

JEE Main 2021 Paper 1 (B.E./B.Tech) 27 August 2021 Shift 1 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2021 - Session 4**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 27 August 2021, Shift 1. The total duration is **3 hours**.
- (ii) The paper has **90 questions**, of which **75 are to be attempted**, and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 30 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **10 numerical value questions**; attempt **any 5** of them.
- (iv) **All Section A questions are compulsory**. Attempt only the specified number of Section B questions. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme**: MCQs carry **+4** for a correct answer and **-1** for an incorrect answer. Numerical questions carry **+4** for a correct answer and **0** for an incorrect answer. Unanswered questions carry **0**.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

Mathematics - Section A · Multiple Choice Questions

1. Let A be a fixed point (0, 6) and B be a moving point (2t, 0). Let M be the mid-point of AB and the perpendicular bisector of AB meets the y-axis at C. The locus of the mid-point P of MC is :

- (A) $3x^2 - 2y - 6 = 0$ (B) $3x^2 + 2y - 6 = 0$
 (C) $2x^2 + 3y - 9 = 0$ (D) $2x^2 - 3y + 9 = 0$

2. If

$$(\sin^{-1}x)^2 - (\cos^{-1}x)^2 = a$$

; $0 < x < 1$, $a \neq 0$, then the value of $2x^2 - 1$ is :

- (A) $\cos\left(\frac{4a}{\pi}\right)$
 (B) $\sin\left(\frac{2a}{\pi}\right)$
 (C) $\cos\left(\frac{2a}{\pi}\right)$
 (D) $\sin\left(\frac{4a}{\pi}\right)$

3. If the matrix

$$A = \begin{pmatrix} 0 & 2 \\ K & -1 \end{pmatrix}$$

satisfies

$$A(A^3 + 3I) = 2I$$

, then the value of K is :

- (A) $\frac{1}{2}$
 (B) $-\frac{1}{2}$
 (C) -1
 (D) 1

4. If

$$S = \left\{ z \in \mathbb{C} : \frac{z-i}{z+2i} \in \mathbb{R} \right\}$$

, then :

- (A) S contains exactly two elements (B) S contains only one element
 (C) S is a circle in the complex plane (D) S is a straight line in the complex plane

5. Let $y = y(x)$ be the solution of the differential equation

$$\frac{dy}{dx} = 2(y + 2 \sin x - 5)x - 2 \cos x$$

such that $y(0) = 7$. Then $y(\pi)$ is equal to :

- (A) $2e^{\pi^2} + 5$
 (B) $e^{\pi^2} + 5$
 (C) $3e^{\pi^2} + 5$
 (D) $7e^{\pi^2} + 5$

6. Let us consider a curve, $y = f(x)$ passing through the point $(-2, 2)$ and the slope of the tangent to the curve at any point $(x, f(x))$ is given by $f(x) + xf(x) = x^2$. Then :

- (A) $x^2 + 2xf(x) - 12 = 0$
 (B) $x^3 + xf(x) + 12 = 0$
 (C) $x^3 - 3xf(x) - 4 = 0$
 (D) $x^2 + 2xf(x) + 4 = 0$

7. If α, β are the distinct roots of $x^2 + bx + c = 0$, then

$$\lim_{x \rightarrow \beta} \frac{e^{2(x^2 + bx + c)} - 1 - 2(x^2 + bx + c)}{(x - \beta)^2}$$

is equal to :

- (A) $b^2 + 4c$ (B) $2(b^2 + 4c)$
 (C) $2(b^2 - 4c)$ (D) $b^2 - 4c$

8. When a certain biased die is rolled, a particular face occurs with probability

$$\frac{1}{6} - x$$

and its opposite face occurs with probability

$$\frac{1}{6} + x$$

. All other faces occur with probability $\frac{1}{6}$. Note that opposite faces sum to 7 in any die. If $0 < x < \frac{1}{6}$, and the probability of obtaining total sum = 7, when such a die is rolled twice, is $\frac{13}{96}$, then the value of x is :

- (A) $\frac{1}{16}$
 (B) $\frac{1}{8}$
 (C) $\frac{1}{9}$
 (D) $\frac{1}{12}$

9. If $x^2 + 9y^2 - 4x + 3 = 0$, $x, y \in \mathbb{R}$, then x and y respectively lie in the intervals :

(A) $\left[-\frac{1}{3}, \frac{1}{3}\right]$

and

$$\left[-\frac{1}{3}, \frac{1}{3}\right]$$

(B) $\left[-\frac{1}{3}, \frac{1}{3}\right]$

and $[1, 3]$

(C) $[1, 3]$ and $[1, 3]$

(D) $[1, 3]$ and

$$\left[-\frac{1}{3}, \frac{1}{3}\right]$$

10.

$$\int_6^{16} \frac{\log_e x^2}{\log_e x^2 + \log_e (x^2 - 44x + 484)} dx$$

is equal to :

- (A) 6 (B) 8
 (C) 5 (D) 10

11. A wire of length 20 m is to be cut into two pieces. One of the pieces is to be made into a square and the other into a regular hexagon. Then the length of the side (in meters) of the hexagon, so that the combined area of the square and the hexagon is minimum, is :

- (A) $\frac{5}{2 + \sqrt{3}}$
 (B) $\frac{10}{2 + 3\sqrt{3}}$
 (C) $\frac{5}{3 + \sqrt{3}}$
 (D) $\frac{10}{3 + 2\sqrt{3}}$

12. If $0 < x < 1$, then

$$\frac{3}{2}x^2 + \frac{5}{3}x^3 + \frac{7}{4}x^4 + \dots$$

, is equal to :

- (A) $x\left(\frac{1+x}{1-x}\right) + \log_e(1-x)$
 (B) $x\left(\frac{1-x}{1+x}\right) + \log_e(1-x)$
 (C) $\frac{1-x}{1+x} + \log_e(1-x)$
 (D) $\frac{1+x}{1-x} + \log_e(1-x)$

13. If for $x, y \in \mathbb{R}$, $x > 0$, $y = \log_{10}x + \log_{10}x^{1/3} + \log_{10}x^{1/9} + \dots$ upto ∞ terms and

$$\frac{2 + 4 + 6 + \dots + 2y}{3 + 6 + 9 + \dots + 3y} = \frac{4}{\log_{10}x}$$

, then the ordered pair (x, y) is equal to :

- (A) $(10^6, 6)$ (B) $(10^4, 6)$
 (C) $(10^2, 3)$ (D) $(10^6, 9)$

14. The distance of the point $(1, -2, 3)$ from the plane $x - y + z = 5$ measured parallel to a line, whose direction ratios are 2, 3, -6 is :

- (A) 3 (B) 5
(C) 2 (D) 1

15. Equation of a plane at a distance

$$\sqrt{\frac{2}{21}}$$

from the origin, which contains the line of intersection of the planes $x - y - z - 1 = 0$ and $2x + y - 3z + 4 = 0$, is :

- (A) $3x - y - 5z + 2 = 0$
(B) $3x - 4z + 3 = 0$
(C) $-x + 2y + 2z - 3 = 0$
(D) $4x - y - 5z + 2 = 0$

16. If

$$U_n = \left(1 + \frac{1}{n^2}\right) \left(1 + \frac{2^2}{n^2}\right)^2 \dots \left(1 + \frac{n^2}{n^2}\right)^n$$

, then

$$\lim_{n \rightarrow \infty} (U_n)^{\frac{-4}{n^2}}$$

is equal to :

- (A) $\frac{e^2}{16}$
(B) $\frac{4}{e}$
(C) $\frac{16}{e^2}$
(D) $\frac{4}{e^2}$

17. The statement $(p \wedge (p \rightarrow q) \wedge (q \rightarrow r)) \rightarrow r$ is :

- (A) a tautology (B) equivalent to $p \rightarrow \sim r$
(C) a fallacy (D) equivalent to $q \rightarrow \sim r$

18.

$$\sum_{k=0}^{20} ({}^{20}C_k)^2$$

is equal to :

- (A) ${}^{40}C_{21}$
(B) ${}^{40}C_{19}$
(C) ${}^{40}C_{20}$
(D) ${}^{41}C_{20}$

19. A tangent and a normal are drawn at the point $P(2, -4)$ on the parabola $y^2 = 8x$, which meet the directrix of the parabola at the points A and B respectively. If $Q(a, b)$ is a point such that AQBP is a square, then $2a + b$ is equal to :

- (A) -16 (B) -18
(C) -12 (D) -20

20. Let

$$\frac{\sin A}{\sin B} = \frac{\sin(A - C)}{\sin(C - B)}$$

, where A, B, C are angles of triangle ABC. If the lengths of the sides opposite these angles are a, b, c respectively, then :

- (A) $b^2 - a^2 = a^2 + c^2$ (B) b^2, c^2, a^2 are in A.P.
(C) c^2, a^2, b^2 are in A.P. (D) a^2, b^2, c^2 are in A.P.

21. Let

$$\vec{a} = \hat{i} + 5\hat{j} + \alpha\hat{k}$$

,

$$\vec{b} = \hat{i} + 3\hat{j} + \beta\hat{k}$$

and

$$\vec{c} = -\hat{i} + 2\hat{j} - 3\hat{k}$$

be three vectors such that,

$$|\vec{b} \times \vec{c}| = 5\sqrt{3}$$

and $\vec{a}$ is perpendicular to $\vec{b}$. Then the greatest amongst the values of $|\vec{a}|^2$ is _____.

22. The number of distinct real roots of the equation $3x^4 + 4x^3 - 12x^2 + 4 = 0$ is _____.

23. If $A = \{x \in \mathbb{R} : |x - 2| > 1\}$,

$B = \{x \in \mathbb{R} :$

$$\sqrt{x^2 - 3}$$

$> 1\}$,

$C = \{x \in \mathbb{R} : |x - 4| \geq 2\}$ and Z is the set of all integers, then the number of subsets of the set $(A \cap B \cap C)^c \cap Z$ is _____.

24. If

$$\int \frac{dx}{(x^2 + x + 1)^2} = a \tan^{-1}\left(\frac{2x+1}{\sqrt{3}}\right) + b \left(\frac{2x+1}{x^2+x+1}\right) + C$$

, $x > 0$ where C is the constant of integration, then the value of $9(\sqrt{3}a + b)$ is equal to _____.

25. If the system of linear equations

$$2x + y - z = 3$$

$$x - y - z = \alpha$$

$$3x + 3y + \beta z = 3$$

has infinitely many solution, then $\alpha + \beta - \alpha\beta$ is equal to _____.

26. Let n be an odd natural number such that the variance of 1, 2, 3, 4, ..., n is 14. Then n is equal to _____.

27. A number is called a palindrome if it reads the same backward as well as forward. For example 285582 is a six digit palindrome. The number of six digit palindromes, which are divisible by 55, is _____.

28. Let the equation $x^2 + y^2 + px + (1 - p)y + 5 = 0$ represent circles of varying radius $r \in (0, 5]$. Then the number of elements in the set $S = \{q : q = p^2 \text{ and } q \text{ is an integer}\}$ is _____.

29. If the minimum area of the triangle formed by a tangent to the ellipse

$$\frac{x^2}{b^2} + \frac{y^2}{4a^2} = 1$$

and the co-ordinate axis is kab , then k is equal to _____.

30. If

$$y^{1/4} + y^{-1/4} = 2x$$

, and

$$(x^2 - 1)\frac{d^2y}{dx^2} + \alpha x \frac{dy}{dx} + \beta y = 0$$

, then $|\alpha - \beta|$ is equal to _____.

31. A uniformly charged disc of radius R having surface charge density σ is placed in the xy plane with its center at the origin. Find the electric field intensity along the z -axis at a distance Z from origin :-

- (A) $E = \frac{\sigma}{2\epsilon_0} \left(1 - \frac{Z}{(Z^2 + R^2)^{1/2}} \right)$
- (B) $E = \frac{\sigma}{2\epsilon_0} \left(1 + \frac{Z}{(Z^2 + R^2)^{1/2}} \right)$
- (C) $E = \frac{2\epsilon_0}{\sigma} \left(\frac{1}{(Z^2 + R^2)^{1/2}} + Z \right)$
- (D) $E = \frac{\sigma}{2\epsilon_0} \left(\frac{1}{(Z^2 + R^2)} + \frac{1}{Z^2} \right)$

32. Which of the following is not a dimensionless quantity?

- (A) Relative magnetic permeability (μ_r)
- (B) Power factor
- (C) Permeability of free space (μ_0)
- (D) Quality factor

33. If E and H represents the intensity of electric field and magnetising field respectively, then the unit of E/H will be :

- (A) ohm
- (B) mho
- (C) joule
- (D) newton

34. The resultant of these forces $\vec{OP}$, $\vec{OQ}$, $\vec{OR}$, $\vec{OS}$ and $\vec{OT}$ is approximately N.

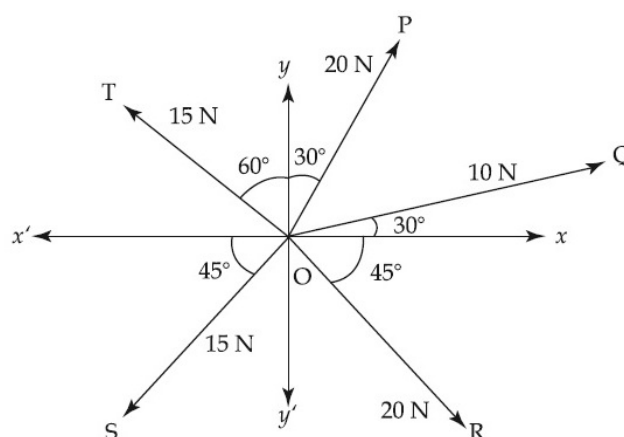
[Take

$$\sqrt{3} = 1.7$$

,

$$\sqrt{2} = 1.4$$

Given $\hat{i}$ and $\hat{j}$ unit vectors along x , y axis]



- (A) $9.25\hat{i} + 5\hat{j}$
- (B) $3\hat{i} + 15\hat{j}$
- (C) $2.5\hat{i} - 14.5\hat{j}$
- (D) $-1.5\hat{i} - 15.5\hat{j}$

35. A balloon carries a total load of 185 kg at normal pressure and temperature of 27°C . What load will the balloon carry on rising to a height at which the barometric pressure is 45 cm of Hg and the temperature is -7°C . Assuming the volume constant?

- (A) 181.46 kg
- (B) 214.15 kg
- (C) 219.07 kg
- (D) 123.54 kg

36. An object is placed beyond the centre of curvature C of the given concave mirror. If the distance of the object is d_1 from C and the distance of the image formed is d_2 from C, the radius of curvature of this mirror is :

- (A) $\frac{2d_1d_2}{d_1 - d_2}$
 (B) $\frac{2d_1d_2}{d_1 + d_2}$
 (C) $\frac{d_1d_2}{d_1 + d_2}$
 (D) $\frac{d_1d_2}{d_1 - d_2}$

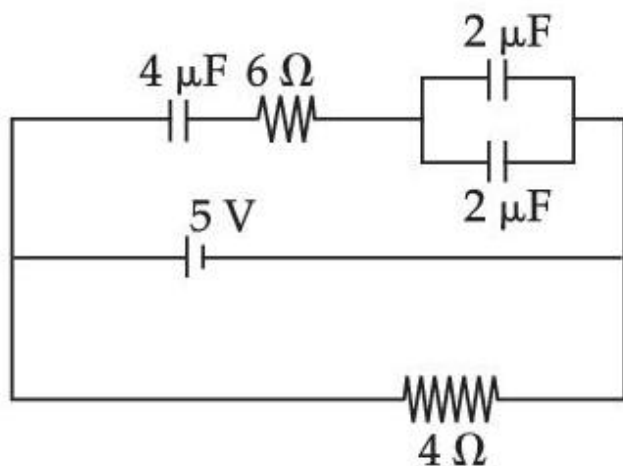
37. A huge circular arc of length 4.4 ly subtends an angle '4s' at the centre of the circle. How long it would take for a body to complete 4 revolution if its speed is 8 AU per second?

Given : 1 ly = 9.46×10^{15} m

1 AU = 1.5×10^{11} m

- (A) 4.1×10^8 s (B) 4.5×10^{10} s
 (C) 3.5×10^6 s (D) 7.2×10^8 s

38. Calculate the amount of charge on capacitor of $4\mu\text{F}$. The internal resistance of battery is 1Ω :



- (A) $8\mu\text{C}$ (B) zero
 (C) $16\mu\text{C}$ (D) $4\mu\text{C}$

39. Moment of inertia of a square plate of side l about the axis passing through one of the corner and perpendicular to the plane of square plate is given by :

- (A) $\frac{Ml^2}{6}$
 (B) Ml^2
 (C) $\frac{Ml^2}{12}$
 (D) $\frac{2}{3}Ml^2$

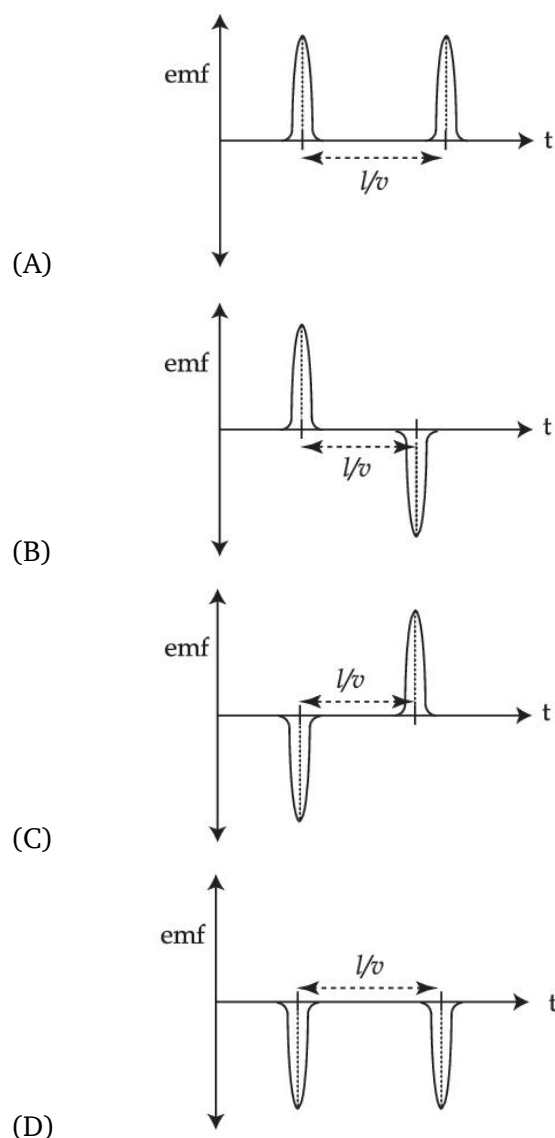
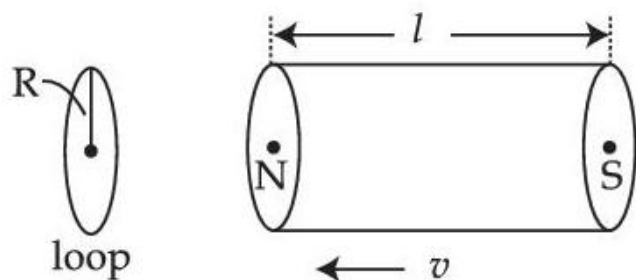
40. An ideal gas is expanding such that $PT^3 = \text{constant}$. The coefficient of volume expansion of the gas is :

- (A) $\frac{1}{T}$
 (B) $\frac{2}{T}$
 (C) $\frac{4}{T}$
 (D) $\frac{3}{T}$

41. In a photoelectric experiment, increasing the intensity of incident light :

- (A) increases the number of photons incident and also increases the K.E. of the ejected electrons
 (B) increases the frequency of photons incident and increases the K.E. of the ejected electrons
 (C) increases the frequency of photons incident and the K.E. of the ejected electrons remains unchanged
 (D) increases the number of photons incident and the K.E. of the ejected electrons remains unchanged

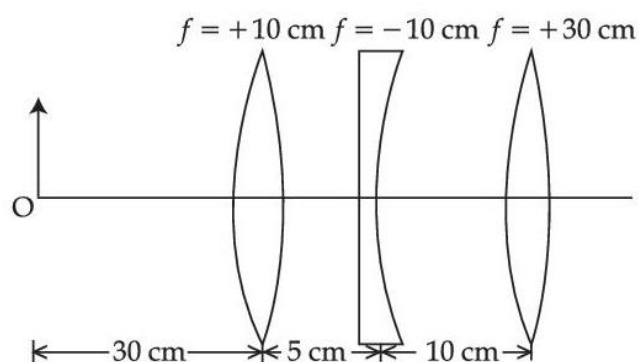
42. A bar magnet is passing through a conducting loop of radius R with velocity v . The radius of the bar magnet is such that it just passes through the loop. The induced e.m.f. in the loop can be represented by the approximate curve :



43. Two ions of masses 4 amu and 16 amu have charges $+2e$ and $+3e$ respectively. These ions pass through the region of constant perpendicular magnetic field. The kinetic energy of both ions is same. Then :

- (A) lighter ion will be deflected less than heavier ion (B) lighter ion will be deflected more than heavier ion
(C) both ions will be deflected equally (D) no ion will be deflected

44. Find the distance of the image from object O, formed by the combination of lenses in the figure :



- (A) 75 cm (B) 10 cm
(C) 20 cm (D) infinity

45. In Millikan's oil drop experiment, what is viscous force acting on an uncharged drop of radius 2.0×10^{-5} m and density $1.2 \times 10^3 \text{ kg m}^{-3}$? Take viscosity of liquid = $1.8 \times 10^{-5} \text{ N s m}^{-2}$. (Neglect buoyancy due to air).

- (A) $3.8 \times 10^{-11} \text{ N}$ (B) $3.9 \times 10^{-10} \text{ N}$
(C) $1.8 \times 10^{-10} \text{ N}$ (D) $5.8 \times 10^{-10} \text{ N}$

46. Electric field in a plane electromagnetic wave is given by $E = 50 \sin(500x - 10 \times 10^{10} t) \text{ V/m}$. The velocity of electromagnetic wave in this medium is :

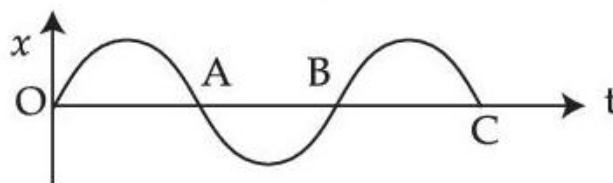
(Given C = speed of light in vacuum)

- (A) $\frac{3}{2}C$
(B) C
(C) $\frac{2}{3}C$
(D) $\frac{C}{2}$

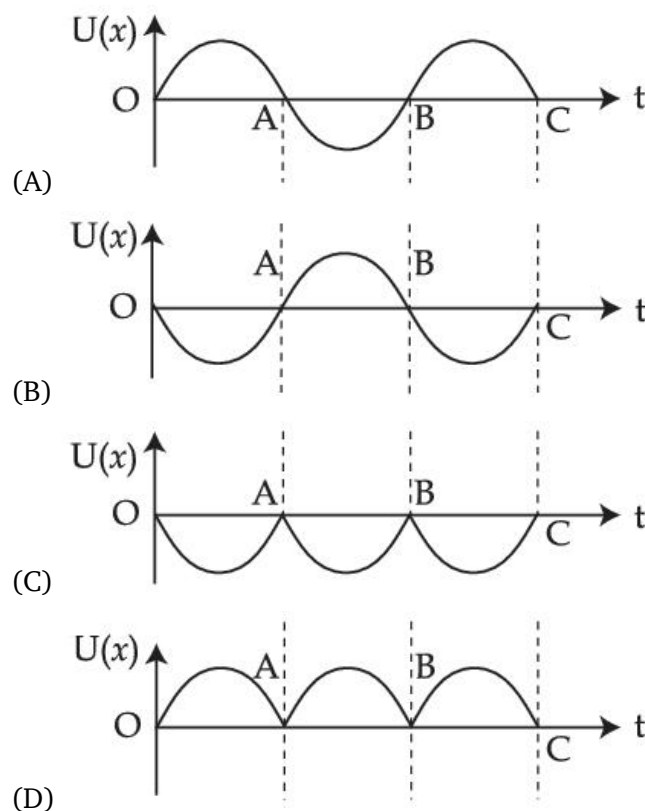
47. Five identical cells each of internal resistance 1Ω and emf 5V are connected in series and in parallel with an external resistance ' R '. For what value of ' R ', current in series and parallel combination will remain the same?

- (A) 1Ω (B) 25Ω
(C) 5Ω (D) 10Ω

48. The variation of displacement with time of a particle executing free simple harmonic motion is shown in the figure.



The potential energy $U(x)$ versus time (t) plot of the particle is correctly shown in figure :



49. There are 10^{10} radioactive nuclei in a given radioactive element, its half-life time is 1 minute. How many nuclei will remain after 30 seconds?

$$(\sqrt{2} = 1.414)$$

- (A) 2×10^{10} (B) 7×10^9
(C) 10^5 (D) 4×10^{10}

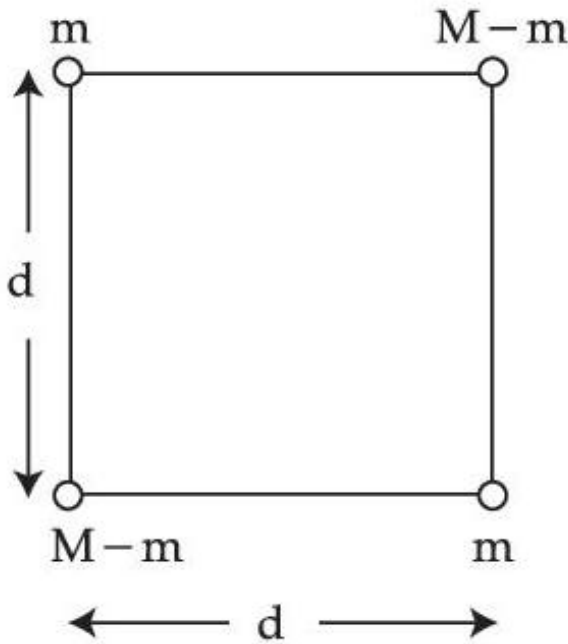
50. For a transistor in CE mode to be used as an amplifier, it must be operated in :

- (A) Both cut-off and Saturation (B) Saturation region only
(C) Cut-off region only (D) The active region only

51. A body of mass $(2M)$ splits into four masses $(m, M - m, m, M - m)$, which are rearranged to form a square as shown in the figure. The ratio of

$$\frac{M}{m}$$

for which, the gravitational potential energy of the system becomes maximum is $x : 1$. The value of x is



52. The alternating current is given by

$$i = \left\{ \sqrt{42} \sin\left(\frac{2\pi}{T}t\right) + 10 \right\} A$$

The r.m.s. value of this current is A.

53. A uniform conducting wire of length is $24a$, and resistance R is wound up as a current carrying coil in the shape of an equilateral triangle of side ' a ' and then in the form of a square of side ' a '. The coil is connected to a voltage source V_0 . The ratio of magnetic moment of the coils in case of equilateral triangle to that for square is $1 :$

$$\sqrt{y}$$

where y is

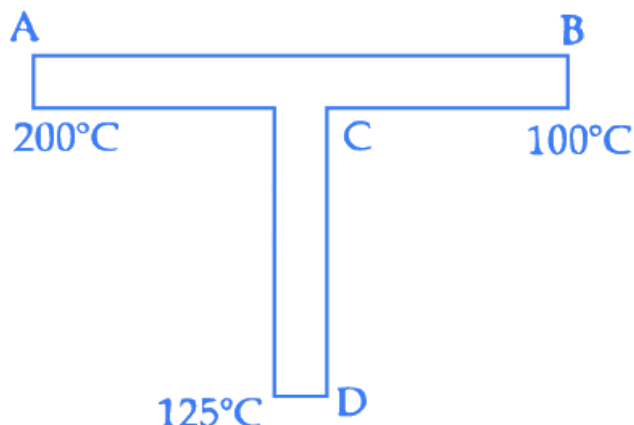
54. First, a set of n equal resistors of 10Ω each are connected in series to a battery of emf $20V$ and internal resistance 10Ω . A current I is observed to flow. Then, the n resistors are connected in parallel to the same battery. It is observed that the current is increased 20 times, then the value of n is

55. If the velocity of a body related to displacement x is given by

$$v = \sqrt{5000 + 24x}$$

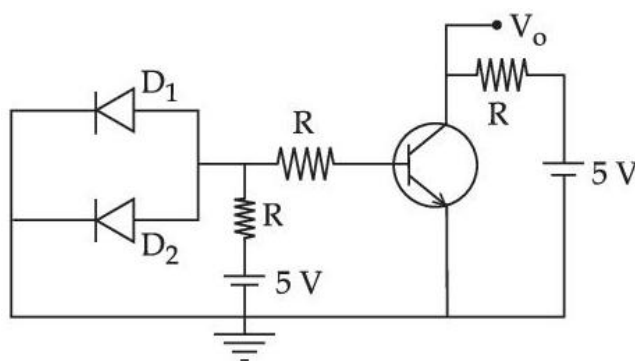
m/s, then the acceleration of the body is m/s^2 .

56. A rod CD of thermal resistance 10.0 KW^{-1} is joined at the middle of an identical rod AB as shown in figure. The end A, B and D are maintained at 200°C , 100°C and 125°C respectively. The heat current in CD is P watt. The value of P is



57. Two persons A and B perform same amount of work in moving a body through a certain distance d with application of forces acting at angle 45° and 60° with the direction of displacement respectively. The ratio of force applied by person A to the force applied by person B is $\frac{1}{\sqrt{x}}$. The value of x is

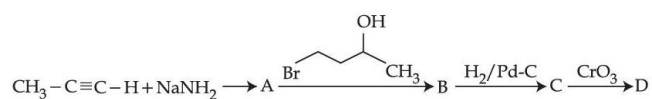
58. A circuit is arranged as shown in figure. The output voltage V_0 is equal to V.



59. Two cars X and Y are approaching each other with velocities 36 km/h and 72 km/h respectively. The frequency of a whistle sound as emitted by a passenger in car X, heard by the passenger in car Y is 1320 Hz. If the velocity of sound in air is 340 m/s, the actual frequency of the whistle sound produced is Hz.

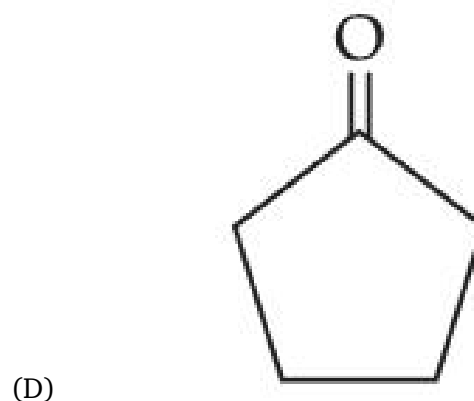
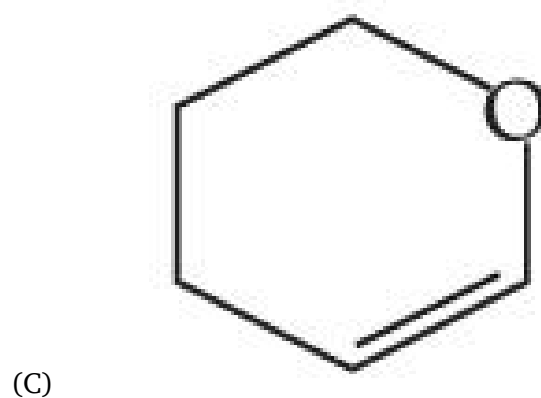
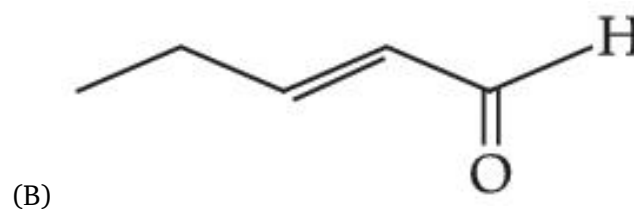
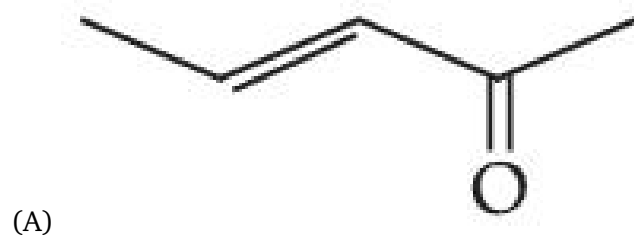
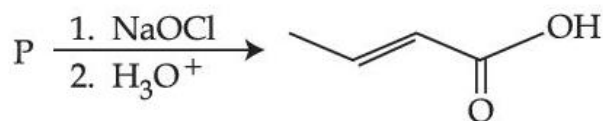
60. A transmitting antenna has a height of 320 m and that of receiving antenna is 2000 m. The maximum distance between them for satisfactory communication in line of sight mode is 'd'. The value of 'd' is km.

61. In the following sequence of reactions, the final product D is :



- (A) $\text{H}_3\text{C} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \overset{\text{O}}{\parallel} \text{C} - \text{H}$
- (B) $\text{CH}_3 - \text{CH} = \text{CH} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{COOH}$
- (C) $\text{H}_3\text{C} - \text{CH} = \text{CH} - \text{CH}(\text{OH}) - \text{CH}_2 - \text{CH}_2 - \text{CH}_3$
- (D) $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \overset{\text{O}}{\parallel} \text{C} - \text{CH}_3$

62. The structure of the starting compound P used in the reaction given below is :



63. Match List - I with List - II :

List - I (Species)	List - II (No. of lone pairs of electrons on the central atom)
(a) XeF_2	(i) 0
(b) XeO_2F_2	(ii) 1
(c) XeO_3F_2	(iii) 2
(d) XeF_4	(iv) 3

Choose the most appropriate answer from the options given below :

- (A) (a)-(iv), (b)-(i), (c)-(ii), (d)-(iii) (B) (a)-(iii), (b)-(iv), (c)-(ii), (d)-(i)
- (C) (a)-(iii), (b)-(ii), (c)-(iv), (d)-(i) (D) (a)-(iv), (b)-(ii), (c)-(i), (d)-(iii)

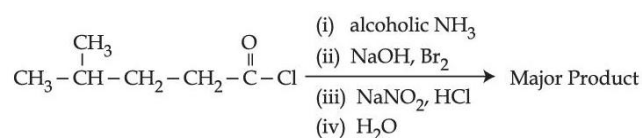
64. Match items of List-I with those of List-II :

List - I (Property)	List - II (Example)
(a) Diamagnetism	(i) MnO
(b) Ferrimagnetism	(ii) O_2
(c) Paramagnetism	(iii) NaCl
(d) Antiferromagnetism	(iv) Fe_3O_4

Choose the most appropriate answer from the options given below :

- (A) (a)-(ii), (b)-(i), (c)-(iii), (d)-(iv) (B) (a)-(i), (b)-(iii), (c)-(iv), (d)-(ii)
- (C) (a)-(iii), (b)-(iv), (c)-(ii), (d)-(i) (D) (a)-(iv), (b)-(ii), (c)-(i), (d)-(iii)

65. The major product of the following reaction is :



- (A) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \overset{\text{Br}}{\text{CH}} - \text{CH}_2\text{OH}$
- (B) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2\text{OH}$
- (C) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_2 - \text{CH}_2\text{OH}$
- (D) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{CH}_2 - \text{CH}_2 - \text{Cl}$

66. Which of the following is not a correct statement for primary aliphatic amines?

- (A) The intermolecular association in primary amines is less than the intermolecular association in secondary amines. (B) Primary amines on treating with nitrous acid solution from corresponding alcohols except methyl amine.
- (C) Primary amines are less basic than the secondary amines. (D) Primary amines can be prepared by the Gabriel phthalimide synthesis.

67. Acidic ferric chloride solution on treatment with excess of potassium ferrocyanide gives a Prussian blue coloured colloidal species. It is :

- (A) $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$ (B) $\text{K}_5\text{Fe}[\text{Fe}(\text{CN})_6]_2$
- (C) $\text{HFe}[\text{Fe}(\text{CN})_6]$ (D) $\text{KFe}[\text{Fe}(\text{CN})_6]$

68. The nature of oxides V_2O_3 and CrO is indexed as 'X' and 'Y' type respectively. The correct set of X and Y is :

- (A)

X = basic		Y = am-pho-teric
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 (B)

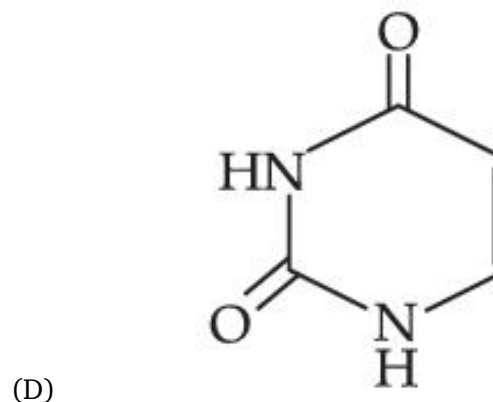
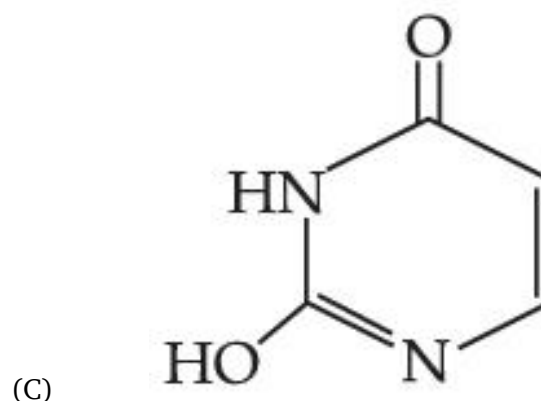
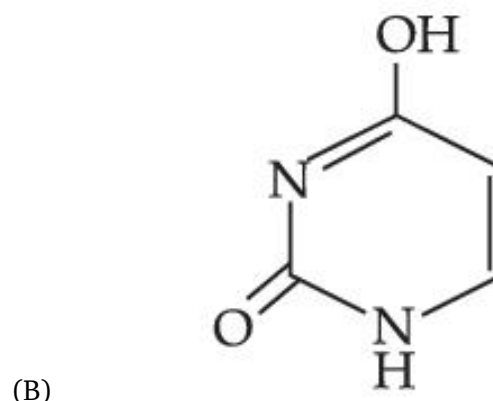
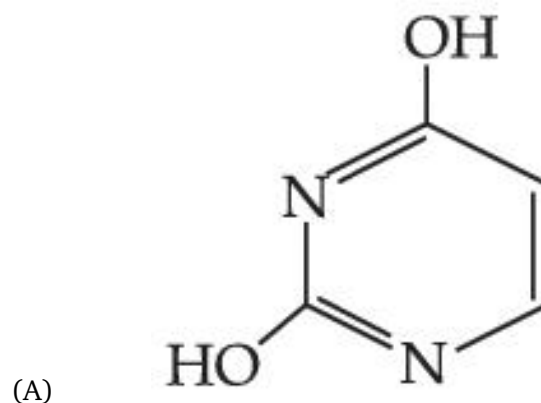
X = am-pho-teric		Y = basic
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- (C)

X = acidi c		Y = acidi c
-------------	--	-------------

 (D)

X = basic		Y = basic
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69. Out of following isomeric forms of uracil, which one is present in RNA?



70. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

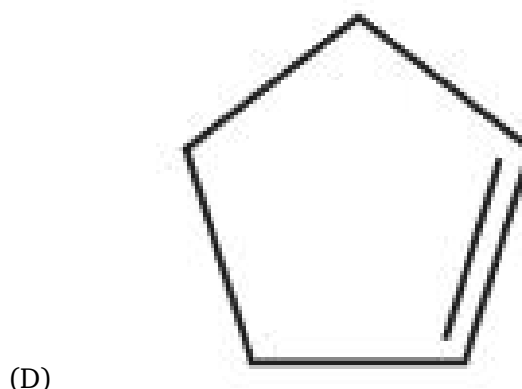
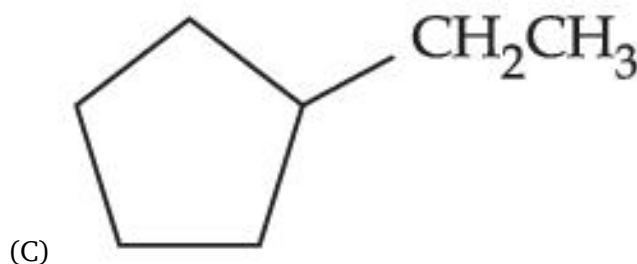
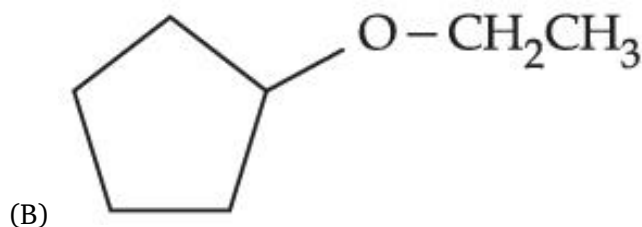
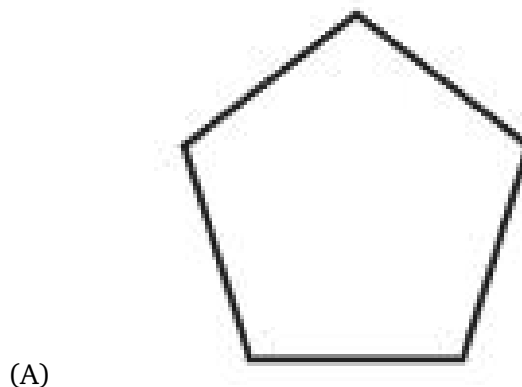
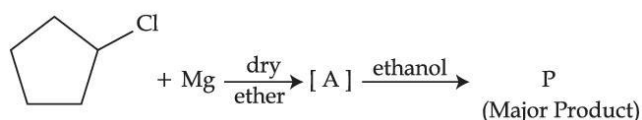
Assertion (A) : Synthesis of ethyl phenyl ether may be achieved by Williamson synthesis.

Reason (R) : Reaction of bromobenzene with sodium ethoxide yields ethyl phenyl ether.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|--|---|
| (A) Both (A) and (R) are correct and (R) is the correct explanation of (A) | (B) (A) is correct but (R) is not correct |
| (C) (A) is not correct but (R) is correct | (D) Both (A) and (R) are correct but (R) is not the correct explanation of (A). |

71. In the following sequence of reactions the P is :



72. In polythionic acid, $\text{H}_2\text{S}_x\text{O}_6$ ($x = 3$ to 5) the oxidation state(s) of sulphur is/are :

- | | |
|----------------------|--------------------|
| (A) + 5 only | (B) + 6 only |
| (C) + 3 and + 5 only | (D) 0 and + 5 only |

73. In which one of the following molecules strongest back donation of an electron pair from halide to boron is expected?

- (A) BCl_3 (B) BF_3
(C) BBr_3 (D) BI_3

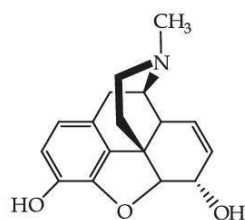
74. Deuterium resembles hydrogen in properties but :

- (A) reacts slower than hydrogen (B) reacts vigorously than hydrogen
(C) reacts just as hydrogen (D) emits β^+ particles

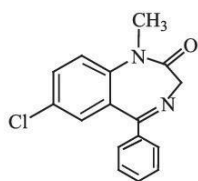
75. Which refining process is generally used in the purification of low melting metals?

- (A) Chromatographic method (B) Liquation
(C) Electrolysis (D) Zone refining

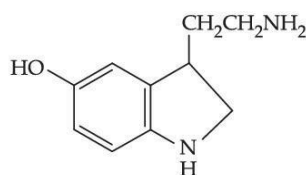
76.



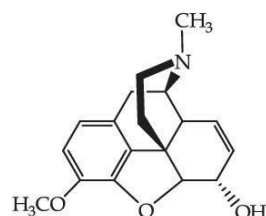
(A)



(B)



(C)



(D)

The correct statement about (A), (B), (C) and (D) is :

- (A) (A), (B) and (C) are narcotic analgesics (B) (B), (C) and (D) are tranquillizers
(C) (A) and (D) are tranquillizers (D) (B) and (C) are narcotic analgesics

77. The gas 'A' is having very low reactivity reaches to stratosphere. It is non-toxic and non-flammable but dissociated by UV-radiations in stratosphere. The intermediates formed initially from the gas 'A' are :

- (A) $\text{Cl}\dot{\text{O}} + \dot{\text{C}}\text{F}_2\text{Cl}$
(B) $\text{Cl}\dot{\text{O}} + \dot{\text{C}}\text{H}_3$
(C) $\dot{\text{C}}\text{H}_3 + \dot{\text{C}}\text{F}_2\text{Cl}$
(D) $\dot{\text{C}}\text{l} + \dot{\text{C}}\text{F}_2\text{Cl}$

78. The number of water molecules in gypsum, dead burnt plaster and plaster of Paris, respectively are :

- (A) 2, 0 and 1 (B) 0.5, 0 and 2
(C) 5, 0 and 0.5 (D) 2, 0 and 0.5

79. The unit of the van der Waals gas equation parameter 'a' in

$$\left(P + \frac{an^2}{V^2}\right)(V - nb) = nRT$$

is :

- (A) kg m s^{-2} (B) $\text{dm}^3 \text{mol}^{-1}$
(C) kg m s^{-1} (D) $\text{atm dm}^6 \text{mol}^{-2}$

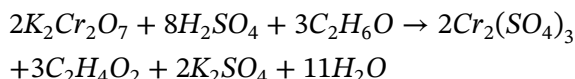
80. Tyndall effect is more effectively shown by :

- (A) true solution (B) lyophilic colloid
(C) lyophobic colloid (D) suspension

Chemistry - Section B · Numerical Value Questions

81. In Carius method for estimation of halogens, 0.2 g of an organic compound gave 0.188 g of AgBr. The percentage of bromine in the compound is _____. (Nearest integer)
[Atomic mass : Ag = 108, Br = 80]

82. The reaction that occurs in a breath analyser, a device used to determine the alcohol level in a person's blood stream is



If the rate of appearance of $Cr_2(SO_4)_3$ is $2.67 \text{ mol min}^{-1}$ at a particular time, the rate of disappearance of C_2H_6O at the same time is _____ mol min^{-1} . (Nearest integer)

83. The kinetic energy of an electron in the second Bohr orbit of a hydrogen atom is equal to $\frac{h^2}{8\pi^2 m a_0^2}$. The value of $10x$ is _____. (a_0 is radius of Bohr's orbit) (Nearest integer) [Given : $\pi = 3.14$]

84. 1 kg of 0.75 molal aqueous solution of sucrose can be cooled up to -4°C before freezing. The amount of ice (in g) that will be separated out is _____. (Nearest integer)
[Given : $K_f(H_2O) = 1.86 \text{ K kg mol}^{-1}$]

85. 1 mol of an octahedral metal complex with formula $MCl_3 \cdot 2L$ on reaction with excess of $AgNO_3$ gives 1 mol of AgCl. The denticity of Ligand L is _____. (Integer answer)

86. The number of moles of CuO, that will be utilized in Dumas method for estimation nitrogen in a sample of 57.5 g of N, N-dimethylaminopentane is _____ $\times 10^{-2}$. (Nearest integer)

87. The number of f electrons in the ground state electronic configuration of Np ($Z = 93$) is _____. (Nearest integer)

88. 200 mL of 0.2 M HCl is mixed with 300 mL of 0.1 M NaOH. The molar heat of neutralization of this reaction is -57.1 kJ . The increase in temperature in $^\circ\text{C}$ of the system on mixing is $x \times 10^{-2}$. The value of x is _____. (Nearest integer)

[Given : Specific heat of water = $4.18 \text{ J g}^{-1} \text{ K}^{-1}$, Density of water = 1.00 g cm^{-3}]
[Assume no volume change on mixing]

89. The number of moles of NH_3 , that must be added to 2L of 0.80 M $AgNO_3$ in order to reduce the concentration of Ag^+ ions to $5.0 \times 10^{-8} \text{ M}$ ($K_{\text{formation}}$ for $[Ag(NH_3)_2]^+ = 1.0 \times 10^8$) is _____. (Nearest integer)
[Assume no volume change on adding NH_3]

90. When 10 mL of an aqueous solution of $KMnO_4$ was titrated in acidic medium, equal volume of 0.1 M of an aqueous solution of ferrous sulphate was required for complete discharge of colour. The strength of $KMnO_4$ in grams per litre is _____ $\times 10^{-2}$. (Nearest integer)
[Atomic mass of K = 39, Mn = 55, O = 16]

- End of Question Paper -