

JEE Main 2023 Paper 1 (B.E./B.Tech) 11 April 2023 Shift 2 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2023 - Session 2**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 11 April 2023, Shift 2. The total duration is **3 hours**.
- (ii) The paper has **90 questions**, of which **75 are to be attempted**, and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 30 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **10 numerical value questions**; attempt **any 5** of them.
- (iv) **All Section A questions are compulsory**. Attempt only the specified number of Section B questions. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme** (Sections A and B): **+4** for a correct answer, **-1** for an incorrect answer, and **0** for an unanswered or marked-for-review question.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

1. Let a, b, c and d be positive real numbers such that $a + b + c + d = 11$. If the maximum value of $a^5 b^3 c^2 d$ is 3750β , then the value of β is

- (A) 110 (B) 108
(C) 90 (D) 55

2. The sum of the coefficients of three consecutive terms in the binomial expansion of $(1 + x)^{n+2}$, which are in the ratio $1 : 3 : 5$, is equal to :

- (A) 63 (B) 92
(C) 25 (D) 41

3. If the system of linear equations
 $7x + 11y + \alpha z = 13$

$$5x + 4y + 7z = \beta$$

$$175x + 194y + 57z = 361$$

has infinitely many solutions, then $\alpha + \beta + 2$ is equal to :

- (A) 6 (B) 4
(C) 5 (D) 3

4. For $a \in \mathbb{C}$, let

$$A = \{z \in \mathbb{C} : \operatorname{Re}(a + \bar{z}) > \operatorname{Im}(\bar{a} + z)\}$$

and

$$B = \{z \in \mathbb{C} : \operatorname{Re}(a + \bar{z}) < \operatorname{Im}(\bar{a} + z)\}$$

. Then among the two statements :

(S1): If

$$\operatorname{Re}(a), \operatorname{Im}(a) > 0$$

, then the set A contains all the real numbers

(S2) : If

$$\operatorname{Re}(a), \operatorname{Im}(a) < 0$$

, then the set B contains all the real numbers,

- (A) both are false (B) only (S1) is true
(C) only (S2) is true (D) both are true

5. If the letters of the word MATHS are permuted and all possible words so formed are arranged as in a dictionary with serial numbers, then the serial number of the word THAMS is :

- (A) 103 (B) 104
(C) 102 (D) 101

6. Let

$$A = \{1, 3, 4, 6, 9\}$$

and

$$B = \{2, 4, 5, 8, 10\}$$

. Let R be a relation defined on $A \times B$ such that

$$R = \{((a_1, b_1), (a_2, b_2)) : a_1 \leq b_2$$

and $b_1 \leq a_2\}$. Then the number of elements in the set R is :

- (A) 180 (B) 26
(C) 52 (D) 160

7. If the 1011^{th} term from the end in the binomial expansion of

$$\left(\frac{4x}{5} - \frac{5}{2x}\right)^{2022}$$

is 1024 times 1011^{th} R term from the beginning, then $|x|$ is equal to

- (A) $\frac{5}{16}$
(B) 8
(C) 12
(D) 15

8.

$$\begin{vmatrix} x+1 & x & x \\ x & x+\lambda & x \\ x & x & x+\lambda^2 \end{vmatrix} = \frac{9}{8}(103x + 81)$$

, then

$$\lambda, \frac{\lambda}{3}$$

are the roots of the equation :

- (A) $4x^2 + 24x - 27 = 0$
(B) $4x^2 - 24x + 27 = 0$
(C) $4x^2 - 24x - 27 = 0$
(D) $4x^2 + 24x + 27 = 0$

9. Let the mean of 6 observations 1, 2, 4, 5, x and y be 5 and their variance be 10. Then their mean deviation about the mean is equal to :

- (A) $\frac{10}{3}$
 (B) $\frac{8}{3}$
 (C) $\frac{7}{3}$
 (D) 3

10. If $f : \mathbb{R} \rightarrow \mathbb{R}$ be a continuous function satisfying

$$\int_0^{\frac{\pi}{2}} f(\sin 2x) \sin x dx + \alpha \int_0^{\frac{\pi}{4}} f(\cos 2x) \cos x dx = 0$$

, then the value of α is :

- (A) $-\sqrt{3}$
 (B) $\sqrt{2}$
 (C) $-\sqrt{2}$
 (D) $\sqrt{3}$

11. Let f and g be two functions defined by

$$f(x) = \begin{cases} x+1, & x < 0 \\ |x-1|, & x \geq 0 \end{cases}$$

and

$$g(x) = \begin{cases} x+1, & x < 0 \\ 1, & x \geq 0 \end{cases}$$

Then $(g \circ f)(x)$ is :

- (A) continuous everywhere but not differentiable at $x = 1$
 (B) differentiable everywhere
 (C) not continuous at $x = -1$
 (D) continuous everywhere but not differentiable exactly at one point

12. Let

$y = y(x)$ be the solution of the differential equation

$$\frac{dy}{dx} + \frac{5}{x(x^5+1)}y = \frac{(x^5+1)^2}{x^7}, x > 0$$

. If

$$y(1) = 2$$

, then $y(2)$ is equal to :

- (A) $\frac{693}{128}$
 (B) $\frac{697}{128}$
 (C) $\frac{637}{128}$
 (D) $\frac{679}{128}$

13. Let the function $f : [0, 2] \rightarrow \mathbb{R}$ be defined as

$$f(x) = \begin{cases} e^{\min\{x^2, x - [x]\}}, & x \in [0, 1) \\ e^{[x - \log_e x]}, & x \in [1, 2] \end{cases}$$

where $[t]$ denotes the greatest integer less than or equal to t . Then the value of the integral

$$\int_0^2 xf(x)dx$$

is :

- (A) $2e - 1$
 (B) $2e - \frac{1}{2}$
 (C) $1 + \frac{3e}{2}$
 (D) $(e-1)\left(e^2 + \frac{1}{2}\right)$

14. The domain of the function

$$f(x) = \frac{1}{\sqrt{[x]^2 - 3[x] - 10}}$$

is : (where $[x]$ denotes the greatest integer less than or equal to x)

- (A) $(-\infty, -2) \cup [6, \infty)$
 (B) $(-\infty, -3] \cup [6, \infty)$
 (C) $(-\infty, -2) \cup (5, \infty)$
 (D) $(-\infty, -3] \cup (5, \infty)$

15. If the radius of the largest circle with centre $(2,0)$ inscribed in the ellipse $x^2 + 4y^2 = 36$ is r , then $12r^2$ is equal to :

- (A) 72 (B) 92
 (C) 115 (D) 69

16. The converse of $((\sim p) \wedge q) \Rightarrow r$ is

- (A) $((\sim p) \vee q) \Rightarrow r$
 (B) $(\sim r) \Rightarrow p \wedge q$
 (C) $(p \vee (\sim q)) \Rightarrow (\sim r)$
 (D) $(\sim r) \Rightarrow ((\sim p) \wedge q)$

17. If four distinct points with position vectors $\vec{a}, \vec{b}, \vec{c}$ and $\vec{d}$ are coplanar, then $[\vec{a} \vec{b} \vec{c}]$ is equal to :

- (A) $[\vec{d} \vec{b} \vec{a}] + [\vec{a} \vec{c} \vec{d}] + [\vec{d} \vec{b} \vec{c}]$
 (B) $[\vec{b} \vec{c} \vec{d}] + [\vec{d} \vec{a} \vec{c}] + [\vec{d} \vec{b} \vec{a}]$
 (C) $[\vec{a} \vec{d} \vec{b}] + [\vec{d} \vec{c} \vec{a}] + [\vec{d} \vec{b} \vec{c}]$
 (D) $[\vec{d} \vec{c} \vec{a}] + [\vec{b} \vec{d} \vec{a}] + [\vec{c} \vec{d} \vec{b}]$

18. Let the line passing through the points $P(2, -1, 2)$ and $Q(5, 3, 4)$ meet the plane $x - y + z = 4$ at the point R. Then the distance of the point R from the plane $x + 2y + 3z + 2 = 0$ measured parallel to the line

$$\frac{x-7}{2} = \frac{y+3}{2} = \frac{z-2}{1}$$

is equal to :

- (A) $\sqrt{31}$
 (B) $\sqrt{189}$
 (C) $\sqrt{61}$
 (D) 3

19. Let P be the plane passing through the points $(5, 3, 0)$, $(13, 3, -2)$ and $(1, 6, 2)$. For $\alpha \in \mathbb{N}$, if the distances of the points $A(3, 4, \alpha)$ and $B(2, \alpha, a)$ from the plane P are 2 and 3 respectively, then the positive value of a is :

- (A) 6 (B) 4
 (C) 5 (D) 3

20. The angle of elevation of the top P of a tower from the feet of one person standing due South of the tower is 45° and from the feet of another person standing due west of the tower is 30° . If the height of the tower is 5 meters, then the distance (in meters) between the two persons is equal to

- (A) 10
 (B) $\frac{5}{2}\sqrt{5}$
 (C) $5\sqrt{5}$
 (D) 5

21. Let

$$\vec{a} = \hat{i} + 2\hat{j} + 3\hat{k}$$

and

$$\vec{b} = \hat{i} + \hat{j} - \hat{k}$$

. If $\vec{c}$ is a vector such that

$$\vec{a} \cdot \vec{c} = 11, \vec{b} \cdot (\vec{a} \times \vec{c}) = 27$$

and

$$\vec{b} \cdot \vec{c} = -\sqrt{3} |\vec{b}|$$

, then $|\vec{a} \times \vec{c}|^2$ is equal to _____.

22. Let

$$A = \{1, 2, 3, 4, 5\}$$

and

$$B = \{1, 2, 3, 4, 5, 6\}$$

. Then the number of functions $f : A \rightarrow B$ satisfying

$$f(1) + f(2) = f(4) - 1$$

is equal to _____.

23. Let the probability of getting head for a biased coin be $\frac{1}{4}$. It is tossed repeatedly until a head appears. Let N be the number of tosses required. If the probability that the equation

$$64x^2 + 5Nx + 1 = 0$$

has no real root is

$$\frac{p}{q}$$

, where p and q are coprime, then $q - p$ is equal to _____.

24. If A is the area in the first quadrant enclosed by the curve

$$C : 2x^2 - y + 1 = 0$$

, the tangent to C at the point $(1, 3)$ and the line

$$x + y = 1$$

, then the value of $60A$ is _____.

25. Let

$$S = \left\{ z \in \mathbb{C} - \{i, 2i\} : \frac{z^2 + 8iz - 15}{z^2 - 3iz - 2} \in \mathbb{R} \right\}$$

. If

$$\alpha - \frac{13}{11}i \in S, \alpha \in \mathbb{R} - \{0\}$$

, then $242\alpha^2$ is equal to _____.

26. The number of points, where the curve $f(x) = e^{8x} - e^{6x} - 3e^{4x} - e^{2x} + 1, x \in \mathbb{R}$ cuts x -axis, is equal to _____.

27. Let the line

$$l : x = \frac{1-y}{-2} = \frac{z-3}{\lambda}, \lambda \in \mathbb{R}$$

meet the plane

$$P : x + 2y + 3z = 4$$

at the point (α, β, γ) . If the angle between the line l and the plane P is

$$\cos^{-1}\left(\sqrt{\frac{5}{14}}\right)$$

, then $\alpha + 2\beta + 6\gamma$ is equal to _____.

28. For $k \in \mathbb{N}$, if the sum of the series

$$1 + \frac{4}{k} + \frac{8}{k^2} + \frac{13}{k^3} + \frac{19}{k^4} + \dots$$

is 10, then the value of k is _____.

29. Let the tangent to the parabola $y^2 = 12x$

at the point $(3, \alpha)$ be perpendicular to the line $2x + 2y = 3$. Then the square of distance of the point $(6, -4)$ from the normal to the hyperbola $\alpha^2 x^2 - 9y^2 = 9\alpha^2$ at its point $(\alpha - 1, \alpha + 2)$ is equal to _____.

30. If the line

$$l_1 : 3y - 2x = 3$$

is the angular bisector of the lines

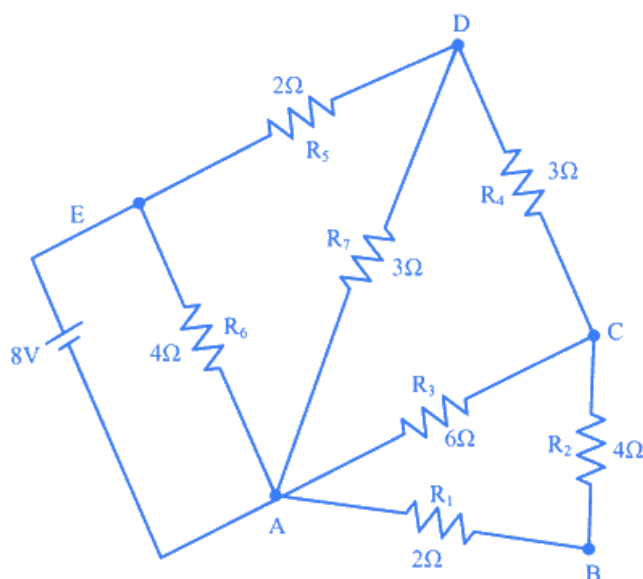
$$l_2 : x - y + 1 = 0$$

and

$$l_3 : \alpha x + \beta y + 17 = 0$$

, then $\alpha^2 + \beta^2 - \alpha - \beta$ is equal to _____.

31.


 The current flowing through R_2 is :

- (A) $\frac{1}{2}$ A
 (B) $\frac{1}{3}$ A
 (C) $\frac{1}{4}$ A
 (D) $\frac{2}{3}$ A

32. A plane electromagnetic wave of frequency 20 MHz propagates in free space along x-direction. At a particular space and time,

$$\vec{E} = 6.6\hat{j} \text{ V/m}$$

. What is $\vec{B}$ at this point?

- (A) $-2.2 \times 10^{-8}\hat{i}T$
 (B) $2.2 \times 10^{-8}\hat{i}T$
 (C) $2.2 \times 10^{-8}\hat{k}T$
 (D) $-2.2 \times 10^{-8}\hat{k}T$

33. An electron is allowed to move with constant velocity along the axis of current carrying straight solenoid.

- A. The electron will experience magnetic force along the axis of the solenoid.
 B. The electron will not experience magnetic force.
 C. The electron will continue to move along the axis of the solenoid.
 D. The electron will be accelerated along the axis of the solenoid.
 E. The electron will follow parabolic path-inside the solenoid.

Choose the correct answer from the options given below:

- (A) B, C and D only (B) B and C only
 (C) A and D only (D) B and E only

34. The Thermodynamic process, in which internal energy of the system remains constant is

- (A) Isobaric (B) Isochoric
 (C) Adiabatic (D) Isothermal

35. Given below are two statements: one is labelled as Assertion **A** and the other is labelled as Reason **R**

Assertion A: A bar magnet dropped through a metallic cylindrical pipe takes more time to come down compared to a non-magnetic bar with same geometry and mass.

Reason R: For the magnetic bar, Eddy currents are produced in the metallic pipe which oppose the motion of the magnetic bar.

In the light of the above statements, choose the correct answer from the options given below

- (A) A is true but R is false (B) A is false but R is true
 (C) Both A and R are true but R is NOT the correct explanation of A (D) Both A and R are true and R is the correct explanation of A

36. The energy of He^+ ion in its first excited state is, (The ground state energy for the Hydrogen atom is -13.6 eV) :

- (A) -13.6 eV
- (B) -27.2 eV
- (C) -3.4 eV
- (D) -54.4 eV

37. When one light ray is reflected from a plane mirror with 30° angle of reflection, the angle of deviation of the ray after reflection is :

- (A) 140°
- (B) 130°
- (C) 120°
- (D) 110°

38. A capacitor of capacitance C is charged to a potential V . The flux of the electric field through a closed surface enclosing the positive plate of the capacitor is :

- (A) Zero
- (B) $\frac{CV}{\epsilon_0}$
- (C) $\frac{CV}{2\epsilon_0}$
- (D) $\frac{2CV}{\epsilon_0}$

39. The ratio of the de-Broglie wavelengths of proton and electron having same Kinetic energy : (Assume

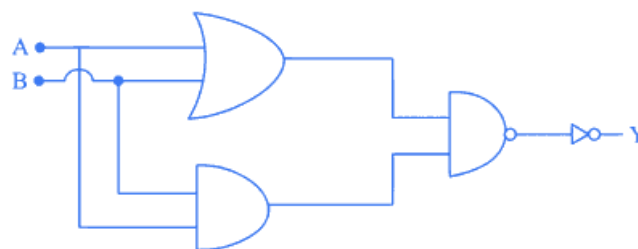
$$m_p = m_e \times 1849$$

-)
- (A) 1:43
 - (B) 1:62
 - (C) 2:43
 - (D) 1:30

40. Eight equal drops of water are falling through air with a steady speed of 10 cm/s . If the drops coalesce, the new velocity is:-

- (A) 40 cm/s
- (B) 16 cm/s
- (C) 10 cm/s
- (D) 5 cm/s

41. The logic operations performed by the given digital circuit is equivalent to:



- (A) NOR
- (B) AND
- (C) NAND
- (D) OR

42. A projectile is projected at 30° from horizontal with initial velocity 40 ms^{-1} . The velocity of the projectile at

from the start will be : (Given $t = 2 \text{ s}$
 $g = 10 \text{ m/s}^2$)

- (A) $20\sqrt{3} \text{ ms}^{-1}$
- (B) Zero
- (C) 20 ms^{-1}
- (D) $40\sqrt{3} \text{ ms}^{-1}$

43. A body of mass 500 g moves along x -axis such that its velocity varies with displacement x according to the relation

$$v = 10\sqrt{x} \text{ m/s}$$

the force acting on the body is:-

- (A) 166 N
- (B) 5 N
- (C) 25 N
- (D) 125 N

44. A space ship of mass 2×10^4 kg is launched into a circular orbit close to the earth surface. The additional velocity to be imparted to the space ship in the orbit to overcome the gravitational pull will be (if

$$g = 10 \text{ m/s}^2$$

and radius of earth

$$= 6400 \text{ km}$$

):

- (A) $7.9(\sqrt{2} - 1)\text{km/s}$
- (B) $11.2(\sqrt{2} - 1)\text{km/s}$
- (C) $7.4(\sqrt{2} - 1)\text{km/s}$
- (D) $8(\sqrt{2} - 1)\text{km/s}$

45. When vector

$$\vec{A} = 2\hat{i} + 3\hat{j} + 2\hat{k}$$

is subtracted from vector $\vec{B}$, it gives a vector equal to $2\hat{j}$. Then the magnitude of vector $\vec{B}$ will be :

- (A) 3
- (B) $\sqrt{33}$
- (C) $\sqrt{6}$
- (D) $\sqrt{5}$

46. If force (F), velocity (V) and time (T) are considered as fundamental physical quantity, then dimensional formula of density will be :

- (A) $FV^{-2} T^2$
- (B) $FV^4 T^{-6}$
- (C) $F^2 V^{-2} T^6$
- (D) $FV^{-4} T^{-2}$

47. If V is the gravitational potential due to sphere of uniform density on it's surface, then it's value at the center of sphere will be:-

- (A) $\frac{3V}{2}$
- (B) $\frac{V}{2}$
- (C) $\frac{4}{3}V$
- (D) V

48. The root mean square speed of molecules of nitrogen gas at 27°C is approximately : (Given mass of a nitrogen molecule

$$= 4.6 \times 10^{-26} \text{ kg}$$

and take Boltzmann constant

$$k_B = 1.4 \times 10^{-23} \text{ JK}^{-1}$$

)

- (A) 91 m/s
- (B) 1260 m/s
- (C) 27.4 m/s
- (D) 523 m/s

49. A car P travelling at 20 ms^{-1} sounds its horn at a frequency of 400 Hz. Another car Q is travelling behind the first car in the same direction with a velocity 40 ms^{-1} . The frequency heard by the passenger of the car Q is approximately [Take, velocity of sound

$$= 360 \text{ ms}^{-1}$$

]

- (A) 485 Hz
- (B) 514 Hz
- (C) 421 Hz
- (D) 471 Hz

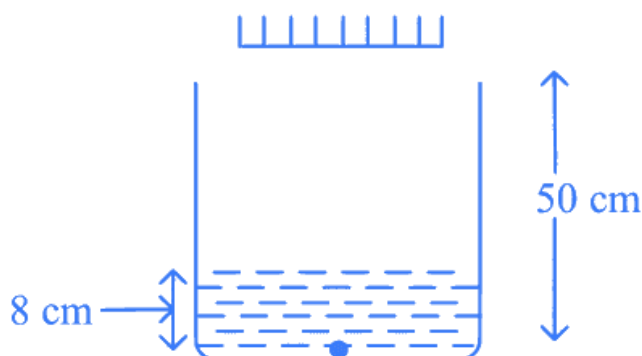
50. In satellite communication, the uplink frequency band used is :

- (A) 3.7 – 4.2 GHz
- (B) 5.925 – 6.425 GHz
- (C) 76 – 88 MHz
- (D) 420 – 890 MHz

51. As shown in the figure, a plane mirror is fixed at a height of 50 cm from the bottom of tank containing water

$$\left(\mu = \frac{4}{3}\right)$$

. The height of water in the tank is 8 cm. A small bulb is placed at the bottom of the water tank. The distance of image of the bulb formed by mirror from the bottom of the tank is _____ cm.



52. A circular plate is rotating in horizontal plane, about an axis passing through its center and perpendicular to the plate, with an angular velocity ω . A person sits at the center having two dumbbells in his hands. When he stretches out his hands, the moment of inertia of the system becomes triple. If E be the initial Kinetic energy of the system, then final Kinetic energy will be $\frac{E}{x}$. The value of x is _____

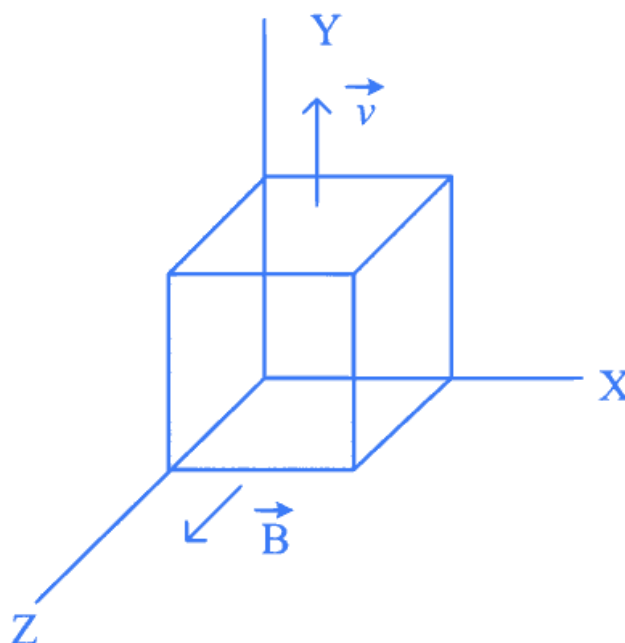
53. The surface tension of soap solution is $3.5 \times 10^{-2} \text{ Nm}^{-1}$. The amount of work done required to increase the radius of soap bubble from 10 cm to 20 cm is _____ $\times 10^{-4} \text{ J}$.
(take $\pi = 22/7$)

54. A wire of density $8 \times 10^3 \text{ kg/m}^3$ is stretched between two clamps 0.5 m apart. The extension developed in the wire is $3.2 \times 10^{-4} \text{ m}$. If

$$Y = 8 \times 10^{10} \text{ N/m}^2$$

, the fundamental frequency of vibration in the wire will be _____ Hz.

55. A metallic cube of side 15 cm moving along y -axis at a uniform velocity of 2 ms^{-1} . In a region of uniform magnetic field of magnitude 0.5 T directed along z -axis. In equilibrium the potential difference between the faces of higher and lower potential developed because of the motion through the field will be _____ mV.



56. A nucleus disintegrates into two nuclear parts, in such a way that ratio of their nuclear sizes is $1 : 2^{1/3}$. Their respective speed have a ratio of $n : 1$. The value of n is _____.

57. A block of mass 5 kg starting from rest pulled up on a smooth incline plane making an angle of 30° with horizontal with an effective acceleration of 1 ms^{-2} . The power delivered by the pulling force at

from the start is _____ W.

[use

$$g = 10 \text{ ms}^{-2}$$

]

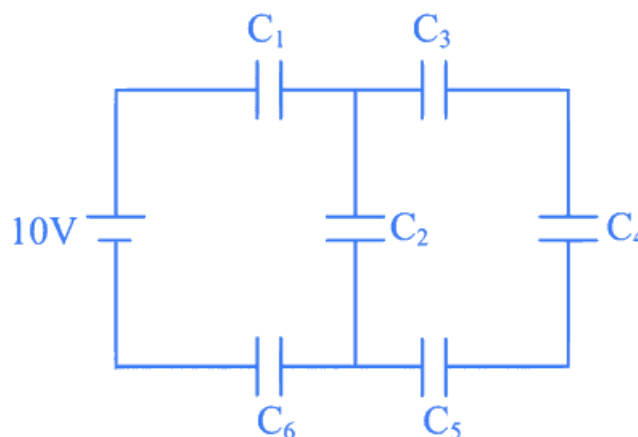
(calculate the nearest integer value)

58. In the given circuit,

$$C_1 = 2\mu\text{F}, C_2 = 0.2\mu\text{F}, C_3 = 2\mu\text{F}, C_4 = 4\mu\text{F},$$

$$C_5 = 2\mu\text{F}, C_6 = 2\mu\text{F}$$

, The charge stored on capacitor C_4 is _____ μC .



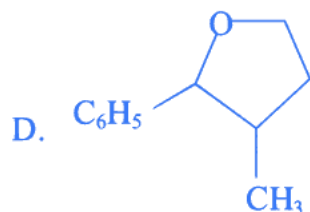
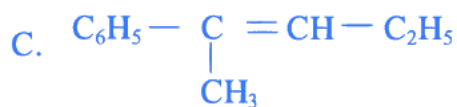
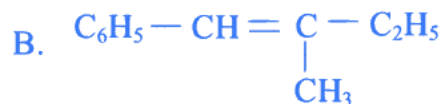
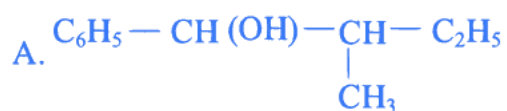
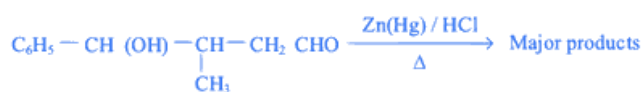
59. A coil has an inductance of 2H and resistance of 4Ω . A 10V is applied across the coil. The energy stored in the magnetic field after the current has built up to its equilibrium value will be _____ $\times 10^{-2}\text{J}$.

60. Two identical cells each of emf 1.5V are connected in series across a 10Ω resistance. An ideal voltmeter connected across 10Ω resistance reads 1.5V . The internal resistance of each cell is _____ Ω .

61. Which one of the following pairs is an example of polar molecular solids?

- (A) HCl(s), AlN(s)
 (B) MgO(s), SO₂(s)
 (C) SO₂(s), NH₃(s)
 (D) SO₂(s), CO₂(s)

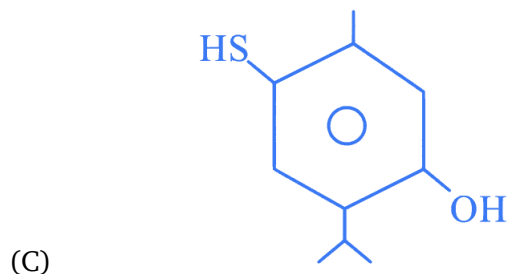
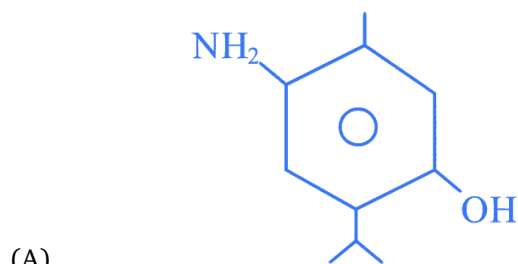
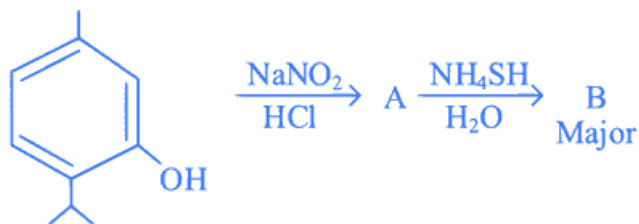
62. The major product formed in the following reaction is



Choose the correct answer from the options given below :

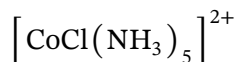
- (A) D only (B) B only
 (C) A only (D) C only

63. Compound 'B' is



64. Given below are two statements, one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A :



absorbs at lower wavelength of light with respect to $[\text{Co}(\text{NH}_3)_5(\text{H}_2\text{O})]^{3+}$

Reason R : It is because the wavelength of the light absorbed depends on the oxidation state of the metal ion.

In the light of the above statements, choose the correct answer from the options given below:

- (A) A is false but R is true
(B) Both A and R are true but R is NOT the correct explanation of A
(C) Both A and R are true and R is the correct explanation of A
(D) A is true but R is false

65. If Ni^{2+} is replaced by Pt^{2+} in the complex $[\text{NiCl}_2\text{Br}_2]^{2-}$, which of the following properties are expected to get changed ?

- A. Geometry
B. Geometrical isomerism
C. Optical isomerism
D. Magnetic properties

- (A) A and D
(B) A, B and D
(C) A, B and C
(D) B and C

66. Given below are two statements, one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A : A solution of the product obtained by heating a mole of glycine with a mole of chlorine in presence of red phosphorous generates chiral carbon atom.

Reason R : A molecule with 2 chiral carbons is always optically active.

In the light of the above statements, choose the correct answer from the options given below:

- (A) Both A and R are true and R is the correct explanation of A
(B) Both A and R are true but R is NOT the correct explanation of A
(C) A is true but R is false
(D) A is false but R is true

67. Match List I with List II

	LIST I Complex		LIST II Colour
A.	$\text{Mg}(\text{NH}_4)\text{PO}_4$	I.	brown
B.	$\text{K}_3[\text{Co}(\text{NO}_2)_6]$	II.	white
C.	$\text{MnO}(\text{OH})_2$	III.	yellow
D.	$\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$	IV.	blue

Choose the correct answer from the options given below :

- (A) A-II, B-IV, C-I, D-III
(B) A-II, B-III, C-I, D-IV
(C) A-III, B-IV, C-II, D-I
(D) A-II, B-III, C-IV, D-I

68. What weight of glucose must be dissolved in 100 g of water to lower the vapour pressure by 0.20 mm Hg ?

(Assume dilute solution is being formed)

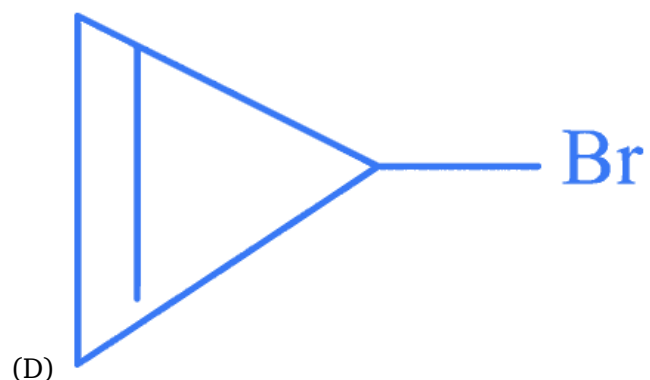
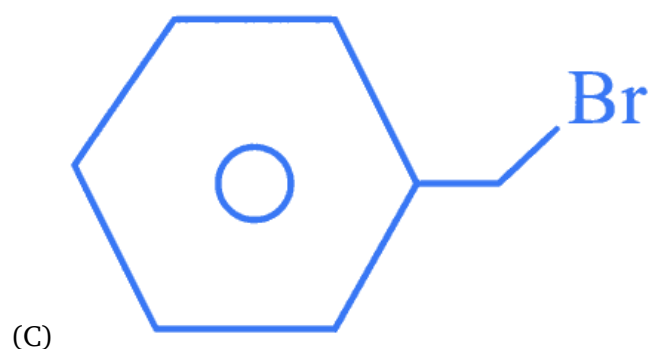
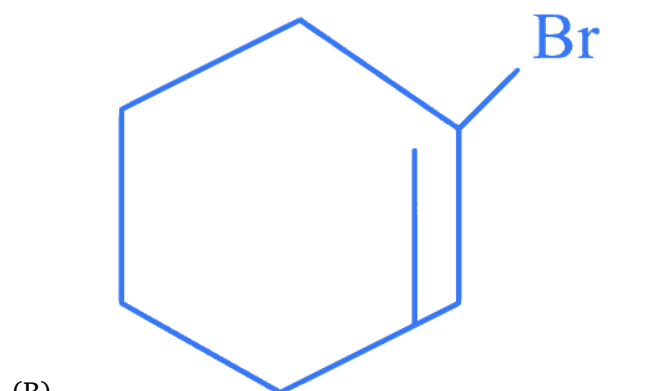
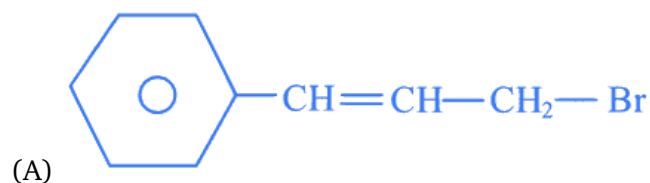
Given : Vapour pressure of pure water is 54.2 mm Hg at room temperature. Molar mass of glucose is 180 g mol^{-1}

- (A) 3.69 g
(B) 2.59 g
(C) 3.59 g
(D) 4.69 g

69. A solution is prepared by adding 2 g of "X" to 1 mole of water. Mass percent of "X" in the solution is :

- (A) 20% (B) 10%
(C) 2% (D) 5%

70. Compound from the following that will not produce precipitate on reaction with AgNO_3 is :



71. The magnetic moment is measured in Bohr Magnetons (BM).

Spin only magnetic moment of Fe in $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ complexes respectively is :

- (A) 3.87 B.M. and 1.732 B.M. (B) 5.92 B.M. and 1.732 B.M.
(C) 6.92 B.M. in both (D) 4.89 B.M. and 6.92 B.M.

72. Given below are two statements, one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A: can be subjected to Wolff-Kishner reduction to give



Reason R: Wolff-Kishner reduction is used to convert into

In the light of the above statements, choose the correct answer from the options given below:

- (A) Both A and R are true and R is the correct explanation of A
(B) A is false but R is true
(C) A is true but R is false
(D) Both A and R are true but R is NOT the correct explanation of A

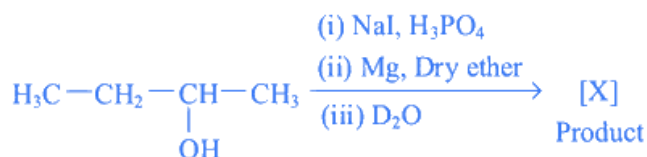
73. For a chemical reaction $\text{A} + \text{B} \rightarrow \text{Product}$, the order is 1 with respect to A and B.

Rate $\text{mol L}^{-1} \text{S}^{-1}$	[A] mol L^{-1}	[B] mol L^{-1}
0.10	20	0.5
0.40	x	0.5
0.80	40	y

What is the value of x and y ?

- (A) 80 and 4 (B) 160 and 4
(C) 80 and 2 (D) 40 and 4

74.



Product [X] formed in the above reaction is :

- (A)
$$\begin{array}{c} \text{H} \\ | \\ \text{H}_3\text{C}-\text{CH}_2-\text{C}-\text{CH}_3 \\ | \\ \text{OH} \end{array}$$
- (B)
$$\begin{array}{c} \text{H}_3\text{C}-\text{CH}_2-\underset{\text{D}}{\text{CH}}-\text{CH}_3 \end{array}$$
- (C) $\text{H}_3\text{C}-\text{CH}=\text{CH}-\text{CH}_3$
- (D) $\text{H}_3\text{C}-\text{CH}_2-\text{CH}=\text{CH}_2$

75. Alkali metal from the following with least melting point is :

- (A) Rb
(B) Na
(C) K
(D) Cs

76. Given below are two statements :
Statement I : In the metallurgy process, sulphide ore is converted to oxide before reduction.
Statement II : Oxide ores in general are easier to reduce.
In the light of the above statements, choose the most appropriate answer from the options given below:

- (A) Both Statement I and Statement II are correct
(B) Both Statement I and Statement II are incorrect
(C) Statement I is correct but Statement II is incorrect
(D) Statement I is incorrect but Statement II is correct

77. Which of the following compounds is an example of Freon?

- (A) $\text{C}_2\text{H}_2\text{F}_2$
(B) C_2HF_3
(C) C_2F_4
(D) $\text{C}_2\text{Cl}_2\text{F}_2$

78. Which hydride among the following is less stable?

- (A) NH_3
(B) HF
(C) LiH
(D) BeH_2

79. Given below are two statements :

Statement I : Ethene at 333 to 343 K and 6-7 atm pressure in the presence of AlEt_3 and TiCl_4 undergoes addition polymerization to give LDP.
Statement II : Caprolactam at 533-543 K in H_2O through step growth polymerizes to give Nylon 6.
In the light of the above statements, choose the correct answer from the options given below:

- (A) Both Statement I and Statement II are true
(B) Statement I is true but Statement II is false
(C) Statement I is false but Statement II is true
(D) Both Statement I and Statement II are false

80. One mole of P_4 reacts with 8 moles of SOCl_2 to give 4 moles of A, x mole of SO_2 and 2 moles of B. A, B and x respectively are :

- (A) POCl_3 , S_2Cl_2 and 4
(B) PCl_3 , S_2Cl_2 and 4
(C) POCl_3 , S_2Cl_2 and 2
(D) PCl_3 , S_2Cl_2 and 2

81. The number of possible isomeric products formed when 3-chloro-1-butene reacts with HCl through carbocation formation is _____.

82. The maximum number of lone pairs of electrons on the central atom from the following species is _____.
 ClO_3^- , XeF_4 , SF_4 and I_3^-

83. The total number of intensive properties from the following is _____.
 Volume, Molar heat capacity, Molarity, E^θ cell, Gibbs free energy change, Molar mass, Mole

84. Number of compounds from the following which will not produce orange red precipitate with Benedict solution is _____.
 Glucose, maltose, sucrose, ribose, 2-deoxyribose, amylose, lactose

85. The number of correct statements from the following is _____.
 A. For 1s orbital, the probability density is maximum at the nucleus
 B. For 2s orbital, the probability density first increases to maximum and then decreases sharply to zero.
 C. Boundary surface diagrams of the orbitals encloses a region of 100% probability of finding the electron.
 D. p and d-orbitals have 1 and 2 angular nodes respectively.
 E. probability density of p-orbital is zero at the nucleus

86. 4.5 moles each of hydrogen and iodine is heated in a sealed ten litre vessel. At equilibrium, 3 moles of HI were found. The equilibrium constant for $\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \rightleftharpoons 2\text{HI}(\text{g})$ is _____.

87. The number of correct statements from the following is _____

- A. E_{cell} is an intensive parameter
- B. A negative $E^\ominus$ means that the redox couple is a stronger reducing agent than the H^+/H_2 couple.
- C. The amount of electricity required for oxidation or reduction depends on the stoichiometry of the electrode reaction.
- D. The amount of chemical reaction which occurs at any electrode during electrolysis by a current is proportional to the quantity of electricity passed through the electrolyte.

88. The volume of hydrogen liberated at STP by treating 2.4 g of magnesium with excess of hydrochloric acid is _____ $\times 10^{-2}$ L
 Given : Molar volume of gas is 22.4 L at STP
 Molar mass of magnesium is 24 g mol^{-1}

89. The number of correct statements about modern adsorption theory of heterogeneous catalysis from the following is _____

- A. The catalyst is diffused over the surface of reactants.
- B. Reactants are adsorbed on the surface of the catalyst.
- C. Occurrence of chemical reaction on the catalyst's surface through formation of an intermediate.
- D. It is a combination of intermediate compound formation theory and the old adsorption theory.
- E. It explains the action of the catalyst as well as those of catalytic promoters and poisons.

90. $\text{Mg}(\text{NO}_3)_2 \cdot \text{XH}_2\text{O}$ and $\text{Ba}(\text{NO}_3)_2 \cdot \text{YH}_2\text{O}$, represent formula of the crystalline forms of nitrate salts. Sum of X and Y is _____.

- End of Question Paper -