

JEE Main 2025 Paper 1 (B.E./B.Tech) 4 April 2025 Shift 2 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2025 - Session 2**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 4 April 2025, Shift 2. The total duration is **3 hours**.
- (ii) The paper has **75 questions** and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 25 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **5 numerical value questions**.
- (iv) **All questions are compulsory**. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme** (Sections A and B): **+4** for a correct answer, **–1** for an incorrect answer, and **0** for an unanswred or marked-for-review question.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

1. Let $A = \{-3, -2, -1, 0, 1, 2, 3\}$ and R be a relation on A defined by xRy if and only if $2x - y \in \{0, 1\}$. Let l be the number of elements in R . Let m and n be the minimum number of elements required to be added in R to make it reflexive and symmetric relations, respectively. Then $l + m + n$ is equal to:

- (A) 17 (B) 18
(C) 15 (D) 16

2. Let the domains of the functions $f(x) = \log_4 \log_3 \log_7 (8 - \log_2 (x^2 + 4x + 5))$ and $g(x) = \sin^{-1} \left(\frac{7x+10}{x-2} \right)$ be (α, β) and $[\gamma, \delta]$, respectively. Then $\alpha^2 + \beta^2 + \gamma^2 + \delta^2$ is equal to:

- (A) 15 (B) 13
(C) 16 (D) 14

3. Let the mean and the standard deviation of the observation 2, 3, 3, 4, 5, 7, a, b be 4 and $\sqrt{2}$ respectively. Then the mean deviation about the mode of these observations is:

- (A) $\frac{1}{2}$ (B) $\frac{3}{4}$
(C) 1 (D) 2

4. Let the sum of the focal distances of the point $P(4, 3)$ on the hyperbola $H: \frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ be $8\sqrt{\frac{5}{3}}$. If for H , the length of the latus rectum is l and the product of the focal distances of the point P is m , then $9l^2 + 6m$ is equal to:

- (A) 187 (B) 184
(C) 186 (D) 185

5. Let $a > 0$. If the function $f(x) = 6x^3 - 45ax^2 + 108a^2x + 1$ attains its local maximum and minimum values at the points x_1 and x_2 respectively such that $x_1x_2 = 54$, then $a + x_1 + x_2$ is equal to:

- (A) 15 (B) 13
(C) 24 (D) 18

6. Let the matrix $A = \begin{bmatrix} 1 & 0 & 0 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{bmatrix}$ satisfy $A^n = A^{n-2} + A^2 - I$ for $n \geq 3$. Then the sum of all the elements of A^{50} is:

- (A) 44 (B) 39
(C) 52 (D) 53

7. If a curve $y = y(x)$ passes through the point $(1, \frac{\pi}{2})$ and satisfies the differential equation $(7x^4 \cot y - e^x \operatorname{cosec} y) \frac{dx}{dy} = x^5, x \geq 1$, then at $x = 2$, the value of $\cos y$ is:

- (A) $\frac{2e^2+e}{64}$ (B) $\frac{2e^2-e}{64}$
(C) $\frac{2e^2-e}{128}$ (D) $\frac{2e^2+e}{128}$

8. If the sum of the first 20 terms of the series $\frac{4 \cdot 1}{4+3 \cdot 1^2+1^4} + \frac{4 \cdot 2}{4+3 \cdot 2^2+2^4} + \frac{4 \cdot 3}{4+3 \cdot 3^2+3^4} + \frac{4 \cdot 4}{4+3 \cdot 4^2+4^4} + \dots$ is $\frac{m}{n}$, where m and n are coprime, then $m + n$ is equal to:

- (A) 423 (B) 421
(C) 422 (D) 420

9. Let for two distinct values of p the lines $y = x + p$ touch the ellipse $E: \frac{x^2}{4^2} + \frac{y^2}{3^2} = 1$ at the points A and B . Let the line $y = x$ intersect E at the points C and D . Then the area of the quadrilateral $ABCD$ is equal to:

- (A) 48 (B) 20
(C) 24 (D) 36

10. The centre of a circle C is at the centre of the ellipse $E: \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1, a > b$. Let C pass through the foci F_1 and F_2 of E such that the circle C and the ellipse E intersect at four points. Let P be one of these four points. If the area of the triangle PF_1F_2 is 30 and the length of the major axis of E is 17, then the distance between the foci of E is:

- (A) 12 (B) 26
(C) 13 (D) $\frac{13}{2}$

11. A line passing through the point $A(-2, 0)$, touches the parabola $P : y^2 = x - 2$ at the point B in the first quadrant. The area, of the region bounded by the line AB , parabola P and the x -axis, is :

- (A) 3 (B) $\frac{7}{3}$
(C) $\frac{8}{3}$ (D) 2

12. Let the product of $\omega_1 = (8 + i) \sin \theta + (7 + 4i) \cos \theta$ and $\omega_2 = (1 + 8i) \sin \theta + (4 + 7i) \cos \theta$ be $\alpha + i\beta$, $i = \sqrt{-1}$. Let p and q be the maximum and the minimum values of $\alpha + \beta$ respectively. Then $p + q$ is equal to :

- (A) 130 (B) 150
(C) 160 (D) 140

13. If

$$1^2 \cdot ({}^{15}C_1) + 2^2 \cdot ({}^{15}C_2) + 3^2 \cdot ({}^{15}C_3) + \dots + 15^2 \cdot ({}^{15}C_{15}) = 2^m \cdot 3^n \cdot 5^k$$

, where $m, n, k \in \mathbf{N}$, then $m + n + k$ is equal to :

- (A) 20 (B) 19
(C) 18 (D) 21

14. Let A be the point of intersection of the lines $L_1 : \frac{x-7}{1} = \frac{y-5}{0} = \frac{z-3}{-1}$ and $L_2 : \frac{x-1}{3} = \frac{y+3}{4} = \frac{z+7}{5}$. Let B and C be the points on the lines L_1 and L_2 respectively such that $AB = AC = \sqrt{15}$. Then the square of the area of the triangle ABC is :

- (A) 63 (B) 57
(C) 60 (D) 54

15. The sum of the infinite series

$$\cot^{-1}\left(\frac{7}{4}\right) + \cot^{-1}\left(\frac{19}{4}\right) + \cot^{-1}\left(\frac{39}{4}\right) + \cot^{-1}\left(\frac{67}{4}\right) + \dots$$

is :

- (A) $\frac{\pi}{2} + \cot^{-1}\left(\frac{1}{2}\right)$ (B) $\frac{\pi}{2} - \cot^{-1}\left(\frac{1}{2}\right)$
(C) $\frac{\pi}{2} - \tan^{-1}\left(\frac{1}{2}\right)$ (D) $\frac{\pi}{2} + \tan^{-1}\left(\frac{1}{2}\right)$

16. The axis of a parabola is the line $y = x$ and its vertex and focus are in the first quadrant at distances $\sqrt{2}$ and $2\sqrt{2}$ units from the origin, respectively. If the point $(1, k)$ lies on the parabola, then a possible value of k is :

- (A) 8 (B) 3
(C) 9 (D) 4

17. Let $f(x) + 2f\left(\frac{1}{x}\right) = x^2 + 5$ and $2g(x) - 3g\left(\frac{1}{x}\right) = x, x > 0$. If $\alpha = \int_1^2 f(x)dx$, and $\beta = \int_1^2 g(x)dx$, then the value of $9\alpha + \beta$ is :

- (A) 0 (B) 10
(C) 1 (D) 11

18. Consider two sets A and B , each containing three numbers in A.P. Let the sum and the product of the elements of A be 36 and p respectively and the sum and the product of the elements of B be 36 and q respectively. Let d and D be the common differences of AP's in A and B respectively such that $D = d + 3, d > 0$. If $\frac{p+q}{p-q} = \frac{19}{5}$, then $p - q$ is equal to

- (A) 540 (B) 450
(C) 600 (D) 630

19. Let the values of p , for which the shortest distance between the lines $\frac{x+1}{3} = \frac{y}{4} = \frac{z}{5}$ and $\vec{r} = (p\hat{i} + 2\hat{j} + \hat{k}) + \lambda(2\hat{i} + 3\hat{j} + 4\hat{k})$ is $\frac{1}{\sqrt{6}}$, be $a, b, (a < b)$. Then the length of the latus rectum of the ellipse $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ is :

- (A) $\frac{3}{2}$ (B) 9
(C) 18 (D) $\frac{2}{3}$

20. Let f be a differentiable function on $\mathbf{R}$ such that $f(2) = 1, f'(2) = 4$. Let $\lim_{x \rightarrow 0} (f(2+x))^{3/x} = e^\alpha$. Then the number of times the curve $y = 4x^3 - 4x^2 - 4(\alpha - 7)x - \alpha$ meets x -axis is :

- (A) 3 (B) 1
(C) 2 (D) 0

21. If

$$\int \frac{(\sqrt{1+x^2} + x)^{10}}{(\sqrt{1+x^2} - x)^9} dx = \frac{1}{m} \left((\sqrt{1+x^2} + x)^n (n\sqrt{1+x^2} - x) \right) + C$$

where C is the constant of integration and $m, n \in \mathbf{N}$, then $m + n$ is equal to _____.

22. A card from a pack of 52 cards is lost. From the remaining 51 cards, n cards are drawn and are found to be spades. If the probability of the lost card to be a spade is $\frac{11}{50}$, then n is equal to _____.

23. Let the three sides of a triangle ABC be given by the vectors $2\hat{i} - \hat{j} + \hat{k}$, $\hat{i} - 3\hat{j} - 5\hat{k}$ and $3\hat{i} - 4\hat{j} - 4\hat{k}$. Let G be the centroid of the triangle ABC. Then $6 \left(|\overrightarrow{AG}|^2 + |\overrightarrow{BG}|^2 + |\overrightarrow{CG}|^2 \right)$ is equal to _____.

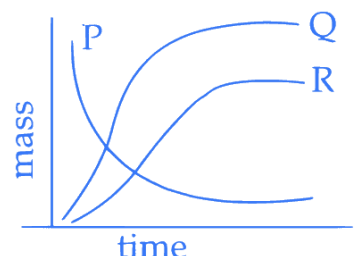
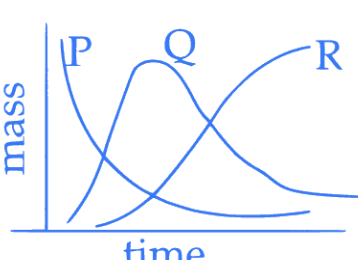
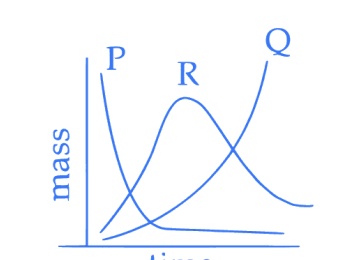
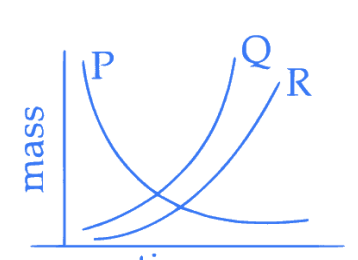
24. If α is a root of the equation $x^2 + x + 1 = 0$ and $\sum_{k=1}^n \left(\alpha^k + \frac{1}{\alpha^k} \right)^2 = 20$, then n is equal to _____.

25. Let m and n , ($m < n$), be two 2-digit numbers. Then the total numbers of pairs (m, n) , such that $\gcd(m, n) = 6$, is _____.

26. A finite size object is placed normal to the principal axis at a distance of 30 cm from a convex mirror of focal length 30 cm. A plane mirror is now placed in such a way that the image produced by both the mirrors coincide with each other. The distance between the two mirrors is :

- (A) 45 cm (B) 15 cm
(C) 22.5 cm (D) 7.5 cm

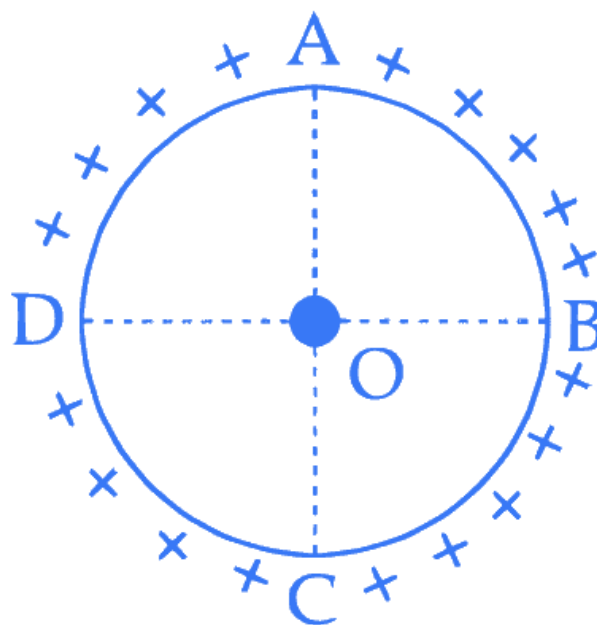
27. A radioactive material P first decays into Q and then Q decays to non-radioactive material R . Which of the following figure represents time dependent mass of P , Q and R ?

- (A) 
- (B) 
- (C) 
- (D) 

28. Displacement of a wave is expressed as $x(t) = 5 \cos(628t + \frac{\pi}{2})$ m. The wavelength of the wave when its velocity is 300 m/s is :
($\pi = 3.14$)

- (A) 0.33 m (B) 0.5 m
(C) 3 m (D) 5 m

29. A metallic ring is uniformly charged as shown in figure. AC and BD are two mutually perpendicular diameters. Electric field due to arc AB at 'O' is 'E' in magnitude. What would be the magnitude of electric field at 'O' due to arc ABC ?



- (A) 2E (B) Zero
(C) E/2 (D) $\sqrt{2}E$

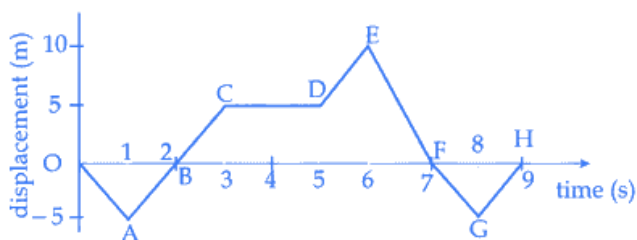
30. A wheel is rolling on a plane surface. The speed of a particle on the highest point of the rim is 8 m/s. The speed of the particle on the rim of the wheel at the same level as the centre of wheel, will be :

- (A) $4\sqrt{2}$ m/s (B) 8 m/s
(C) 4 m/s (D) $8\sqrt{2}$ m/s

31. In an electromagnetic system, a quantity defined as the ratio of electric dipole moment and magnetic dipole moment has dimension of $[M^P L^Q T^R A^S]$. The value of P and Q are :

- (A) -1, 0 (B) 0, -1
(C) -1, 1 (D) 1, -1

32. The displacement x versus time graph is shown below.



- (A) The average velocity during 0 to 3 s is 10 m/s
(B) The average velocity during 3 to 5 s is 0 m/s
(C) The instantaneous velocity at $t = 2$ s is 5 m/s
(D) The average velocity during 5 to 7 s and instantaneous velocity at $t = 6.5$ s are equal
(E) The average velocity from $t = 0$ to $t = 9$ s is zero

Choose the correct answer from the options given below :

- (A) (B), (C), (D) only (B) (B), (D), (E) only
(C) (A), (D), (E) only (D) (B), (C), (E) only

33. Consider a rectangular sheet of solid material of length $l = 9$ cm and width $d = 4$ cm. The coefficient of linear expansion is $\alpha = 3.1 \times 10^{-5} \text{ K}^{-1}$ at room temperature and one atmospheric pressure. The mass of sheet $m = 0.1$ kg and the specific heat capacity $C_v = 900 \text{ J kg}^{-1} \text{ K}^{-1}$. If the amount of heat supplied to the material is $8.1 \times 10^2 \text{ J}$ then change in area of the rectangular sheet is :

- (A) $2.0 \times 10^{-6} \text{ m}^2$ (B) $6.0 \times 10^{-7} \text{ m}^2$
(C) $3.0 \times 10^{-7} \text{ m}^2$ (D) $4.0 \times 10^{-7} \text{ m}^2$

34. An object is kept at rest at a distance of $3R$ above the earth's surface where R is earth's radius. The minimum speed with which it must be projected so that it does not return to earth is : (Assume $M =$ mass of earth, $G =$ Universal gravitational constant)

- (A) $\sqrt{\frac{3GM}{R}}$ (B) $\sqrt{\frac{2GM}{R}}$
(C) $\sqrt{\frac{GM}{2R}}$ (D) $\sqrt{\frac{GM}{R}}$

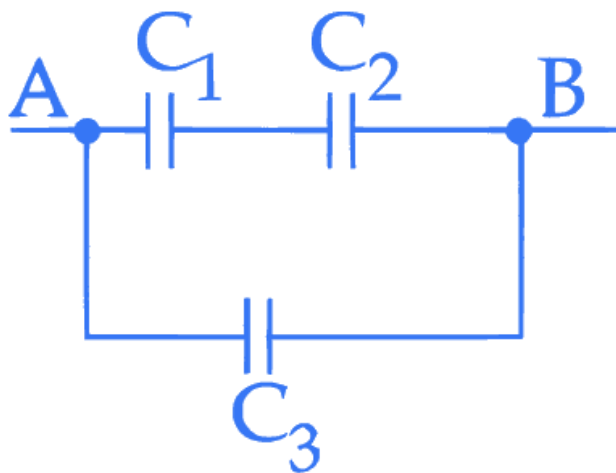
35. A cylindrical rod of length 1 m and radius 4 cm is mounted vertically. It is subjected to a shear force of 10^5 N at the top. Considering infinitesimally small displacement in the upper edge, the angular displacement θ of the rod axis from its original position would be : (shear moduli, $G = 10^{10} \text{ N/m}^2$)

- (A) $1/160\pi$ (B) $1/2\pi$
(C) $1/4\pi$ (D) $1/40\pi$

36. For the determination of refractive index of glass slab, a travelling microscope is used whose main scale contains 300 equal divisions equals to 15 cm. The vernier scale attached to the microscope has 25 divisions equals to 24 divisions of main scale. The least count (LC) of the travelling microscope is (in cm) :

- (A) 0.002 (B) 0.0025
(C) 0.0005 (D) 0.001

37. Three parallel plate capacitors C_1 , C_2 and C_3 each of capacitance $5\mu\text{F}$ are connected as shown in figure. The effective capacitance between points A and B, when the space between the parallel plates of C_1 capacitor is filled with a dielectric medium having dielectric constant of 4, is :



- (A) $9\mu\text{F}$ (B) $30\mu\text{F}$
(C) $7.5\mu\text{F}$ (D) $22.5\mu\text{F}$

38. A block of mass 25 kg is pulled along a horizontal surface by a force at an angle 45° with the horizontal. The friction coefficient between the block and the surface is 0.25. The block travels at a uniform velocity. The work done by the applied force during a displacement of 5 m of the block is :

- (A) 490 J (B) 970 J
(C) 735 J (D) 245 J

39. There are two vessels filled with an ideal gas where volume of one is double the volume of other. The large vessel contains the gas at 8 kPa at 1000 K while the smaller vessel contains the gas at 7 kPa at 500 K. If the vessels are connected to each other by a thin tube allowing the gas to flow and the temperature of both vessels is maintained at 600 K, at steady state the pressure in the vessels will be (in kPa).

- (A) 24 (B) 4.4
(C) 18 (D) 6

40. Two polarisers P_1 and P_2 are placed in such a way that the intensity of the transmitted light will be zero. A third polariser P_3 is inserted in between P_1 and P_2 , at particular angle between P_2 and P_3 . The transmitted intensity of the light passing through all three polarisers is maximum. The angle between the polarisers P_2 and P_3 is :

- (A) $\pi/6$ (B) $\frac{\pi}{3}$
(C) $\frac{\pi}{4}$ (D) $\pi/8$

41. Match List - I with List - II.

	List - I		List - II
(A)	Isobaric	(I)	$\Delta Q = \Delta W$
(B)	Isochoric	(II)	$\Delta Q = \Delta U$
(C)	Adiabatic	(III)	$\Delta Q = \text{zero}$
(D)	Isothermal	(IV)	$\Delta Q = \Delta U + P\Delta V$

ΔQ = Heat supplied

ΔW = Work done by the system

ΔU = Change in internal energy

P = Pressure of the system

ΔV = Change in volume of the system

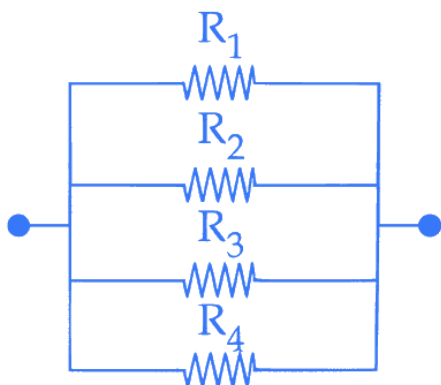
Choose the correct answer from the options given below :

- (A) (A)-(IV), (B)-(I), (C)-(III), (D)-(II) (B) (A)-(IV), (B)-(II), (C)-(III), (D)-(I)
(C) (A)-(IV), (B)-(III), (C)-(II), (D)-(I) (D) (A)-(II), (B)-(IV), (C)-(III), (D)-(I)

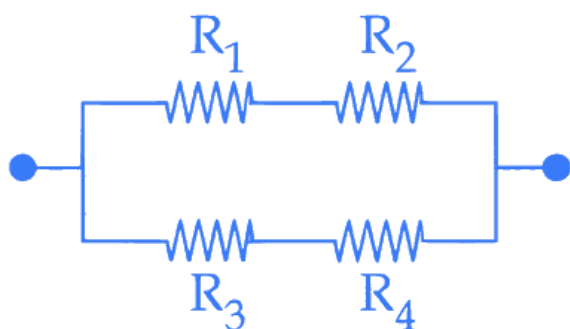
42. There are 'n' number of identical electric bulbs, each is designed to draw a power p independently from the mains supply. They are now joined in series across the mains supply. The total power drawn by the combination is :

- (A) np (B) p
(C) $\frac{p}{n^2}$ (D) $\frac{p}{n}$

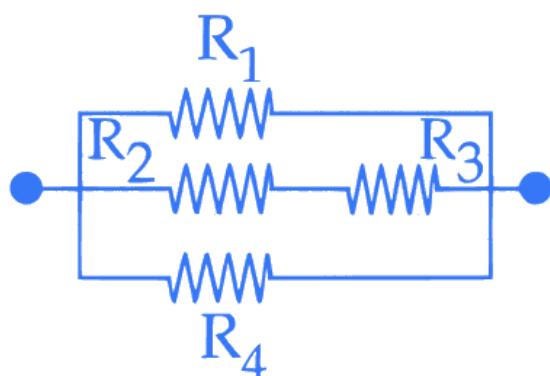
43. From the combination of resistors with resistances values $R_1 = R_2 = R_3 = 5\Omega$ and $R_4 = 10\Omega$, which of the following combination is the best circuit to get an equivalent resistance of 6Ω ?



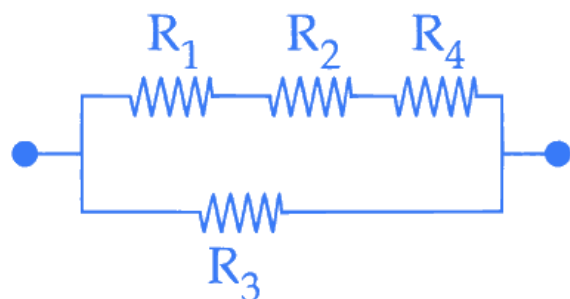
(A)



(B)



(C)



(D)

44. Consider a n-type semiconductor in which n_e and n_h are number of electrons and holes, respectively.

(A) Holes are minority carriers

(B) The dopant is a pentavalent atom

(C) $n_e n_h \neq n_i^2$

(where n_i is number of electrons or holes in semiconductor when it is intrinsic form)

(D) $n_e n_h \geq n_i^2$

(E) The holes are not generated due to the donors
Choose the correct answer from the options given below :

(A) (A), (B), (E) only

(B) (A), (C), (E) only

(C) (A), (B), (C) only

(D) (A), (C), (D) only

45. Given below are two statements :

Statement (I) : The dimensions of Planck's constant and angular momentum are same.

Statement (II) : In Bohr's model electron revolve around the nucleus only in those orbits for which angular momentum is integral multiple of Planck's constant.

In the light of the above statements, choose the most appropriate answer from the options given below :

(A) Both Statement I and Statement II are correct

(B) Statement I is correct but Statement II is incorrect

(C) Both Statement I and Statement II are incorrect

(D) Statement I is incorrect but Statement II is correct

46. A solid sphere with uniform density and radius R is rotating initially with constant angular velocity (ω_1) about its diameter. After some time during the rotation it starts losing mass at a uniform rate, with no change in its shape. The angular velocity of the sphere when its radius becomes $R/2$ is $x\omega_1$. The value of x is _____.

47. An inductor of self inductance 1 H is connected in series with a resistor of 100π ohm and an ac supply of 100π volt, 50 Hz. Maximum current flowing in the circuit is _____ A.

48. If an optical medium possesses a relative permeability of $\frac{10}{\pi}$ and relative permittivity of $\frac{1}{0.0885}$, then the velocity of light is greater in vacuum than that in this medium by _____ times.

$$\begin{aligned} (\mu_0 &= 4\pi \times 10^{-7} \text{ H/m}, \epsilon_0 = 8.85 \times 10^{-12} \text{ F/m}, c \\ &= 3 \times 10^8 \text{ m/s}) \end{aligned}$$

49. A particle of charge $1.6\mu\text{C}$ and mass $16\mu\text{g}$ is present in a strong magnetic field of 6.28 T. The particle is then fired perpendicular to magnetic field. The time required for the particle to return to original location for the first time is _____ s. ($\pi = 3.14$)

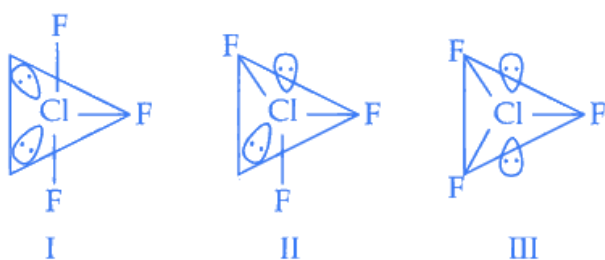
50. In a Young's double slit experiment, two slits are located 1.5 mm apart. The distance of screen from slits is 2 m and the wavelength of the source is 400 nm. If the 20 maxima of the double slit pattern are contained within the central maximum of the single slit diffraction pattern, then the width of each slit is $x \times 10^{-3}$ cm, where x -value is _____.

51. Given below are two statements:

Statement (I): For



, all three possible structures may be drawn as follows.

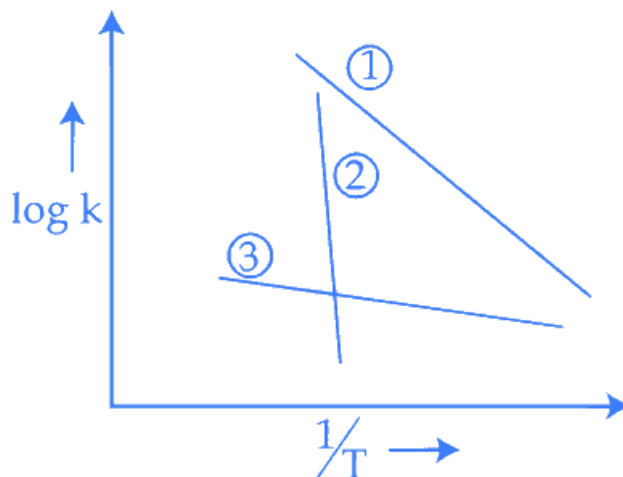


Statement (II): Structure III is most stable, as the orbitals having the lone pairs are axial, where the lp – bp repulsion is minimum.

In the light of the above statements, choose the **most appropriate** answer from the options given below:

- | | |
|--|--|
| (A) Both Statement I and Statement II are correct | (B) Both Statement I and Statement II are incorrect |
| (C) Statement I is correct but Statement II is incorrect | (D) Statement I is incorrect but Statement II is correct |

52. Consider the following plots of log of rate constant k ($\log k$) vs $\frac{1}{T}$ for three different reactions. The correct order of activation energies of these reactions is :



- | | |
|--------------------------------|--------------------------------|
| (A) $E_{a2} > E_{a1} > E_{a3}$ | (B) $E_{a1} > E_{a3} > E_{a2}$ |
| (C) $E_{a3} > E_{a2} > E_{a1}$ | (D) $E_{a1} > E_{a2} > E_{a3}$ |

53. Match List - I with List - II

	List - I (Separation of)		List - II (Separation Technique)
(A)	Aniline from aniline-water mixture	(I)	Simple distillation
(B)	Glycerol from spent-lye in soap industry	(II)	Fractional distillation
(C)	Different fractions of crude oil in petroleum industry	(III)	Distillation at reduced pressure
(D)	Chloroform-Aniline mixture	(IV)	Steam distillation

Choose the correct answer from the options given below :

- | | |
|--|--|
| (A) (A)-(IV), (B)-(III), (C)-(II), (D)-(I) | (B) (A)-(I), (B)-(II), (C)-(III), (D)-(IV) |
| (C) (A)-(III), (B)-(IV), (C)-(I), (D)-(II) | (D) (A)-(II), (B)-(I), (C)-(IV), (D)-(III) |

54. Given below are two statements :

Statement (I) : The first ionisation enthalpy of group 14 elements is higher than the corresponding elements of group 13.

Statement (II) : Melting points and boiling points of group 13 elements are in general much higher than those of corresponding elements of group 14.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|--|---|
| (A) Statement I is correct but Statement II is incorrect | (B) Both Statement I and Statement II are incorrect |
| (C) Statement I is incorrect but Statement II is correct | (D) Both Statement I and Statement II are correct |

55. Which among the following compounds give yellow solid when reacted with NaOI/NaOH ?

- (A) $\text{CH}_3-\overset{\text{OH}}{\underset{|}{\text{CH}}}-\text{C}_2\text{H}_5$
- (B) $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{OH}$
- (C) $\text{CH}_3-\overset{\text{O}}{\underset{||}{\text{C}}}-\text{C}_2\text{H}_5$
- (D) $\text{CH}_3-\overset{\text{O}}{\underset{||}{\text{C}}}-\text{OH}$
- (E) $\text{CH}_3-\text{CH}_2-\overset{\text{O}}{\underset{||}{\text{C}}}-\text{H}$

Choose the correct answer from the options given below :

- | | |
|---------------------------|---------------------------|
| (A) (A) and (C) Only | (B) (C) and (D) Only |
| (C) (B), (C) and (E) Only | (D) (A), (C) and (D) Only |

56. Consider the ground state of chromium atom ($Z = 24$). How many electrons are with Azimuthal quantum number $l = 1$ and $l = 2$ respectively ?

- | | |
|--------------|--------------|
| (A) 16 and 5 | (B) 12 and 5 |
| (C) 12 and 4 | (D) 16 and 4 |

57. Given below are two statements:

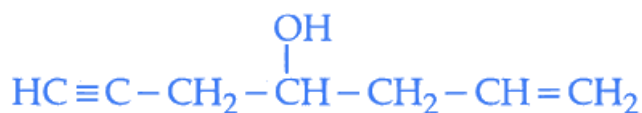
Statement (I) : Molal depression constant K_f is given by $\frac{M_1 RT_f}{\Delta S_{\text{fus}}}$, where symbols have their usual meaning.

Statement (II) : K_f for benzene is less than the K_f for water.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|--|---|
| (A) Statement I is correct but Statement II is incorrect | (B) Both Statement I and Statement II are incorrect |
| (C) Statement I is incorrect but Statement II is correct | (D) Both Statement I and Statement II are correct |

58. The IUPAC name of the following compound is:



- | | |
|------------------------------|------------------------------|
| (A) 4-Hydroxyhept-6-en-1-yne | (B) Hept-1-en-6-yn-4-ol |
| (C) Hept-6-en-1-yn-4-ol | (D) 4-Hydroxyhept-1-en-6-yne |

59. Half life of zero order reaction $A \rightarrow \text{product}$ is 1 hour, when initial concentration of reactant is 2.0 mol L^{-1} . The time required to decrease concentration of A from 0.50 to 0.25 mol L^{-1} is :

- (A) 0.5 hour (B) 15 min
(C) 60 min (D) 4 hour

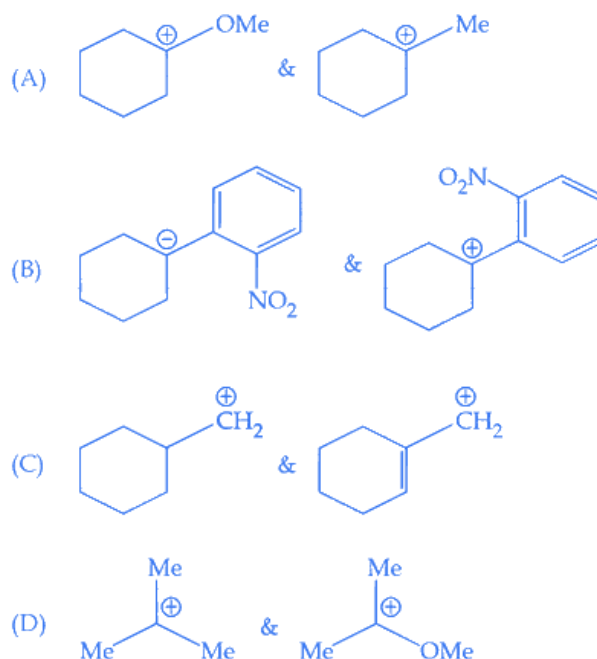
60. The incorrect relationship in the following pairs in relation to ionisation enthalpies is :

- (A) $\text{Mn}^{2+} < \text{Fe}^{2+}$ (B) $\text{Mn}^+ < \text{Mn}^{2+}$
(C) $\text{Mn}^+ < \text{Cr}^+$ (D) $\text{Fe}^{2+} < \text{Fe}^{3+}$

61. A toxic compound "A" when reacted with NaCN in aqueous acidic medium yields an edible cooking component and food preservative "B". "B" is converted to "C" by diborane and can be used as an additive to petrol to reduce emission. "C" upon reaction with oleum at 140°C yields an inhalable anesthetic "D". Identify "A", "B", "C" & "D", respectively :

- | | |
|---|---|
| (A) Methanol; acetic acid; ethanol; diethyl ether | (B) Ethanol; acetonitrile; ethylamine; ethylene |
| (C) Methanol; formaldehyde; methyl chloride; chloroform | (D) Acetaldehyde; 2-hydroxypropanoic acid; propanoic acid; dipropyl ether |

62. In which pairs, the first ion is more stable than the second?

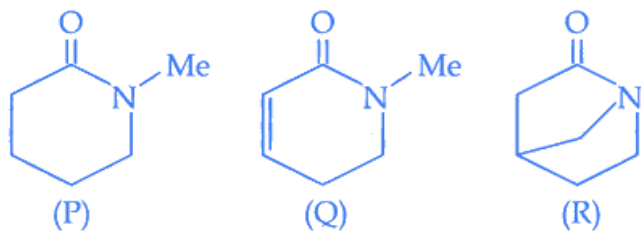


- (A) (A) & (B) only (B) (B) & (D) only
(C) (A) & (C) only (D) (B) & (C) only

63. 'X' is the number of electrons in t_{2g} orbitals of the most stable complex ion among $[\text{Fe}(\text{NH}_3)_6]^{3+}$, $[\text{FeCl}_6]^{3-}$, $[\text{Fe}(\text{C}_2\text{O}_4)_3]^{3-}$ and $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$. The nature of oxide of vanadium of the type V_2O_X is :

- (A) Amphoteric (B) Acidic
(C) Basic (D) Neutral

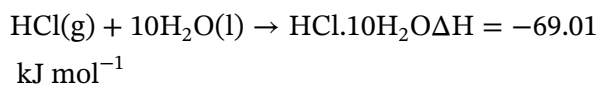
64. The correct order of basicity for the following molecules is :



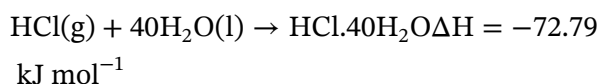
- (A) $R > P > Q$ (B) $P > Q > R$
 (C) $R > Q > P$ (D) $Q > P > R$

65. Consider the given data :

(a)



(b)

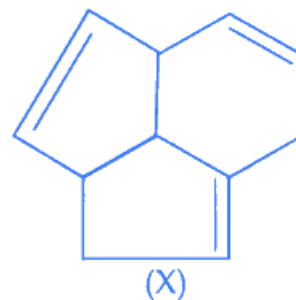


Choose the correct statement :

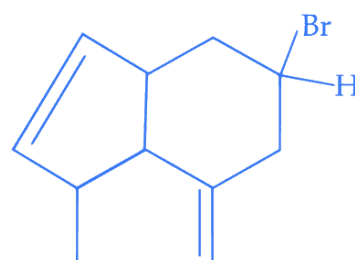
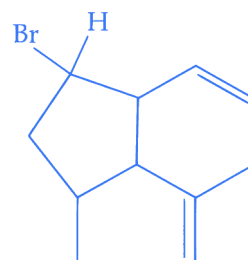
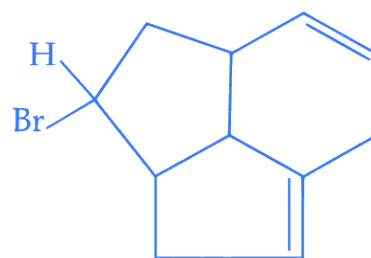
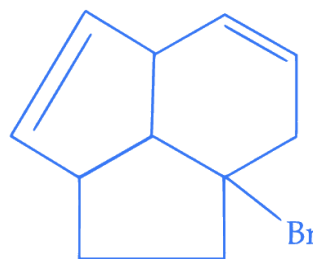
- (A) The heat of dilution for the $\text{HCl}(\text{HCl.10H}_2\text{O}$ to $\text{HCl.40H}_2\text{O})$ is 3.78 kJ mol^{-1} .
 (B) Dissolution of gas in water is an endothermic process.
 (C) The heat of solution depends on the amount of solvent.
 (D) The heat of formation of HCl solution is represented by both (a) and (b).

66. Consider the following molecule(X).

The structure of X is



The major product formed when the given molecule (X) is treated with $\text{HBr}(\text{1eq})$ is :



67. The correct order of $[\text{FeF}_6]^{3-}$, $[\text{CoF}_6]^{3-}$, $[\text{Ni}(\text{CO})_4]$ and $[\text{Ni}(\text{CN})_4]^{2-}$ complex species based on the number of unpaired electrons present is:

- (A) $[\text{CoF}_6]^{3-} > [\text{FeF}_6]^{3-} > [\text{Ni}(\text{CO})_4] > [\text{Ni}(\text{CN})_4]^{2-}$
- (B) $[\text{FeF}_6]^{3-} > [\text{CoF}_6]^{3-} > [\text{Ni}(\text{CN})_4]^{2-} > [\text{Ni}(\text{CO})_4]$
- (C) $[\text{FeF}_6]^{3-} > [\text{CoF}_6]^{3-} > [\text{Ni}(\text{CN})_4]^{2-} = [\text{Ni}(\text{CO})_4]$
- (D) $[\text{Ni}(\text{CN})_4]^{2-} > [\text{FeF}_6]^{3-} > [\text{CoF}_6]^{3-} > [\text{Ni}(\text{CO})_4]$

68. Given below are two statements :

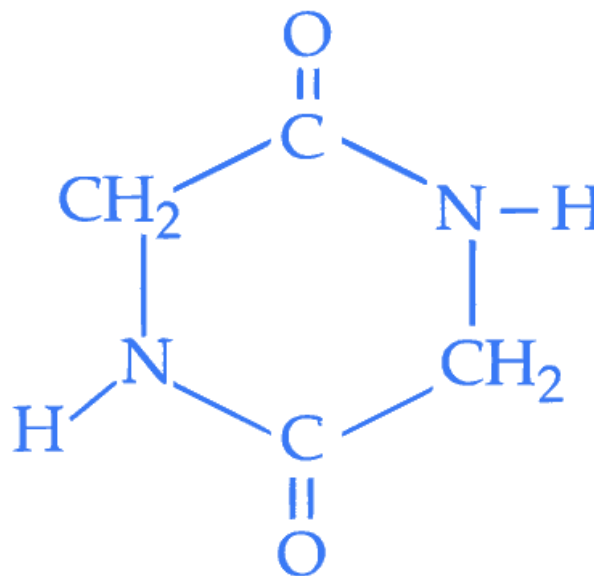
Statement (I) : Alcohols are formed when alkyl chlorides are treated with aqueous potassium hydroxide by elimination reaction.

Statement (II) : In alcoholic potassium hydroxide, alkyl chlorides form alkenes by abstracting the hydrogen from the β -carbon.

In the light of the above statements, choose the most appropriate answer from the options given below :

- | | |
|--|--|
| (A) Both Statement I and Statement II are incorrect | (B) Both Statement I and Statement II are correct |
| (C) Statement I is incorrect but Statement II is correct | (D) Statement I is correct but Statement II is incorrect |

69. A dipeptide, " x " on complete hydrolysis gives " y " and " z ". " y " on treatment with aq. HNO_2 produces lactic acid. On the other hand " z " on heating gives the following cyclic molecule.



Based on the information given, the dipeptide X is :

- | | |
|---------------------|---------------------|
| (A) alanine-alanine | (B) valine-glycine |
| (C) valine-leucine | (D) alanine-glycine |

70. The elements of Group 13 with highest and lowest first ionisation enthalpies are respectively :

- | | |
|------------|------------|
| (A) B & Tl | (B) Tl & B |
| (C) B & In | (D) B & Ga |

71. Sea water, which can be considered as a 6 molar (6M) solution of NaCl, has a density of 2 g mL^{-1} . The concentration of dissolved oxygen (O_2) in sea water is 5.8 ppm. Then the concentration of dissolved oxygen (O_2) in sea water, is $x \times 10^{-4} \text{ m}$.

$x =$ _____. (Nearest integer)

Given: Molar mass of NaCl is 58.5 g mol^{-1}

Molar mass of O_2 is 32 g mol^{-1}

72. The amount of calcium oxide produced on heating 150 kg limestone (75% pure) is _____ kg. (Nearest integer)

Given: Molar mass (in gmol^{-1}) of Ca-40, O-16, C-12

73. A metal complex with a formula $\text{MCl}_4 \cdot 3\text{NH}_3$ is involved in $\text{sp}^3 \text{d}^2$ hybridisation. It upon reaction with excess of AgNO_3 solution gives ' x ' moles of AgCl . Consider ' x ' is equal to the number of lone pairs of electron present in central atom of BrF_5 . Then the number of geometrical isomers exhibited by the complex is _____.

74. The molar conductance of an infinitely dilute solution of ammonium chloride was found to be $185 \text{ S cm}^2 \text{ mol}^{-1}$ and the ionic conductance of hydroxyl and chloride ions are 170 and $70 \text{ S cm}^2 \text{ mol}^{-1}$, respectively. If molar conductance of 0.02 M solution of ammonium hydroxide is $85.5 \text{ S cm}^2 \text{ mol}^{-1}$, its degree of dissociation is given by $x \times 10^{-1}$. The value of x is _____. (Nearest integer)

75. $x \text{ mg}$ of $\text{Mg}(\text{OH})_2$ (molar mass = 58) is required to be dissolved in 1.0 L of water to produce a pH of 10.0 at 298 K. The value of x is _____ mg. (Nearest integer)

(Given : $\text{Mg}(\text{OH})_2$ is assumed to dissociate completely in H_2O]

- End of Question Paper -