

JEE Main 2023 Paper 1 (B.E./B.Tech) 31 January 2023 Shift 1 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2023 - Session 1**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 31 January 2023, Shift 1. The total duration is **3 hours**.
- (ii) The paper has **90 questions**, of which **75 are to be attempted**, and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 30 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **10 numerical value questions**; attempt **any 5** of them.
- (iv) **All Section A questions are compulsory**. Attempt only the specified number of Section B questions. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme** (Sections A and B): **+4** for a correct answer, **-1** for an incorrect answer, and **0** for an unanswered or marked-for-review question.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

1. The number of real roots of the equation $\sqrt{x^2 - 4x + 3} + \sqrt{x^2 - 9} = \sqrt{4x^2 - 14x + 6}$, is :

- (A) 0 (B) 1
(C) 3 (D) 2

2. A bag contains 6 balls. Two balls are drawn from it at random and both are found to be black. The probability that the bag contains at least 5 black balls is :

- (A) $\frac{3}{7}$
(B) $\frac{5}{6}$
(C) $\frac{5}{7}$
(D) $\frac{2}{7}$

3. If $\sin^{-1} \frac{\alpha}{17} + \cos^{-1} \frac{4}{5} - \tan^{-1} \frac{77}{36} = 0, 0 < \alpha < 13$, then $\sin^{-1}(\sin \alpha) + \cos^{-1}(\cos \alpha)$ is equal to :

- (A) 16
(B) π
(C) $16 - 5\pi$
(D) 0

4. If the sum and product of four positive consecutive terms of a G.P., are 126 and 1296, respectively, then the sum of common ratios of all such GPs is

- (A) 7
(B) 14
(C) 3
(D) $\frac{9}{2}$

5. For the system of linear equations

$$\begin{aligned} x + y + z &= 6 \\ \alpha x + \beta y + 7z &= 3 \\ x + 2y + 3z &= 14 \end{aligned}$$

which of the following is **NOT** true ?

- (A) If $\alpha = \beta = 7$, then the system has no solution
(B) For every point $(\alpha, \beta) \neq (7, 7)$ on the line $x - 2y + 7 = 0$, the system has infinitely many solutions
(C) There is a unique point (α, β) on the line $x + 2y + 18 = 0$ for which the system has infinitely many solutions
(D) If $\alpha = \beta$ and $\alpha \neq 7$, then the system has a unique solution

6. Let R be a relation on $\mathbb{N} \times \mathbb{N}$ defined by $(a, b) R (c, d)$ if and only if $ad(b - c) = bc(a - d)$.

. Then R is

- (A) symmetric and transitive but not reflexive
(B) reflexive and symmetric but not transitive
(C) transitive but neither reflexive nor symmetric
(D) symmetric but neither reflexive nor transitive

7. If the domain of the function

$$f(x) = \frac{[x]}{1 + x^2}$$

, where $[x]$ is greatest integer $\leq x$, is $[2, 6)$, then its range is

- (A) $\left(\frac{5}{37}, \frac{2}{5}\right] - \left\{\frac{9}{29}, \frac{27}{109}, \frac{18}{89}, \frac{9}{53}\right\}$
(B) $\left(\frac{5}{37}, \frac{2}{5}\right]$
(C) $\left[\frac{5}{26}, \frac{2}{5}\right]$
(D) $\left(\frac{5}{26}, \frac{2}{5}\right] - \left\{\frac{9}{29}, \frac{27}{109}, \frac{18}{89}, \frac{9}{53}\right\}$

8. A wire of length 20 m is to be cut into two pieces. A piece of length l_1 is bent to make a square of area A_1 and the other piece of length l_2 is made into a circle of area A_2 . If $2A_1 + 3A_2$ is minimum then $(\pi l_1) : l_2$ is equal to :

- (A) 6 : 1 (B) 1 : 6
(C) 4 : 1 (D) 3 : 1

9. Let $\alpha \in (0, 1)$ and $\beta = \log_e(1 - \alpha)$

. Let

$$P_n(x) = x + \frac{x^2}{2} + \frac{x^3}{3} + \dots + \frac{x^n}{n}, x \in (0, 1)$$

. Then the integral

$$\int_0^\alpha \frac{t^{50}}{1-t} dt$$

is equal to

- (A) $-(\beta + P_{50}(\alpha))$
(B) $\beta - P_{50}(\alpha)$
(C) $P_{50}(\alpha) - \beta$
(D) $\beta + P_{50} - (\alpha)$

10. Let the shortest distance between the lines

$$L : \frac{x-5}{-2} = \frac{y-\lambda}{0} = \frac{z+\lambda}{1}, \lambda \geq 0$$

and

$$L_1 : x+1 = y-1 = 4-z$$

be $2\sqrt{6}$. If (α, β, γ) lies on L ,

then which of the following is NOT possible?

- (A) $\alpha + 2\gamma = 24$
(B) $2\alpha + \gamma = 7$
(C) $\alpha - 2\gamma = 19$
(D) $2\alpha - \gamma = 9$

11. Let

$$\vec{a} = 2\hat{i} + \hat{j} + \hat{k}$$

, and $\vec{b}$ and $\vec{c}$ be two nonzero vectors such that

$$|\vec{a} + \vec{b} + \vec{c}| = |\vec{a} + \vec{b} - \vec{c}|$$

and

$$\vec{b} \cdot \vec{c} = 0$$

. Consider the following two statements:

- (A) $|\vec{a} + \lambda\vec{c}| \geq |\vec{a}|$ for all $\lambda \in \mathbb{R}$.
(B) $\vec{a}$ and $\vec{c}$ are always parallel.

Then,

- (A) only (B) is correct (B) both (A) and (B) are correct
(C) only (A) is correct (D) neither (A) nor (B) is correct

12. Let

$$A = \begin{pmatrix} 1 & 0 & 0 \\ 0 & 4 & -1 \\ 0 & 12 & -3 \end{pmatrix}$$

. Then the sum of the diagonal elements of the matrix $(A + I)^{11}$ is equal to :

- (A) 4094 (B) 2050
(C) 6144 (D) 4097

13. For all $z \in C$ on the curve

$$C_1 : |z| = 4$$

, let the locus of the point

$$z + \frac{1}{z}$$

be the curve C_2 . Then :

- (A) the curves C_1 and C_2 intersect at 4 points (B) the curve C_2 lies inside C_1
(C) the curve C_1 lies inside C_2 (D) the curves C_1 and C_2 intersect at 2 points

14. The value of

$$\int_{\frac{\pi}{3}}^{\frac{\pi}{2}} \frac{(2 + 3 \sin x)}{\sin x(1 + \cos x)} dx$$

is equal to :

- (A) $\frac{10}{3} - \sqrt{3} + \log_e \sqrt{3}$
(B) $\frac{7}{2} - \sqrt{3} - \log_e \sqrt{3}$
(C) $\frac{10}{3} - \sqrt{3} - \log_e \sqrt{3}$
(D) $-2 + 3\sqrt{3} + \log_e \sqrt{3}$

15. Let

$$y = f(x) = \sin^3 \left(\frac{\pi}{3} \left(\cos \left(\frac{\pi}{3\sqrt{2}} \right. \right. \right. \\ \left. \left. \left. (-4x^3 + 5x^2 + 1)^{\frac{3}{2}} \right) \right) \right)$$

. Then, at $x = 1$,

- (A) $2y' + \sqrt{3} \pi^2 y = 0$
 (B) $y' + 3\pi^2 y = 0$
 (C) $\sqrt{2} y' - 3\pi^2 y = 0$
 (D) $2y' + 3\pi^2 y = 0$

16. Let a differentiable function f satisfy

$$f(x) + \int_3^x \frac{f(t)}{t} dt = \sqrt{x+1}, x \geq 3$$

. Then $12f(8)$ is equal to :

- (A) 19 (B) 34
 (C) 17 (D) 1

17. Let a circle C_1 be obtained on rolling the circle

$$x^2 + y^2 - 4x - 6y + 11 = 0$$

upwards 4 units on the tangent T to it at the point $(3, 2)$. Let C_2 be the image of C_1 in T . Let A and B be the centers of circles C_1 and C_2 respectively, and M and N be respectively the feet of perpendiculars drawn from A and B on the x -axis. Then the area of the trapezium $AMNB$ is :

- (A) $2(2 + \sqrt{2})$
 (B) $4(1 + \sqrt{2})$
 (C) $3 + 2\sqrt{2}$
 (D) $2(1 + \sqrt{2})$

18. If the maximum distance of normal to the ellipse

$$\frac{x^2}{4} + \frac{y^2}{b^2} = 1, b < 2$$

, from the origin is 1, then the eccentricity of the ellipse is :

- (A) $\frac{\sqrt{3}}{4}$
 (B) $\frac{1}{2}$
 (C) $\frac{1}{\sqrt{2}}$
 (D) $\frac{\sqrt{3}}{2}$

19. (S1) $(p \Rightarrow q) \vee (p \wedge (\sim q))$ is a tautology

(S2) $((\sim p) \Rightarrow (\sim q)) \wedge ((\sim p) \vee q)$ is a contradiction.

Then

- (A) only (S2) is correct (B) both (S1) and (S2) are correct
 (C) only (S1) is correct (D) both (S1) and (S2) are wrong

20. Let

$y = f(x)$ represent a parabola with focus

$$\left(-\frac{1}{2}, 0\right)$$

and directrix

$$y = -\frac{1}{2}$$

. Then

$$S = \left\{ x \in \mathbb{R} : \tan^{-1}(\sqrt{f(x)}) + \sin^{-1}(\sqrt{f(x)+1}) \right. \\ \left. = \frac{\pi}{2} \right\}$$

:

- (A) is an empty set (B) contains exactly one element
 (C) contains exactly two elements (D) is an infinite set

21. The remainder on dividing 5^{99} by 11 is _____.

22. Let 5 digit numbers be constructed using the digits 0, 2, 3, 4, 7, 9 with repetition allowed, and are arranged in ascending order with serial numbers. Then the serial number of the number 42923 is _____.

23. Let $\alpha > 0$, be the smallest number such that the expansion of

$$\left(x^{\frac{2}{3}} + \frac{2}{x^3}\right)^{30}$$

has a term $\beta x^{-\alpha}$, $\beta \in \mathbb{N}$. Then α is equal to _____.

24. Number of 4-digit numbers that are less than or equal to 2800 and either divisible by 3 or by 11, is equal to _____.

25. Let $a_1, a_2, \dots, a_n$ be in A.P. If $a_5 = 2a_7$ and $a_{11} = 18$, then

$$12 \left(\frac{1}{\sqrt{a_{10}} + \sqrt{a_{11}}} + \frac{1}{\sqrt{a_{11}} + \sqrt{a_{12}}} + \dots + \frac{1}{\sqrt{a_{17}} + \sqrt{a_{18}}} \right)$$

is equal to _____.

26. Let $\vec{a}$ and $\vec{b}$ be two vectors such that

$$|\vec{a}| = \sqrt{14}, |\vec{b}| = \sqrt{6}$$

and

$$|\vec{a} \times \vec{b}| = \sqrt{48}$$

. Then $(\vec{a} \cdot \vec{b})^2$ is equal to _____.

27. Let for $x \in \mathbb{R}$,

$$f(x) = \frac{x + |x|}{2} \text{ and } g(x) = \begin{cases} x, & x < 0 \\ x^2, & x \geq 0 \end{cases}$$

Then area bounded by the curve

$$y = (f \circ g)(x)$$

and the lines

$$y = 0, 2y - x = 15$$

is equal to _____.

28. If the variance of the frequency distribution

x_i	2	3	4	5	6	7	8
Frequency f_i	3	6	16	α	9	5	6

is 3, then α is equal to _____.

29. Let the line

$$L: \frac{x-1}{2} = \frac{y+1}{-1} = \frac{z-3}{1}$$

intersect the plane $2x + y + 3z = 16$ at the point

P . Let the point Q be the foot of perpendicular

from the point $R(1, -1, -3)$ on the line L . If α is

the area of triangle PQR , then α^2 is equal to

_____.

30. Let θ be the angle between the planes

$$P_1: \vec{r} \cdot (\hat{i} + \hat{j} + 2\hat{k}) = 9$$

and

$$P_2: \vec{r} \cdot (2\hat{i} - \hat{j} + \hat{k}) = 15$$

. Let L be the line that meets P_2 at the point

$(4, -2, 5)$ and makes an angle θ with the normal

of P_2 . If α is the angle between L and P_2 , then

$(\tan^2 \theta) (\cot^2 \alpha)$ is equal to _____.

Physics - Section A · Multiple Choice Questions

31. The initial speed of a projectile fired from ground is u . At the highest point during its motion, the speed of projectile is

$$\frac{\sqrt{3}}{2}u$$

. The time of flight of the projectile is :

- (A) $\frac{u}{g}$
 (B) $\frac{2u}{g}$
 (C) $\frac{u}{2g}$
 (D) $\frac{\sqrt{3} u}{g}$

32. A bar magnet with a magnetic moment 5.0 Am^2 is placed in parallel position relative to a magnetic field of 0.4 T . The amount of required work done in turning the magnet from parallel to antiparallel position relative to the field direction is _____.

- (A) zero
 (B) 1 J
 (C) 2 J
 (D) 4 J

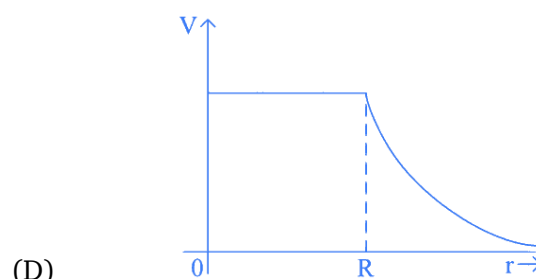
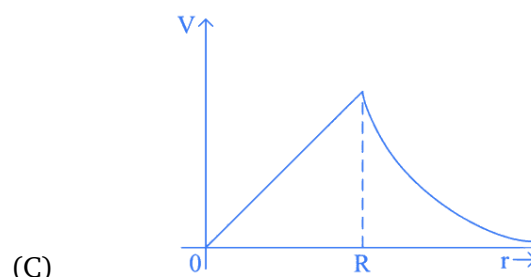
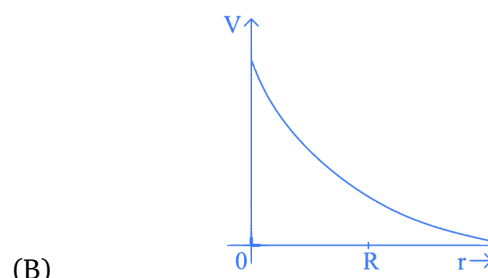
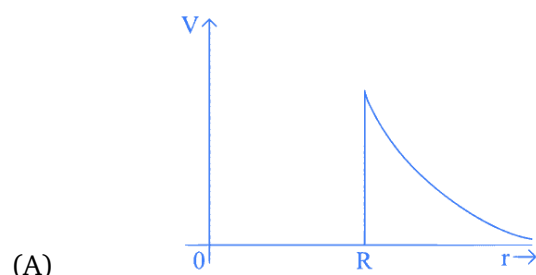
33. Spherical insulating ball and a spherical metallic ball of same size and mass are dropped from the same height. Choose the correct statement out of the following {Assume negligible air friction}

- (A) Metal ball will reach the earth's surface earlier than the insulating ball
 (B) Both will reach the earth's surface simultaneously.
 (C) Insulating ball will reach the earth's surface earlier than the metal ball
 (D) Time taken by them to reach the earth's surface will be independent of the properties of their materials

34. The effect of increase in temperature on the number of electrons in conduction band (n_e) and resistance of a semiconductor will be as:

- (A) n_e decreases, resistance increases
 (B) Both n_e and resistance increase
 (C) n_e increases, resistance decreases
 (D) Both n_e and resistance decrease

35. Which of the following correctly represents the variation of electric potential (V) of a charged spherical conductor of radius (R) with radial distance (r) from the center?



36. The drift velocity of electrons for a conductor connected in an electrical circuit is V_d . The conductor is now replaced by another conductor with same material and same length but double the area of cross section. The applied voltage remains same. The new drift velocity of electrons will be

- (A) $\frac{V_d}{4}$
 (B) V_d
 (C) $2 V_d$
 (D) $\frac{V_d}{2}$

37. A rod with circular cross-section area 2 cm^2 and length 40 cm is wound uniformly with 400 turns of an insulated wire. If a current of 0.4 A flows in the wire windings, the total magnetic flux produced inside windings is $4\pi \times 10^{-6} \text{ Wb}$. The relative permeability of the rod is

(Given : Permeability of vacuum $\mu_0 = 4\pi \times 10^{-7} \text{ NA}^{-2}$)

- (A) $\frac{5}{16}$
 (B) 125
 (C) $\frac{32}{5}$
 (D) 12.5

38. At a certain depth "d" below surface of earth, value of acceleration due to gravity becomes four times that of its value at a height $3R$ above earth surface. Where R is Radius of earth (Take

- $R = 6400 \text{ km}$)
). The depth d is equal to
 (A) 5260 km (B) 2560 km
 (C) 640 km (D) 4800 km

39. Two polaroids A and B are placed in such a way that the pass-axis of polaroids are perpendicular to each other. Now, another polaroid C is placed between A and B bisecting angle between them. If intensity of unpolarized light is I_0 then intensity of transmitted light after passing through polaroid B will be:

- (A) $\frac{I_0}{4}$
 (B) $\frac{I_0}{8}$
 (C) Zero
 (D) $\frac{I_0}{2}$

40. 100 balls each of mass m moving with speed v simultaneously strike a wall normally and reflected back with same speed, in time t s. The total force exerted by the balls on the wall is

- (A) $\frac{200mv}{t}$
 (B) $\frac{100mv}{t}$
 (C) $\frac{mv}{100t}$
 (D) $200mvt$

41. If R , X_L , and X_C represent resistance, inductive reactance and capacitive reactance. Then which of the following is dimensionless :

- (A) $\frac{R}{X_L X_C}$
 (B) $R X_L X_C$
 (C) $\frac{R}{\sqrt{X_L X_C}}$
 (D) $R \frac{X_L}{X_C}$

42. The maximum potential energy of a block executing simple harmonic motion is 25 J. A is amplitude of oscillation. At A/2, the kinetic energy of the block is

- (A) 9.75 J (B) 37.5 J
(C) 18.75 J (D) 12.5 J

43. If 1000 droplets of water of surface tension 0.07 N/m, having same radius 1 mm each, combine to form a single drop. In the process the released surface energy is :-

$$\left(\text{Take } \pi = \frac{22}{7} \right)$$

- (A) 7.92×10^{-4} J
(B) 7.92×10^{-6} J
(C) 8.8×10^{-5} J
(D) 9.68×10^{-4} J

44. As shown in figure, a 70 kg garden roller is pushed with a force of

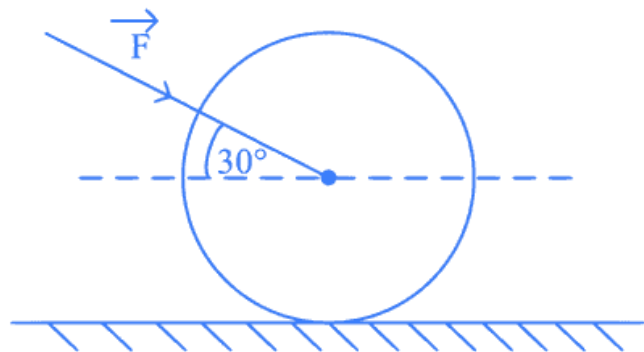
$$\vec{F} = 200 \text{ N}$$

at an angle of 30° with horizontal. The normal reaction on the roller is

(Given

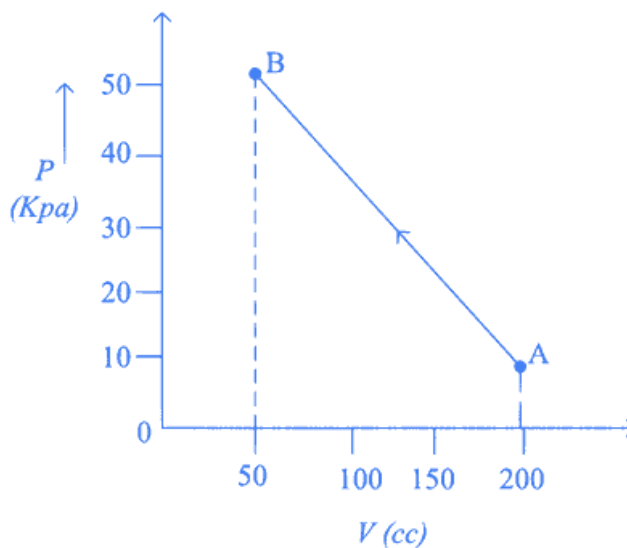
$$g = 10 \text{ m s}^{-2}$$

)



- (A) 800 N (B) 600 N
(C) $200\sqrt{3}$ N (D) $800\sqrt{2}$ N

45. The pressure of a gas changes linearly with volume from A to B as shown in figure. If no heat is supplied to or extracted from the gas then change in the internal energy of the gas will be



- (A) 6 J (B) 4.5 J
(C) zero (D) -4.5 J

46. The correct relation between

$$\gamma = \frac{c_p}{c_v}$$

and temperature T is :

- (A) $\gamma \propto T$
(B) $\gamma \propto \frac{1}{\sqrt{T}}$
(C) $\gamma \propto \frac{1}{T}$
(D) $\gamma \propto T^\circ$

47. If a source of electromagnetic radiation having power 15 kW produces 10^{16} photons per second, the radiation belongs to a part of spectrum is.

(Take Planck constant

$$h = 6 \times 10^{-34} \text{ Js}$$

)

- | | |
|-----------------|----------------------|
| (A) Gamma rays | (B) Radio waves |
| (C) Micro waves | (D) Ultraviolet rays |

48. Given below are two statements : One is labelled as Assertion A and the other is labelled as Reason R

Assertion A : The beam of electrons show wave nature and exhibit interference and diffraction.

Reason R : Davisson Germer Experimentally verified the wave nature of electrons.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|--|--|
| (A) A is not correct but R is correct. | (B) Both A and R are correct and R is the correct explanation of A |
| (C) Both A and R are correct but R is Not the correct explanation of A | (D) A is correct but R is not correct |

49. The amplitude of $15 \sin(1000\pi t)$ is modulated by $10 \sin(4\pi t)$ signal. The amplitude modulated signal contains frequency (ies) of

- A. 500 Hz
- B. 2 Hz
- C. 250 Hz
- D. 498 Hz
- E. 502 Hz

Choose the correct answer from the options given below:

- | | |
|---------------------|------------------|
| (A) B only | (B) A only |
| (C) A, D and E only | (D) A and B only |

50. A free neutron decays into a proton but a free proton does not decay into neutron. This is because

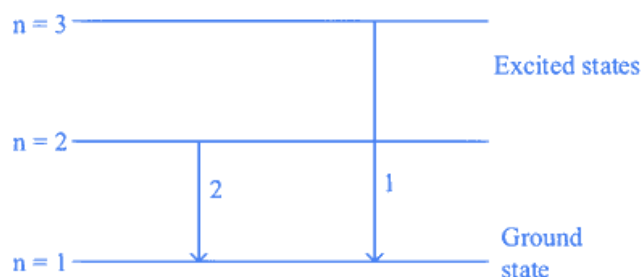
- | | |
|--|--|
| (A) neutron is an uncharged particle | (B) neutron has larger rest mass than proton |
| (C) neutron is a composite particle made of a proton and an electron | (D) proton is a charged particle |

51. A lift of mass

$$M = 500 \text{ kg}$$

is descending with speed of 2 ms^{-1} . Its supporting cable begins to slip thus allowing it to fall with a constant acceleration of 2 ms^{-2} . The kinetic energy of the lift at the end of fall through to a distance of 6 m will be _____ kJ.

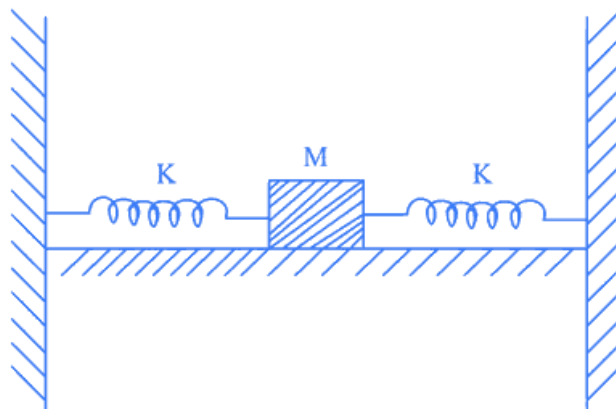
52. For hydrogen atom, λ_1 and λ_2 are the wavelengths corresponding to the transitions 1 and 2 respectively as shown in figure. The ratio of λ_1 and λ_2 is $\frac{x}{32}$. The value of x is _____.



53. In the figure given below, a block of mass $M = 490 \text{ g}$

placed on a frictionless table is connected with two springs having same spring constant ($K = 2 \text{ N m}^{-1}$)

. If the block is horizontally displaced through 'X' m then the number of complete oscillations it will make in 14π seconds will be _____.



54. The speed of a swimmer is 4 km h^{-1} in still water. If the swimmer makes his strokes normal to the flow of river of width 1 km, he reaches a point 750 m down the stream on the opposite bank.

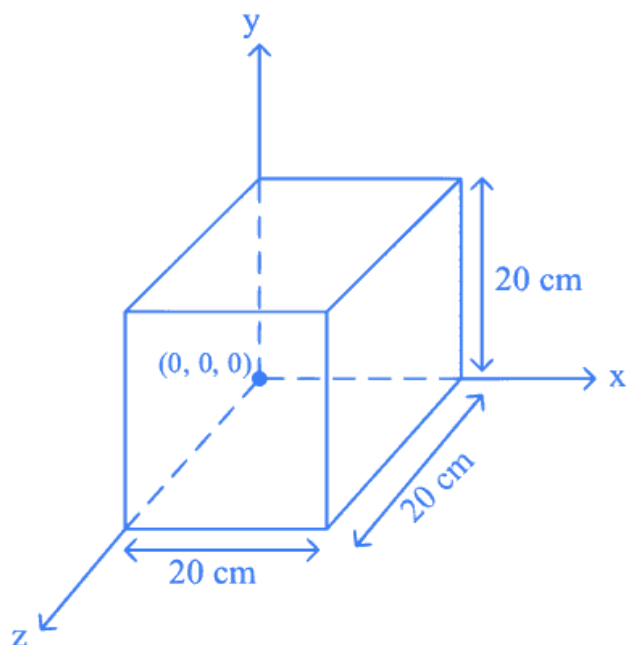
The speed of the river water is _____ km h^{-1}

55. A solid sphere of mass 1 kg rolls without slipping on a plane surface. Its kinetic energy is 7×10^{-3} J. The speed of the centre of mass of the sphere is _____ cm s^{-1}

56. Expression for an electric field is given by

$$\vec{E} = 4000x^2 \hat{i} \frac{\text{V}}{\text{m}}$$

. The electric flux through the cube of side 20 cm when placed in electric field (as shown in the figure) is _____ V cm.



57. A thin rod having a length of 1 m and area of cross-section $3 \times 10^{-6} \text{ m}^2$ is suspended vertically from one end. The rod is cooled from 210°C to 160°C . After cooling, a mass M is attached at the lower end of the rod such that the length of rod again becomes 1 m. Young's modulus and coefficient of linear expansion of the rod are $2 \times 10^{11} \text{ N m}^{-2}$ and $2 \times 10^{-5} \text{ K}^{-1}$, respectively. The value of M is _____ kg.

(Take

$$g = 10 \text{ m s}^{-2}$$

)

58. In a medium the speed of light wave decreases to 0.2 times to its speed in free space. The ratio of relative permittivity to the refractive index of the medium is $x : 1$. The value of x is _____.

(Given speed of light in free space

$$= 3 \times 10^8 \text{ m s}^{-1}$$

and for the given medium

$$\mu_r = 1$$

)

59. Two identical cells, when connected either in parallel or in series gives same current in an external resistance 5Ω . The internal resistance of each cell will be _____ Ω .

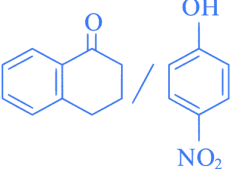
60. An inductor of 0.5 mH , a capacitor of $20 \mu\text{F}$ and resistance of 20Ω are connected in series with a 220 V ac source. If the current is in phase with the emf, the amplitude of current of the circuit is $\sqrt{x} \text{ A}$. The value of x is _____

Chemistry - Section A · Multiple Choice Questions

61. When Cu^{2+} ion is treated with KI, a white precipitate, X appears in solution. The solution is titrated with sodium thiosulphate, the compound Y is formed. X and Y respectively are :

- | | | |
|-----|------------------------------------|--|
| (A) | $\text{X} = \text{CuI}_2$ | $\text{Y} = \text{Na}_2\text{S}_4\text{O}_6$ |
| (B) | $\text{X} = \text{Cu}_2\text{I}_2$ | $\text{Y} = \text{Na}_2\text{S}_4\text{O}_6$ |
| (C) | $\text{X} = \text{CuI}_2$ | $\text{Y} = \text{Na}_2\text{S}_2\text{O}_3$ |
| (D) | $\text{X} = \text{Cu}_2\text{I}_2$ | $\text{Y} = \text{Na}_2\text{S}_4\text{O}_5$ |

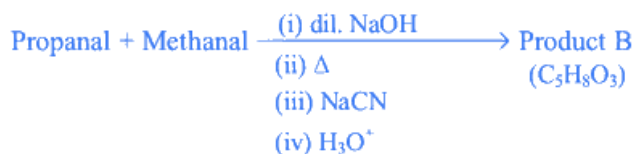
62. Match items of column I and II

	Column I (Mixture of compounds)		Column II (Separation Technique)
A.	$\text{H}_2\text{O}/\text{CH}_2\text{Cl}_2$	i.	Crystallization
B.		ii.	Differential solvent extraction
C.	Kerosene / Naphthalene	iii.	Column chromatography
D.	$\text{C}_6\text{H}_{12}\text{O}_6/\text{NaCl}$	iv.	Fractional Distillation

Correct match is

- | | |
|------------------------------------|------------------------------------|
| (A) A-(ii), B-(iv), C-(i), D-(iii) | (B) A-(iii), B-(iv), C-(ii), D-(i) |
| (C) A-(i), B-(iii), C-(ii), D-(iv) | (D) A-(ii), B-(iii), C-(iv), D-(i) |

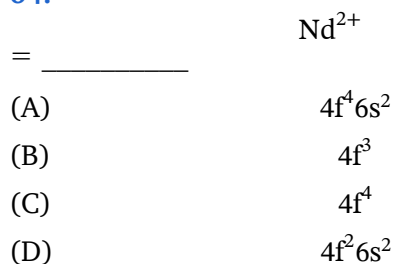
63. Consider the following reaction



The correct statement for product B is. It is

- | | |
|--|---|
| (A) racemic mixture and is neutral | (B) optically active and adds one mole of bromine |
| (C) racemic mixture and gives a gas with saturated NaHCO_3 solution | (D) optically active alcohol and is neutral |

64.



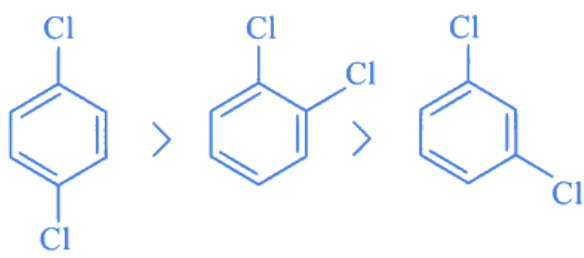
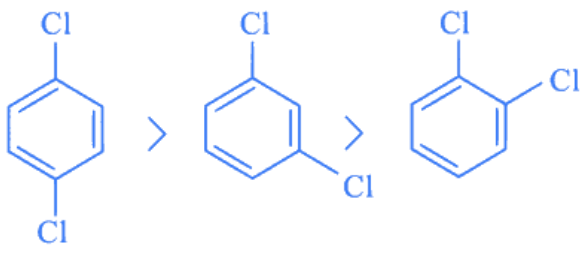
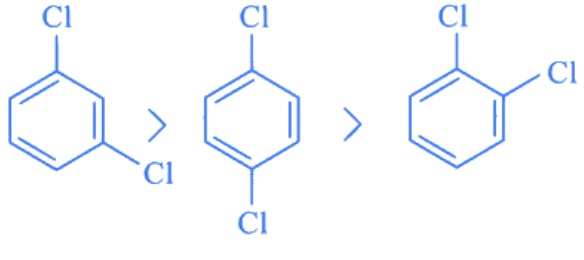
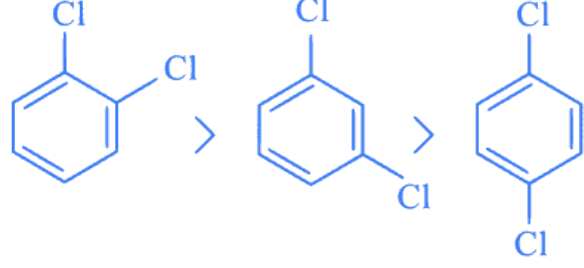
65. The correct order of basicity of oxides of vanadium is :

- | | |
|-----|--|
| (A) | $\text{V}_2\text{O}_3 > \text{V}_2\text{O}_5 > \text{V}_2\text{O}_4$ |
| (B) | $\text{V}_2\text{O}_3 > \text{V}_2\text{O}_4 > \text{V}_2\text{O}_5$ |
| (C) | $\text{V}_2\text{O}_5 > \text{V}_2\text{O}_4 > \text{V}_2\text{O}_3$ |
| (D) | $\text{V}_2\text{O}_4 > \text{V}_2\text{O}_3 > \text{V}_2\text{O}_5$ |

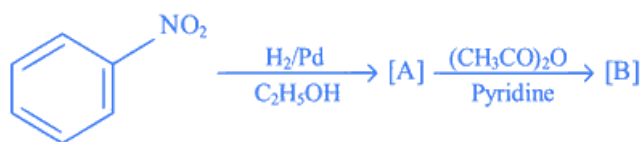
66. The correct increasing order of the ionic radii is

- (A) $\text{Cl}^- < \text{Ca}^{2+} < \text{K}^+ < \text{S}^{2-}$
 (B) $\text{K}^+ < \text{S}^{2-} < \text{Ca}^{2+} < \text{Cl}^-$
 (C) $\text{Ca}^{2+} < \text{K}^+ < \text{Cl}^- < \text{S}^{2-}$
 (D) $\text{S}^{2-} < \text{Cl}^- < \text{Ca}^{2+} < \text{K}^+$

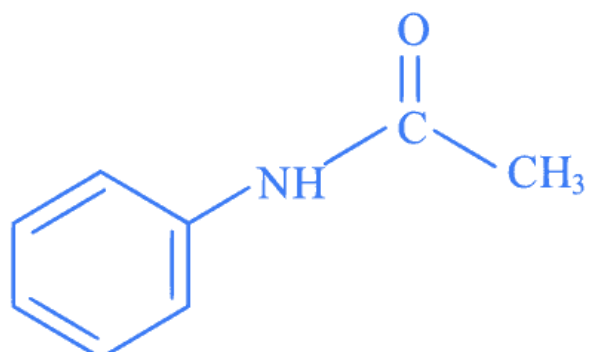
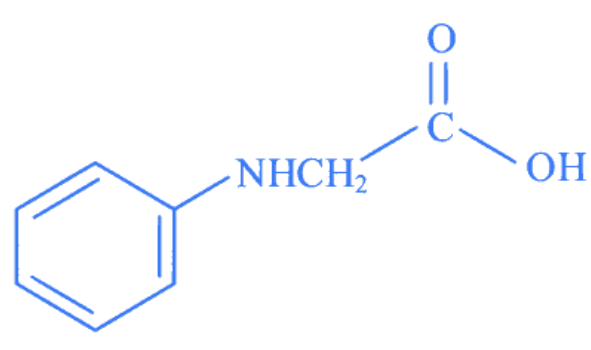
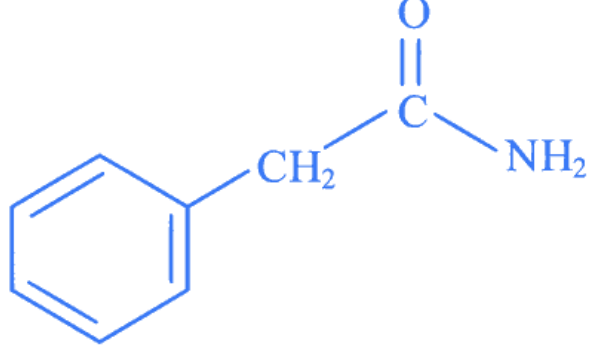
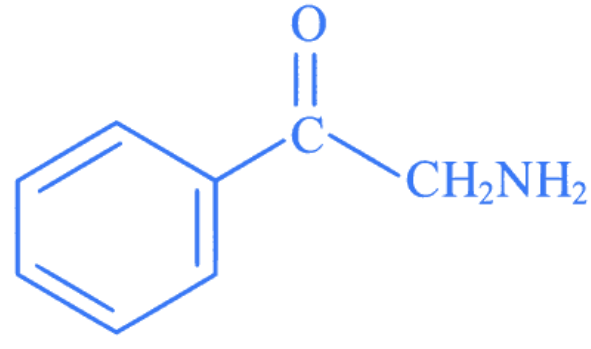
67. The correct order of melting points of dichlorobenzenes is

- (A) 
 (B) 
 (C) 
 (D) 

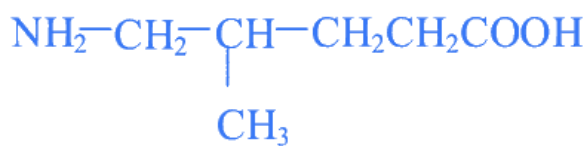
68.



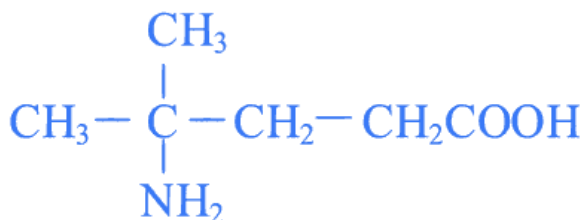
Consider the above reaction and identify the product B.

- (A) 
 (B) 
 (C) 
 (D) 

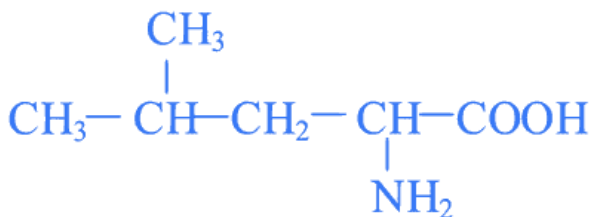
69. A protein 'X' with molecular weight of 70,000 u, on hydrolysis gives amino acids. One of these amino acid is



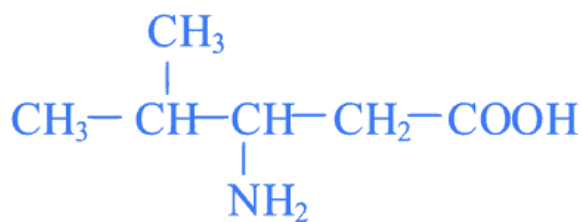
(A)



(B)



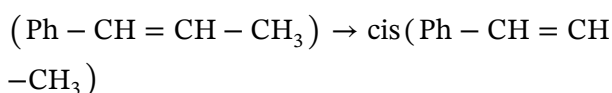
(C)



(D)

70. Choose the correct set of reagents for the following conversion.

trans



(A) Br_2 , alc · KOH, NaNH_2 , H_2 Lindlar Catalyst

(B) Br_2 , aq · KOH, NaNH_2 , Na (Liq NH_3)

(C) Br_2 , alc · KOH, NaNH_2 , Na (Liq NH_3)

(D) Br_2 , aq · KOH, NaNH_2 , H_2 Lindlar Catalyst

71. Which transition in the hydrogen spectrum would have the same wavelength as the Balmer type transition from

$$n = 4$$

to

$$n = 2$$

of He^+ spectrum

(A) to $n = 3$

to

$$n = 4$$

(B) to $n = 2$

to

$$n = 1$$

(C) to $n = 1$

to

$$n = 2$$

(D) to $n = 1$

to

$$n = 3$$

72. Which one of the following statements is correct for electrolysis of brine solution?

(A) O_2 is formed at cathode (B) H_2 is formed at anode

(C) Cl_2 is formed at cathode (D) OH^- is formed at cathode

73. Cobalt chloride when dissolved in water forms pink colored complex X which has octahedral geometry. This solution on treating with conc HCl forms deep blue complex, Y which has a Z geometry. X, Y and Z, respectively, are

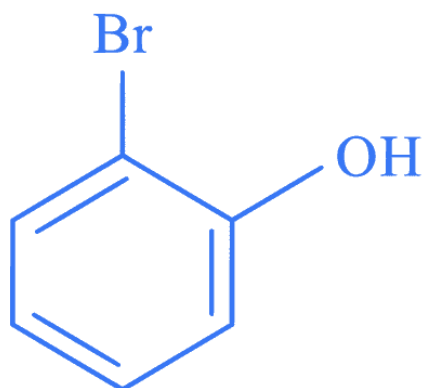
(A) $\text{X} = [\text{Co}(\text{H}_2\text{O})_6]^{2+}$, $\text{Y} = [\text{CoCl}_4]^{2-}$, Z = Tetrahedral

(B) $\text{X} = [\text{Co}(\text{H}_2\text{O})_6]^{2+}$, $\text{Y} = [\text{CoCl}_6]^{3-}$, Z = Octahedral

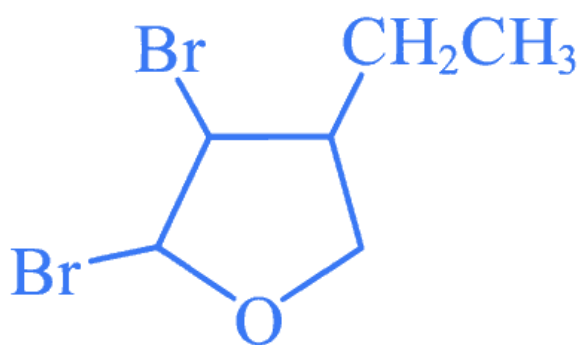
(C) $\text{X} = [\text{Co}(\text{H}_2\text{O})_4\text{Cl}_2]^+$, $\text{Y} = [\text{CoCl}_4]^{2-}$, Z = Tetrahedral

(D) $\text{X} = [\text{Co}(\text{H}_2\text{O})_6]^{3+}$, $\text{Y} = [\text{CoCl}_6]^{3-}$, Z = Octahedral

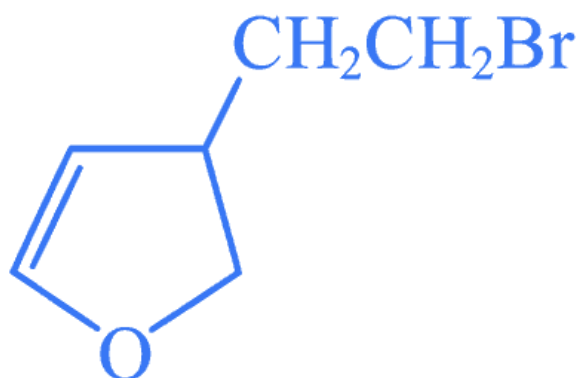
74. An organic compound 'A' with empirical formula C_6H_6O gives sooty flame on burning. Its reaction with bromine solution in low polarity solvent results in high yield of B. B is



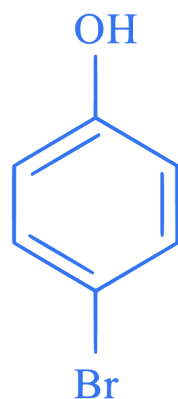
(A)



(B)



(C)



(D)

75. Match List I with List II

	List I		List II
A.	XeF_4	I.	See-saw
B.	SF_4	II.	Square-planar
C.	NH_4^+	III.	Bent T-shaped
D.	BrF_3	IV.	Tetrahedral

Choose the correct answer from the options given below :

- (A) A - II, B - I, C - III, D - IV (B) A - II, B - I, C - IV, D - III
 (C) A - IV, B - I, C - II, D - III (D) A - IV, B - III, C - II, D - I

76. The methods NOT involved in concentration of ore are

- A. Liquefaction
 B. Leaching
 C. Electrolysis
 D. Hydraulic washing
 E. Froth floatation

Choose the correct answer from the options given below :

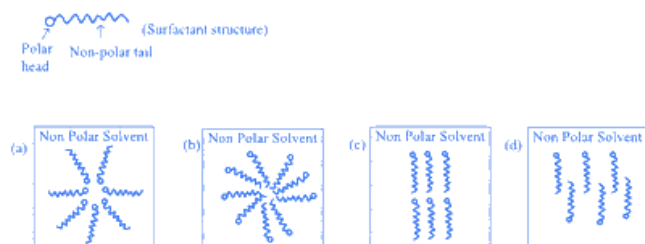
- (A) A and C only (B) C, D and E only
 (C) B, D and C only (D) B, D and E only

77. Identify X, Y and Z in the following reaction.
(Equation not balanced)



- (A) X = ClONO₂, Y = HOCl, Z = NO₂
 (B) X = ClNO₃, Y = Cl₂, Z = NO₂
 (C) X = ClONO₂, Y = HOCl, Z = HNO₃
 (D) X = ClNO₂, Y = HCl, Z = HNO₃

78. Adding surfactants in non polar solvent, the micelles structure will look like



- (A) d (B) a
 (C) c (D) b

79. H₂O₂ acts as a reducing agent in :

- (A) $\text{Mn}^{2+} + 2\text{H}_2\text{O}_2 \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$
 (B) $2\text{Fe}^{2+} + 2\text{H}^+ + \text{H}_2\text{O}_2 \rightarrow 2\text{Fe}^{3+} + 2\text{H}_2\text{O}$
 (C) $\text{Na}_2\text{S} + 4\text{H}_2\text{O}_2 \rightarrow \text{Na}_2\text{SO}_4 + 4\text{H}_2\text{O}$
 (D) $2\text{NaOCl} + \text{H}_2\text{O}_2 \rightarrow 2\text{NaCl} + \text{H}_2\text{O} + \text{O}_2$

80. Which of the following artificial sweeteners has the highest sweetness value in comparison to cane sugar ?

- (A) Alitame (B) Sucralose
 (C) Saccharin (D) Aspartame

Chemistry - Section B · Numerical Value Questions

81. Zinc reacts with hydrochloric acid to give hydrogen and zinc chloride. The volume of hydrogen gas produced at STP from the reaction of 11.5 g of zinc with excess HCl is _____ L (Nearest integer)

(Given : Molar mass of Zn is 65.4 g mol^{-1} and Molar volume of H_2 at STP = 22.7 L)

82. The logarithm of equilibrium constant for the reaction $\text{Pd}^{2+} + 4\text{Cl}^- \rightleftharpoons \text{PdCl}_4^{2-}$ is _____ (Nearest integer)

Given :

$$\frac{2.303RT}{F} = 0.06 \text{ V}$$

$$\text{Pd}_{(\text{aq})}^{2+} + 2\text{e}^- \rightleftharpoons \text{Pd}(\text{s}) \quad E^\ominus = 0.83 \text{ V}$$

$$\text{PdCl}_4^{2-}(\text{aq}) + 2\text{e}^- \rightleftharpoons \text{Pd}(\text{s}) + 4\text{Cl}^-(\text{aq}) \quad E^\ominus = 0.65 \text{ V}$$

83. The enthalpy change for the conversion of $\frac{1}{2}\text{Cl}_2(\text{g})$ to $\text{Cl}^-(\text{aq})$ is $(-)$ _____ kJ mol^{-1} (Nearest integer)

Given :

$$\Delta_{\text{dis}} H_{\text{Cl}_2(\text{g})}^\ominus = 240 \text{ kJ mol}^{-1}, \Delta_{\text{eg}} H_{\text{Cl}(\text{g})}^\ominus = -350 \text{ kJ mol}^{-1}$$

,

$$\Delta_{\text{hyd}} H_{\text{Cl}(\text{g})}^\ominus = -380 \text{ kJ mol}^{-1}$$

84. On complete combustion, 0.492 g of an organic compound gave 0.792 g of CO_2 . The % of carbon in the organic compound is _____ (Nearest integer)

85. At 27°C , a solution containing 2.5 g of solute in 250.0 mL of solution exerts an osmotic pressure of 400 Pa. The molar mass of the solute is _____ g mol^{-1} (Nearest integer)

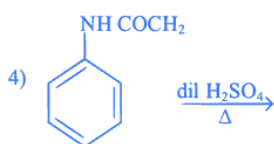
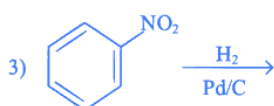
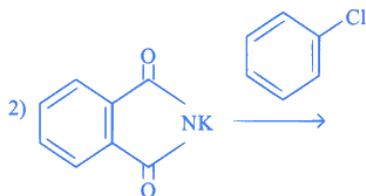
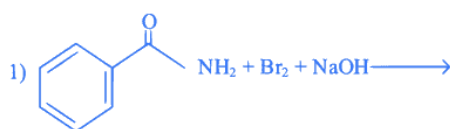
(Given :

$$R = 0.083 \text{ L bar K}^{-1} \text{ mol}^{-1}$$

)

86. The oxidation state of phosphorus in hypophosphoric acid is + _____.

87. How many of the transformations given below would result in aromatic amines?



88. $A \rightarrow B$

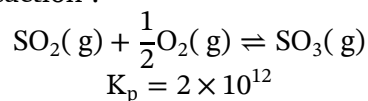
The rate constants of the above reaction at 200 K and 300 K are 0.03 min^{-1} and 0.05 min^{-1} respectively. The activation energy for the reaction is _____ J (Nearest integer)

(Given :

$$\begin{aligned} R &= 8.3 \text{ J K}^{-1} \text{ mol}^{-1} \\ \ln 10 &= 2.3 \\ \log 5 &= 0.70 \\ \log 3 &= 0.48 \\ \log 2 &= 0.30 \end{aligned}$$

)

89. For reaction :



at 27°C and 1 atm pressure. The K_c for the same reaction is _____ $\times 10^{13}$. (Nearest integer)

(Given

$$R = 0.082 \text{ L atm K}^{-1} \text{ mol}^{-1}$$

)

90. The total pressure of a mixture of non-reacting gases X(0.6 g) and Y(0.45 g) in a vessel is 740 mm of Hg.

The partial pressure of the gas X is _____ mm of Hg. (Nearest Integer)

(Given : molar mass

$$X = 20$$

and

$$Y = 45 \text{ g mol}^{-1}$$

)

- End of Question Paper -