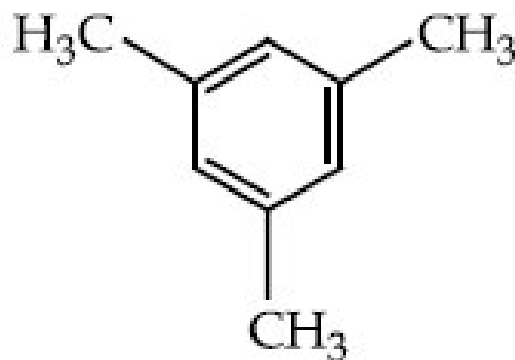
67. Which one of the following lanthanoids does not form MO_2 ? [M is lanthanoid metal]

- | | |
|--------|--------|
| (A) Nd | (B) Dy |
| (C) Yb | (D) Pr |

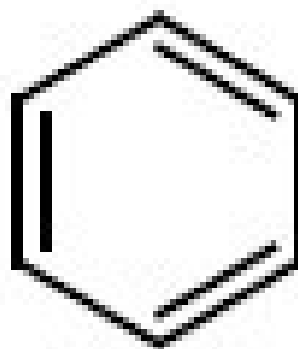
68. For the given reaction :



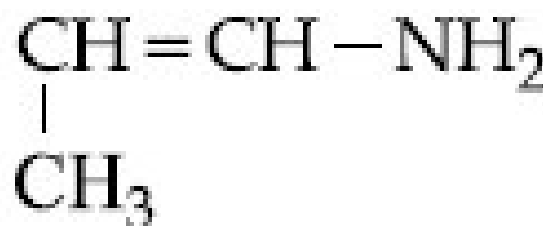
What is 'A'?



(A)



(B)



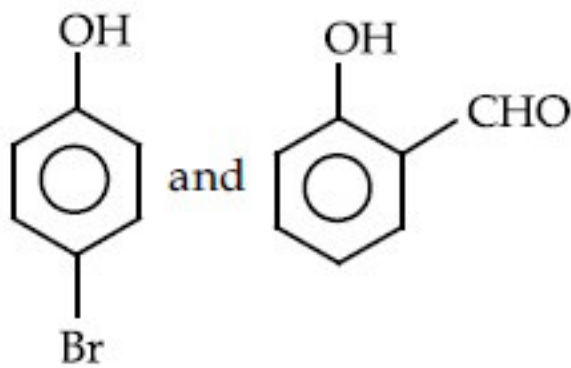
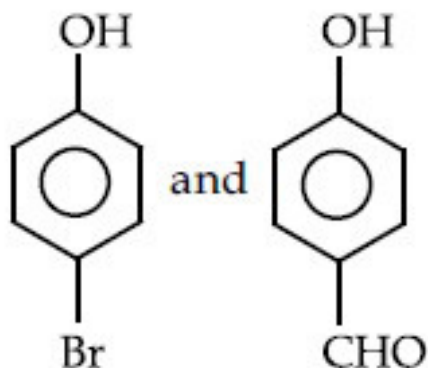
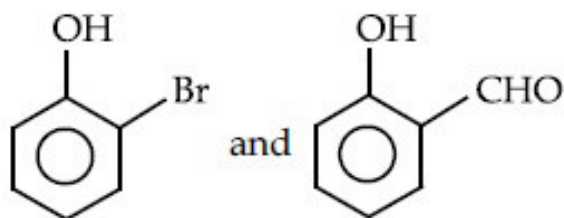
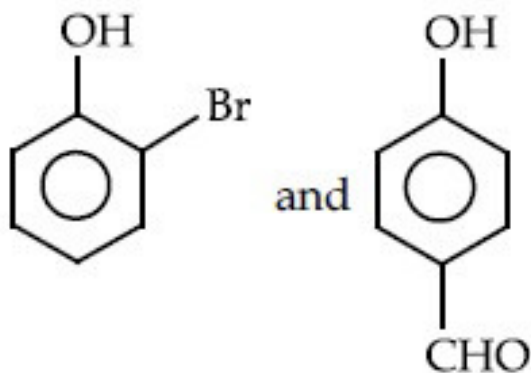
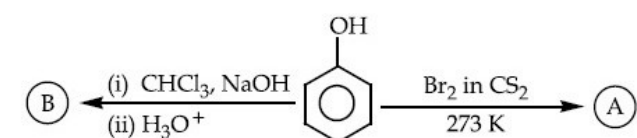
(C)

(D) $\text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2$

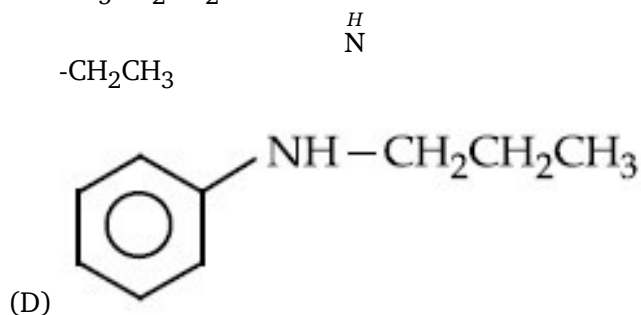
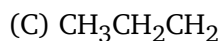
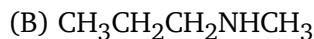
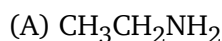
69. Compound A used as a strong oxidizing agent is amphoteric in nature. It is the part of lead storage batteries. Compound A is :

- | | |
|---------------------|-----------------------------|
| (A) PbO | (B) PbO_2 |
| (C) PbSO_4 | (D) Pb_3O_4 |

70. Identify the major products A and B respectively in the following reactions of phenol :



71. An amine on reaction with benzenesulphonyl chloride produces a compound insoluble in alkaline solution. This amine can be prepared by ammonolysis of ethyl chloride. The correct structure of amine is :



72. Given below are two statements :

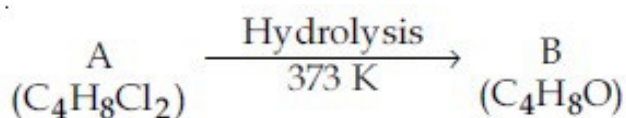
Statement I : o-Nitrophenol is steam volatile due to intramolecular hydrogen bonding.

Statement II : o-Nitrophenol has high melting due to hydrogen bonding.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|---|---|
| (A) Both Statement I and Statement II are true | (B) Statement I is true but Statement II is false |
| (C) Statement I is false but Statement II is true | (D) Both Statement I and Statement II are false |

73.



B reacts with Hydroxyl amine but does not give Tollen's test. Identify A and B.

- | | |
|--|---|
| (A) 1, 1-Dichlorobutane and Butanal | (B) 2, 2-Dichlorobutane and Butan-2-one |
| (C) 1, 1-Dichlorobutane and 2-Butanone | (D) 2, 2-Dichlorobutane and Butanal |

74. Which of the following vitamin is helpful in delaying the blood clotting?

- | | |
|---------------|---------------|
| (A) Vitamin K | (B) Vitamin C |
| (C) Vitamin E | (D) Vitamin B |

75. The orbital having two radial as well as two angular nodes is :

- | | |
|--------|--------|
| (A) 3p | (B) 5d |
| (C) 4d | (D) 4f |

76. Statements about heavy water are given below :

A. Heavy water is used in exchange reactions for the study of reaction mechanisms.

B. Heavy water is prepared by exhaustive electrolysis of water.

(C) Heavy water has higher boiling point than ordinary water.

(D) Viscosity of H_2O is greater than D_2O .

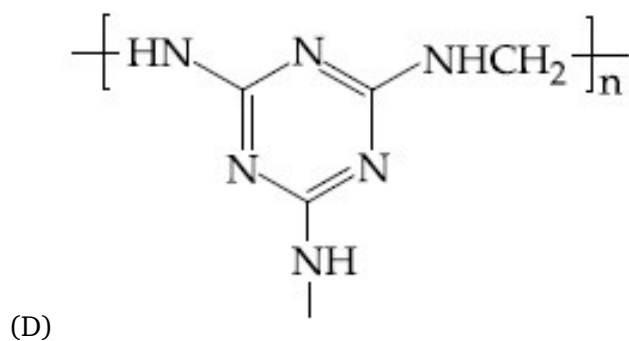
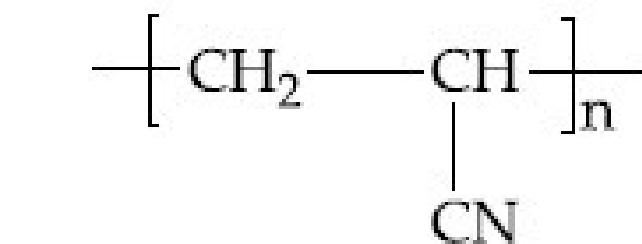
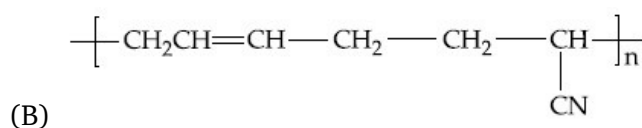
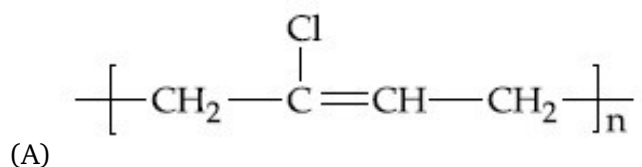
Choose the most appropriate answer from the options given below :

- | | |
|----------------------|------------------|
| (A) A, B, and C only | (B) A and C only |
| (C) A and D only | (D) A and B only |

77. On treating a compound with warm dil. H_2SO_4 , gas X is evolved which turns $\text{K}_2\text{Cr}_2\text{O}_7$ paper acidified with dil. H_2SO_4 to a green compound Y. X and Y respectively are :

- | | |
|---|--|
| (A) $\text{X} = \text{SO}_2$, $\text{Y} = \text{Cr}_2\text{O}_3$ | (B) $\text{X} = \text{SO}_3$, $\text{Y} = \text{Cr}_2(\text{SO}_4)_3$ |
| (C) $\text{X} = \text{SO}_3$, $\text{Y} = \text{Cr}_2\text{O}_3$ | (D) $\text{X} = \text{SO}_2$, $\text{Y} = \text{Cr}_2(\text{SO}_4)_3$ |

78. The structure of Neoprene is :



79. Match List - I with List - II.

	List - I (Ore)		List - II (Element Present)
(a)	Kernite	(i)	Tin
(b)	Cassiterite	(ii)	Boron
(c)	Calamine	(iii)	Fluorine
(d)	Cryolite	(iv)	Zinc

Choose the most appropriate answer from the options given below :

- (A) (a) → (i), (b) → (iii), (c) → (iv), (d) → (ii)
 (B) (a) → (iii), (b) → (i), (c) → (ii), (d) → (iv)
 (C) (a) → (ii), (b) → (i), (c) → (iv), (d) → (iii)
 (D) (a) → (ii), (b) → (iv), (c) → (i), (d) → (iii)

80. The presence of ozone in troposphere :

- (A) protects us from greenhouse effect
 (B) protects us from the UV radiation
 (C) protects us from the X-ray radiation
 (D) generates photochemical smog

81. For a chemical reaction $A + B \rightleftharpoons C + D$ ($\Delta_r H^\ominus = 80 \text{ kJ mol}^{-1}$) the entropy change $\Delta_r S^\ominus$ depends on the temperature T (in K) as $\Delta_r S^\ominus = 2T \text{ (J K}^{-1}\text{mol}^{-1}\text{)}$.

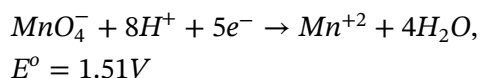
Minimum temperature at which it will become spontaneous is _____ K. (Integer)

82. An exothermic reaction $X \rightarrow Y$ has an activation energy 30 kJ mol^{-1} . If energy change ΔE during the reaction is -20 kJ , then the activation energy for the reverse reaction in kJ is _____. (Integer answer)

83. Dichromate ion is treated with base, the oxidation number of Cr in the product formed is _____.

84. Number of bridging CO ligands in $[\text{Mn}_2(\text{CO})_{10}]$ is _____.

85. Consider the following reaction



The quantity of electricity required in Faraday to reduce five moles of MnO_4^- is _____. (Integer answer)

86. 224 mL of $\text{SO}_{2(g)}$ at 298 K and 1 atm is passed through 100 mL of 0.1 M NaOH solution. The non-volatile solute produced is dissolved in 36g of water. The lowering of vapour pressure of solution (assuming the solution in dilute) ($P_{(\text{H}_2\text{O})}^\ominus - 24 \text{ mm of Hg}$) is $x \times 10^{-2} \text{ mm of Hg}$, the value of x is _____. (Integer answer)

87. A homogeneous ideal gaseous reaction $\text{AB}_{2(g)} \rightleftharpoons \text{A}_{(g)} + 2\text{B}_{(g)}$ is carried out in a 25 litre flask at 27°C . The initial amount of AB_2 was 1 mole and the equilibrium pressure was 1.9 atm. The value of K_p is $x \times 10^{-2}$. The value of x is _____. (Integer answer)

[$R = 0.08206 \text{ dm}^3\text{atm K}^{-1}\text{mol}^{-1}$]

88. A certain gas obeys $P(V_m - b) = RT$. The value of

$$\left(\frac{\partial Z}{\partial P}\right)_T$$

is $\frac{xb}{RT}$. The value of x is _____. (Integer answer) (Z : compressibility factor)

89. The number of significant figures in 50000.020×10^{-3} is _____.

90. 3.12 g of oxygen is adsorbed on 1.2 g of platinum metal. The volume of oxygen adsorbed per gram of the adsorbent at 1 atm and 300 K in L is _____.

[$R = 0.0821 \text{ L atm K}^{-1}\text{mol}^{-1}$]

- End of Question Paper -