

JEE Main 2026 Paper 1 (B.E./B.Tech) 22 January 2026 Shift 2 Question Paper



Conducted by the National Testing Agency (NTA)

Time Allowed: 3 Hours

Paper 1: B.E./B.Tech.

Maximum Marks: 300

General Instructions

- (i) This paper is **Paper 1 (B.E./B.Tech.)** of **JEE (Main) 2026 - Session 1**, conducted by the National Testing Agency (NTA) in Computer Based Test (CBT) mode on 22 January 2026, Shift 2. The total duration is **3 hours**.
- (ii) The paper has **75 questions** and a maximum of **300 marks**: Mathematics, Physics and Chemistry carry 25 questions (100 marks) each, in that order.
- (iii) Each subject has **two sections**. **Section A** has **20 multiple choice questions** (MCQs) with four options each. Select the most appropriate answer. **Section B** has **5 numerical value questions**.
- (iv) **All questions are compulsory**. For Section B, the answer is an integer value; the answer should be **rounded off to the nearest integer**. Use only the constants given in the question while calculating.
- (v) **Marking scheme** (Sections A and B): **+4** for a correct answer, **-1** for an incorrect answer, and **0** for an unanswered or marked-for-review question.
- (vi) All calculations and rough work are to be done on the rough sheets only. Calculators, log tables, slide rules and electronic gadgets are not allowed.

1. Let $f(x) = [x]^2 - [x+3] - 3, x \in \mathbf{R}$, where $[.]$ is the greatest integer function. Then

- (A) $f(x) = 0$ for finitely many values of x (B) $f(x) < 0$ only for $x \in [-1, 3]$
 (C) $\int_0^2 f(x) dx = -6$ (D) $f(x) > 0$ only for $x \in [4, \infty)$

2. Let the locus of the mid-point of the chord through the origin O of the parabola $y^2 = 4x$ be the curve S . Let P be any point on S . Then the locus of the point, which internally divides OP in the ratio $3 : 1$, is :

- (A) $2x^2 = 3y$ (B) $2y^2 = 3x$
 (C) $3y^2 = 2x$ (D) $3x^2 = 2y$

3. Let n be the number obtained on rolling a fair die. If the probability that the system

$$\begin{aligned} x - ny + z &= 6 \\ x + (n-2)y + (n+1)z &= 8 \\ (n-1)y + z &= 1 \end{aligned}$$

has a unique solution is $\frac{k}{6}$, then the sum of k and all possible values of n is :

- (A) 22 (B) 20
 (C) 24 (D) 21

4. Let the domain of the function

$$f(x) = \log_3 \log_5 (7 - \log_2 (x^2 - 10x + 85)) + \sin^{-1} \left(\left| \frac{3x-7}{17-x} \right| \right)$$

be $(\alpha, \beta]$. Then $\alpha + \beta$ is equal to :

- (A) 12 (B) 8
 (C) 10 (D) 9

5. Let S and S' be the foci of the ellipse $\frac{x^2}{25} + \frac{y^2}{9} = 1$ and $P(\alpha, \beta)$ be a point on the ellipse in the first quadrant. If $(SP)^2 + (S'P)^2 - SP \cdot S'P = 37$, then $\alpha^2 + \beta^2$ is equal to :

- (A) 13 (B) 15
 (C) 11 (D) 17

6. If $\lim_{x \rightarrow 0} \frac{e^{(a-1)x} + 2 \cos bx + (c-2)e^{-x}}{x \cos x - \log_e(1+x)} = 2$, then $a^2 + b^2 + c^2$ is equal to :

- (A) 3 (B) 5
 (C) 9 (D) 7

7. Let $\vec{a} = 2\hat{i} - \hat{j} + \hat{k}$ and $\vec{b} = \lambda\hat{j} + 2\hat{k}, \lambda \in \mathbf{Z}$ be two vectors. Let $\vec{c} = \vec{a} \times \vec{b}$ and $\vec{d}$ be a vector of magnitude 2 in yz -plane. If $|\vec{c}| = \sqrt{53}$, then the maximum possible value of $(\vec{c} \cdot \vec{d})^2$ is equal to :

- (A) 52 (B) 26
 (C) 104 (D) 208

8. Let $[.]$ denote the greatest integer function, and let $f(x) = \min \{ \sqrt{2x}, x^2 \}$.

Let $S = \{x \in (-2, 2) : \text{the function } g(x) = |x| [x^2] \text{ is discontinuous at } x\}$.

Then $\sum_{x \in S} f(x)$ equals

- (A) $2 - \sqrt{2}$ (B) $2\sqrt{6} - 3\sqrt{2}$
 (C) $1 - \sqrt{2}$ (D) $\sqrt{6} - 2\sqrt{2}$

9. Among the statements

(S1) : If $A(5, -1)$ and $B(-2, 3)$ are two vertices of a triangle, whose orthocentre is $(0, 0)$, then its third vertex is $(-4, -7)$

and

(S2) : If positive numbers $2a, b, c$ are three consecutive terms of an A.P., then the lines $ax + by + c = 0$ are concurrent at $(2, -2)$,

- (A) both are incorrect (B) only (S2) is correct
 (C) both are correct (D) only (S1) is correct

10. Let α, β be the roots of the quadratic equation $12x^2 - 20x + 3\lambda = 0, \lambda \in \mathbf{Z}$. If $\frac{1}{2} \leq |\beta - \alpha| \leq \frac{3}{2}$, then the sum of all possible values of λ is :

- (A) 3 (B) 6
 (C) 1 (D) 4

11. The number of elements in the relation $R = \{(x, y) : 4x^2 + y^2 < 52, x, y \in \mathbf{Z}\}$ is

- (A) 86 (B) 67
(C) 89 (D) 77

12. Let f and g be functions satisfying $f(x+y) = f(x)f(y)$, $f(1) = 7$ and $g(x+y) = g(xy)$, $g(1) = 1$, for all $x, y \in \mathbf{N}$. If $\sum_{x=1}^n \left(\frac{f(x)}{g(x)}\right) = 19607$, then n is equal to :

- (A) 6 (B) 7
(C) 4 (D) 5

13. If $X = \begin{bmatrix} x \\ y \\ z \end{bmatrix}$ is a solution of the system of equations $AX = B$, where

$$\text{adj } A = \begin{bmatrix} 4 & 2 & 2 \\ -5 & 0 & 5 \\ 1 & -2 & 3 \end{bmatrix} \text{ and } B = \begin{bmatrix} 4 \\ 0 \\ 2 \end{bmatrix}, \text{ then}$$

$|x+y+z|$ is equal to :

- (A) 3 (B) 2
(C) $\frac{3}{2}$ (D) 1

14. The area of the region $A = \{(x, y) : 4x^2 + y^2 \leq 8 \text{ and } y^2 \leq 4x\}$ is:

- (A) $\pi + \frac{2}{3}$ (B) $\frac{\pi}{2} + 2$
(C) $\pi + 4$ (D) $\frac{\pi}{2} + \frac{1}{3}$

15. Let L be the line $\frac{x+1}{2} = \frac{y+1}{3} = \frac{z+3}{6}$ and let S be the set of all points (a, b, c) on L , whose distance from the line $\frac{x+1}{2} = \frac{y+1}{3} = \frac{z-9}{0}$ along the line L is 7. Then $\sum_{(a,b,c) \in S} (a+b+c)$ is equal to :

- (A) 28 (B) 6
(C) 40 (D) 34

16. If $y = y(x)$ satisfies the differential equation

$$16(\sqrt{x+9}\sqrt{x})(4+\sqrt{9+\sqrt{x}})\cos y \, dy =$$

$$y = (1+2\sin y)dx, x > 0$$

and $y(256) = \frac{\pi}{2}$, $y(49) = \alpha$, then $2\sin \alpha$ is equal to :

- (A) $2\sqrt{2} - 1$ (B) $\sqrt{2} - 1$
(C) $2(\sqrt{2} - 1)$ (D) $3(\sqrt{2} - 1)$

17. Let $P(10, 2\sqrt{15})$ be a point on the hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$, whose foci are S and S' . If the length of its latus rectum is 8, then the square of the area of $\Delta PSS'$ is equal to :

- (A) 4200 (B) 1462
(C) 900 (D) 2700

18. If the mean deviation about the median of the numbers $k, 2k, 3k, \dots, 1000k$ is 500, then k^2 is equal to :

- (A) 1 (B) 16
(C) 9 (D) 4

19. Let $S = \{z \in \mathbb{C} : 4z^2 + \bar{z} = 0\}$. Then $\sum_{z \in S} |z|^2$ is equal to:

- (A) $\frac{5}{64}$ (B) $\frac{1}{16}$
(C) $\frac{7}{64}$ (D) $\frac{3}{16}$

20. Let C_r denote the coefficient of x^r in the binomial expansion of $(1+x)^n$, $n \in \mathbf{N}$, $0 \leq r \leq n$. If $P_n = C_0 - C_1 + \frac{2^2}{3}C_2 - \frac{2^3}{4}C_3 + \dots + \frac{(-2)^n}{n+1}C_n$, then the value of $\sum_{n=1}^{25} \frac{1}{P_{2n}}$ equals.

- (A) 675 (B) 580
(C) 525 (D) 650

21. Suppose a, b, c are in A.P. and $a^2, 2b^2, c^2$ are in G.P. If $a < b < c$ and $a + b + c = 1$, then $9(a^2 + b^2 + c^2)$ is equal to ____.

22. Let a vector $\vec{a} = \sqrt{2}\hat{i} - \hat{j} + \lambda\hat{k}, \lambda > 0$, make an obtuse angle with the vector $\vec{b} = -\lambda^2\hat{i} + 4\sqrt{2}\hat{j} + 4\sqrt{2}\hat{k}$ and an angle $\theta, \frac{\pi}{6} < \theta < \frac{\pi}{2}$, with the positive z -axis. If the set of all possible values of λ is $(\alpha, \beta) - \{\gamma\}$, then $\alpha + \beta + \gamma$ is equal to ____.

23. Let $[.]$ be the greatest integer function. If $\alpha = \int_0^{64} (x^{1/3} - [x^{1/3}]) dx$, then $\frac{1}{\pi} \int_0^{\alpha\pi} \left(\frac{\sin^2 \theta}{\sin^6 \theta + \cos^6 \theta} \right) d\theta$ is equal to ____.

24. Let $\cos(\alpha + \beta) = -\frac{1}{10}$ and $\sin(\alpha - \beta) = \frac{3}{8}$, where $0 < \alpha < \frac{\pi}{3}$ and $0 < \beta < \frac{\pi}{4}$. If $\tan 2\alpha = \frac{3(1-r\sqrt{5})}{\sqrt{11}(s+\sqrt{5})}, r, s \in N$, then $r + s$ is equal to ____.

25. Let S be the set of the first 11 natural numbers. Then the number of elements in $A = \{B \subseteq S : n(B) \geq 2 \text{ and the product of all elements of } B \text{ is even}\}$ is ____.

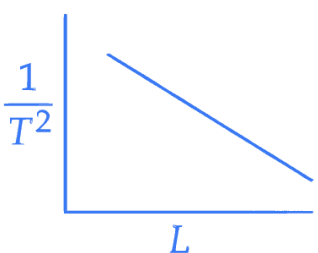
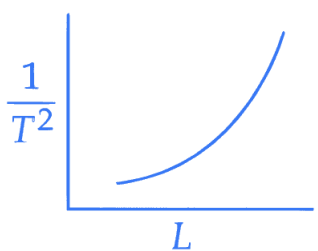
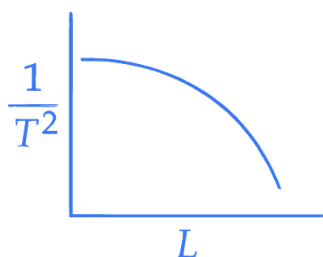
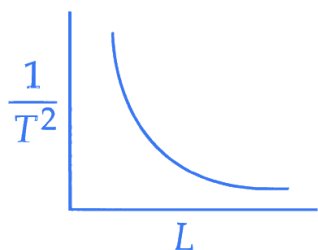
Physics - Section A · Multiple Choice Questions

26. In an open organ pipe ν_3 and ν_6 are 3rd and 6th harmonic frequencies, respectively. If $\nu_6 - \nu_3 = 2200$ Hz then length of the pipe is ____ mm .

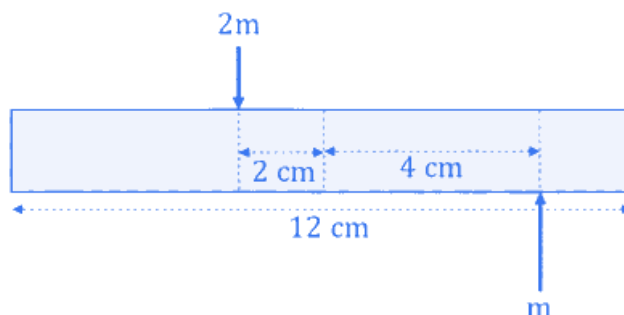
(Take velocity of sound in air is 330 m/s.)

- (A) 200 (B) 225
(C) 275 (D) 250

27. Using a simple pendulum experiment g is determined by measuring its time period T . Which of the following plots represent the correct relation between the pendulum length L and time period T ?



28. A uniform bar of length 12 cm and mass 20 m lies on a smooth horizontal table. Two point masses m and $2m$ are moving in opposite directions with same speed of v and in the same plane as the bar, as shown in figure. These masses strike the bar simultaneously and get stuck to it. After collision the entire system is rotating with angular frequency ω . The ratio of v and ω is :



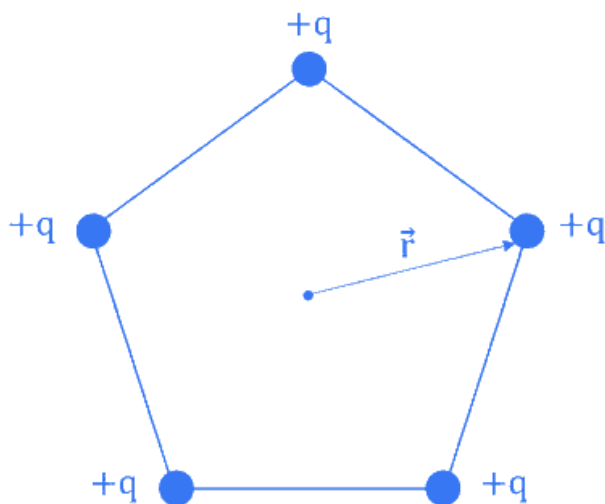
- (A) 33 (B) $2\sqrt{88}$
(C) 32 (D) 66

29. A laser beam has intensity of 4.0×10^{14} W/m². The amplitude of magnetic field associated with beam is ____ T.

(Take $\epsilon_0 = 8.85 \times 10^{-12}$ C²/Nm² and $c = 3 \times 10^8$ m/s)

- (A) 1.83 (B) 2.0
(C) 5.5 (D) 18.3

30. Five positive charges each having charge q are placed at the vertices of a pentagon as shown in the figure. The electric potential (V) and the electric field ($\vec{E}$) at the center O of the pentagon due to these five positive charges are :



- (A) $V = \frac{5q}{4\pi\epsilon_0 r}$ and $\vec{E} = \frac{5\sqrt{3}q}{8\pi\epsilon_0 r^2} \hat{r}$
- (B) $V = \frac{5q}{4\pi\epsilon_0 r}$ and $\vec{E} = 0$
- (C) $V = 0$ and $\vec{E} = 0$
- (D) $V = \frac{5q}{4\pi\epsilon_0 r}$ and $\vec{E} = \frac{5q}{4\pi\epsilon_0 r^2} \hat{r}$

31. Given below are two statements :

Statement I : An object moves from position r_1 to position r_2 under a conservative force field $\vec{F}$. The work done by the force is $W = - \int_{r_1}^{r_2} \vec{F} \cdot d\vec{r}$.

Statement II : Any object moving from one location to another location can follow infinite number of paths. Therefore, the amount of work done by the object changes with the path it follows for a conservative force.

In the light of the above statements, choose the correct answer from the options given below :

- (A) Statement I is true but Statement II is false
- (B) Statement I is false but Statement II is true
- (C) Both Statement I and Statement II are false
- (D) Both Statement I and Statement II are true

32. When a part of a straight capillary tube is placed vertically in a liquid, the liquid raises upto certain height h . If the inner radius of the capillary tube, density of the liquid and surface tension of the liquid decrease by 1% each, then the height of the liquid in the tube will change by ____ %.

- (A) +3 (B) +1
(C) -1 (D) -3

33. Given below are two statements :

Statement I : A satellite is moving around earth in the orbit very close to the earth surface. The time period of revolution of satellite depends upon the density of earth.

Statement II : The time period of revolution of the satellite is $T = 2\pi\sqrt{\frac{R_e}{g}}$ (for satellite very close to the earth surface), where R_e radius of earth and g acceleration due to gravity. In the light of the above statements, choose the correct answer from the options given below :

- (A) Statement I is true but Statement II is false
- (B) Statement I is false but Statement II is true
- (C) Both Statement I and Statement II are true
- (D) Both Statement I and Statement II are false

34. Consider two boxes containing ideal gases A and B such that their temperatures, pressures and number densities are same. The molecular size of A is half of that of B and mass of molecule A is four times that of B . If the collision frequency in gas B is $32 \times 10^{18}/s$ then collision frequency in gas A is ____ /s.

- (A) 8×10^{18} (B) 2×10^{18}
(C) 32×10^{18} (D) 4×10^{18}

35. In parallax method for the determination of focal length of a concave mirror, the object should always be placed:

- | | |
|--|---|
| (A) between the focus (F) and the centre of curvature (C) of the mirror ONLY | (B) between the pole (P) and the focus (F) of the concave mirror ONLY |
| (C) at any point beyond the focus (F) of the mirror | (D) beyond the centre of the curvature (C) of the mirror ONLY |

36. The wavelength of light, while it is passing through water is 540 nm. The refractive index of water is $4/3$. The wavelength of the same light when it is passing through a transparent medium having refractive index of $3/2$ is ____ nm.

- | | |
|---------|---------|
| (A) 540 | (B) 840 |
| (C) 480 | (D) 380 |

37. The smallest wavelength of Lyman series is 91 nm. The difference between the largest wavelengths of Paschen and Balmer series is nearly ____ nm.

- | | |
|----------|----------|
| (A) 1784 | (B) 1875 |
| (C) 1217 | (D) 1550 |

38. An electric power line having total resistance of 2Ω , delivers 1 kW of power at 250 V. The percentage efficiency of transmission line is ____.

- | | |
|----------|----------|
| (A) 96.9 | (B) 92.5 |
| (C) 86.5 | (D) 100 |

39. Three small identical bubbles of water having same charge on each coalesce to form a bigger bubble. Then the ratio of the potentials on one initial bubble and that on the resultant bigger bubble is :

- | | |
|-------------------|-------------------|
| (A) $1 : 3^{2/3}$ | (B) $3^{2/3} : 1$ |
| (C) $1 : 3^{1/3}$ | (D) $1 : 2^{2/3}$ |

40. Given below are two statements :

Statement I : For a mechanical system of many particles total kinetic energy is the sum of kinetic energies of all the particles.

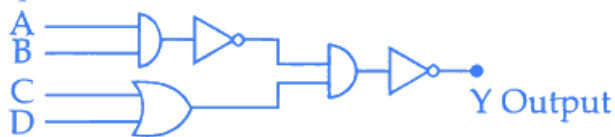
Statement II : The total kinetic energy can be the sum of kinetic energy of the center of mass w.r.t to the origin and the kinetic energy of all the particles w.r.t. the center of mass as the reference.

In the light of the above statements, choose the correct answer from the options given below :

- | | |
|---|---|
| (A) Both Statement I and Statement II are false | (B) Statement I is false but Statement II is true |
| (C) Statement I is true but Statement II is false | (D) Both Statement I and Statement II are true |

41. The correct truth table for the given input data of the following logic gate is :

Inputs



(A)

Inputs				O ut pu t
A	B	C	D	Y
1	1	0	1	1
0	0	1	1	0
1	0	1	0	0
1	1	1	1	1

(B)

Inputs				O ut pu t
A	B	C	D	Y
1	1	0	1	0
0	0	1	1	1
1	0	1	0	1
1	1	1	1	1

(C)

Inputs				O ut pu t
A	B	C	D	Y
1	1	0	1	1
0	0	1	1	0
1	0	1	0	1
1	1	1	1	0

(D)

Inputs				O ut pu t
A	B	C	D	Y
1	1	0	1	0
0	0	1	1	0
1	0	1	0	1
1	1	1	1	1

42. If ϵ , E and t represent the free space permittivity, electric field and time respectively, then the unit of $\frac{\epsilon E}{t}$ will be :

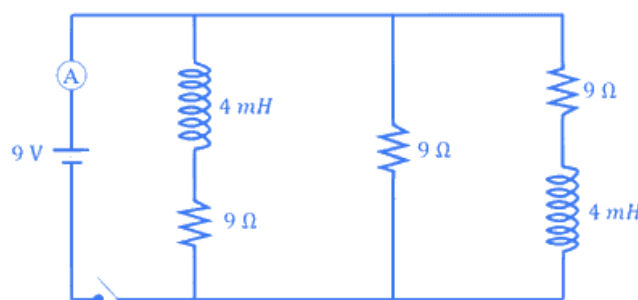
- (A) Am (B) A/m
(C) A/m² (D) Am²

43. Which of the following are true for a single slit diffraction?

- A. Width of central maxima increases with increase in wavelength keeping slit width constant.
B. Width of central maxima increases with decrease in wavelength keeping slit width constant.
C. Width of central maxima increases with decrease in slit width at constant wavelength.
D. Width of central maxima increases with increase in slit width at constant wavelength.
E. Brightness of central maxima increases for decrease in wavelength at constant slit width.

- (A) B, D only (B) A, D only
(C) A, C, E only (D) B, C only

44. Figure shows the circuit that contains three resistances (9Ω each) and two inductors (4 mH each). The reading of ammeter at the moment switch K is turned ON , is ____ A.



- (A) 2 (B) 3
(C) 1 (D) zero

45. Light is incident on a metallic plate having work function $110 \times 10^{-20}\text{ J}$. If the produced photoelectrons have zero kinetic energy then the angular frequency of the incident light is ____ rad/s. ($h = 6.63 \times 10^{-34}\text{ J.s}$).

- (A) 1.04×10^{13} (B) 1.66×10^{16}
(C) 1.66×10^{15} (D) 1.04×10^{16}

46. An insulated cylinder of volume 60 cm^3 is filled with a gas at 27°C and 2 atmospheric pressure. Then the gas is compressed making the final volume as 20 cm^3 while allowing the temperature to rise to 77°C . The final pressure is ____ atmospheric pressure.

47. A conducting circular loop is rotated about its diameter at a constant angular speed of 100 rad/s in a magnetic field of 0.5 T perpendicular to the axis of rotation. When the loop is rotated by 30° from the horizontal position, the induced EMF is 15.4 mV . The radius of the loop is ____ mm.

$$\left(\text{Take } \pi = \frac{22}{7} \right)$$

48. A capacitor P with capacitance $10 \times 10^{-6} \text{ F}$ is fully charged with a potential difference of 6.0 V and disconnected from the battery. The charged capacitor P is connected across another capacitor Q with capacitance $20 \times 10^{-6} \text{ F}$. The charge on capacitor Q when equilibrium is established will be $\alpha \times 10^{-5} \text{ C}$ (assume capacitor Q does not have any charge initially), the value of α is ____.

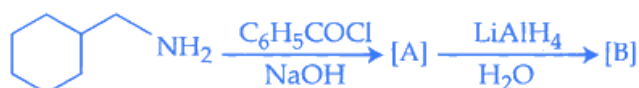
49. A cylindrical conductor of length 2 m and area of cross-section 0.2 mm^2 carries an electric current of 1.6 A when its ends are connected to a 2 V battery. Mobility of electrons in the conductor is $\alpha \times 10^{-3} \text{ m}^2/\text{V.s}$. The value of α is :
(electron concentration = $5 \times 10^{28}/\text{m}^3$ and electron charge = $1.6 \times 10^{-19} \text{ C}$)

50. Two masses m and $2m$ are connected by a light string going over a pulley (disc) of mass 30 m with radius $r = 0.1 \text{ m}$. The pulley is mounted in a vertical plane and it is free to rotate about its axis. The $2m$ mass is released from rest and its speed when it has descended through a height of 3.6 m is ____ m/s . (Assume string does not slip and $g = 10 \text{ m/s}^2$)

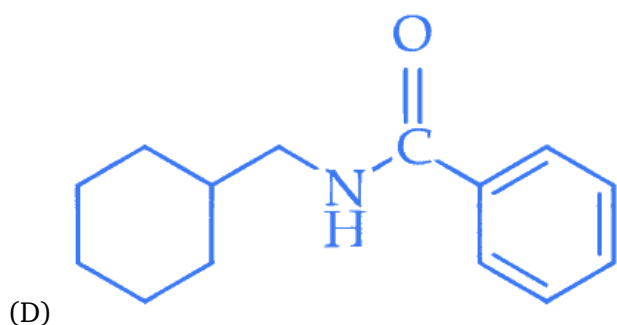
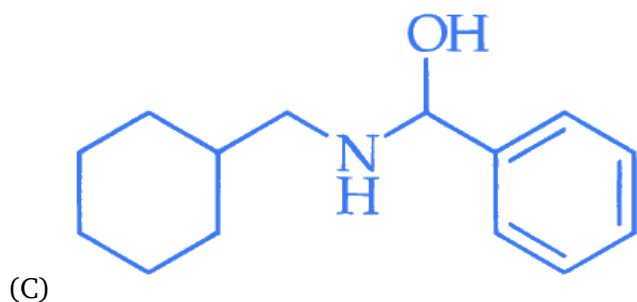
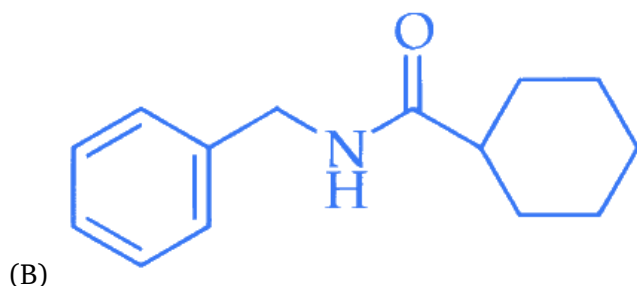
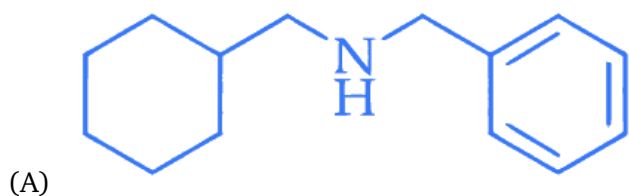
51. Among H_2S , H_2O , NF_3 , NH_3 and CHCl_3 , identify the molecule (X) with lowest dipole moment value. The number of lone pairs of electrons present on the central atom of the molecule (X) is :

- (A) 3 (B) 2
(C) 0 (D) 1

52.



The final product [B] is :



53. Given below are two statements :

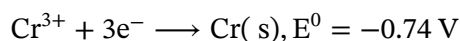
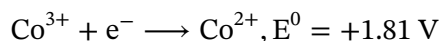
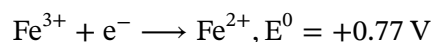
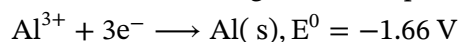
Statement I : $\text{C} < \text{O} < \text{N} < \text{F}$ is the correct order in terms of first ionization enthalpy values.

Statement II : $\text{S} > \text{Se} > \text{Te} > \text{Po} > \text{O}$ is the correct order in terms of the magnitude of electron gain enthalpy values.

In the light of the above statements, choose the correct answer from the options given below :

- (A) Both Statement I and Statement II are false (B) Statement I is false but Statement II is true
(C) Statement I is true but Statement II is false (D) Both Statement I and Statement II are true

54. Consider the following reduction processes :



The tendency to act as reducing agent decreases in the order :

- (A) $\text{Al} > \text{Fe}^{2+} > \text{Cr} > \text{Co}^{2+}$
(B) $\text{Al} > \text{Cr} > \text{Co}^{2+} > \text{Fe}^{2+}$
(C) $\text{Cr} > \text{Fe}^{2+} > \text{Al} > \text{Co}^{2+}$
(D) $\text{Al} > \text{Cr} > \text{Fe}^{2+} > \text{Co}^{2+}$

55. The compound A, $\text{C}_8\text{H}_8\text{O}_2$ reacts with acetophenone to form a single product via cross-Aldol condensation. The compound A on reaction with conc. NaOH forms a substituted benzyl alcohol as one of the two products. The compound A is :

- (A) 2-hydroxy acetophenone (B) 4-hydroxy benzaldehyde
(C) 4-methyl benzoic acid (D) 4-methoxy benzaldehyde

56.

Match List - I with List - II.

List - I Reaction of Glucose with		List - II Product formed	
A.	Hydroxylamine	I.	Gluconic acid
B.	Br ₂ water	II.	Glucose pentacetate
C.	Excess acetic anhydride	III.	Saccharic acid
D.	Concentrated HNO ₃	IV.	Glucosime

Choose the correct answer from the options given below :

- (A) A-IV, B-III, C-II, D-I (B) A-I, B-III, C-IV, D-II
(C) A-III, B-I, C-IV, D-II (D) A-IV, B-I, C-II, D-III

57. Identify the correct statements :

- A. Hydrated salts can be used as primary standard.
B. Primary standard should not undergo any reaction with air.
C. Reactions of primary standard with another substance should be instantaneous and stoichiometric.
D. Primary standard should not be soluble in water.
E. Primary standard should have low relative molar mass.

Choose the correct answer from the options given below :

- (A) D and E Only (B) A, B and E Only
(C) A, B, C and E Only (D) A, B and C Only

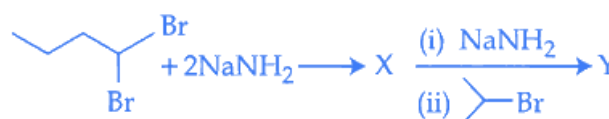
58. Correct statements regarding Arrhenius equation among the following are :

- A. Factor $e^{-E_a/RT}$ corresponds to fraction of molecules having kinetic energy less than E_a .
B. At a given temperature, lower the E_a , faster is the reaction.
C. Increase in temperature by about 10°C doubles the rate of reaction.
D. Plot of $\log k$ vs $\frac{1}{T}$ gives a straight line with slope $= -\frac{E_a}{R}$.

Choose the correct answer from the options given below :

- (A) A and B Only (B) B and D Only
(C) B and C Only (D) A and C Only

59. Consider the following reaction :



The product Y formed is :

- (A) 2-methylhex-3-yne (B) 5-methylhex-2-yne
(C) Isopropylbut-1-yne (D) 2-methylhex-2-yne

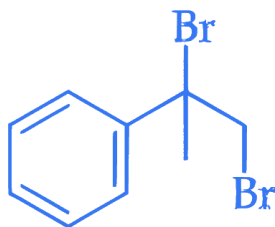
60. $[\text{Ni}(\text{PPh}_3)_2\text{Cl}_2]$ is a paramagnetic complex. Identify the INCORRECT statements about this complex.

- A. The complex exhibits geometrical isomerism.
B. The complex is white in colour.
C. The calculated spin-only magnetic moment of the complex is 2.84 BM.
D. The calculated CFSE (Crystal Field Stabilization Energy) of Ni in this complex is $-0.8\Delta_o$.
E. The geometrical arrangement of ligands in this complex is similar to that in $\text{Ni}(\text{CO})_4$.

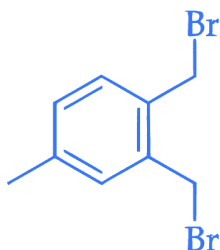
Choose the correct answer from the options given below :

- (A) C and D Only (B) A, B and D Only
(C) C, D and E Only (D) A and B Only

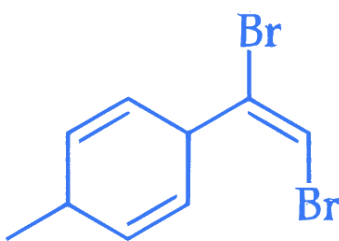
61. The dibromo compound [P] (molecular formula: $C_9H_{10}Br_2$) when heated with excess sodamide followed by treatment with dilute HCl gives [Q]. On warming [Q] with mercuric sulphate and dilute sulphuric acid yield [R] which gives positive Iodoform test but negative Tollen's test. The compound [P] is :



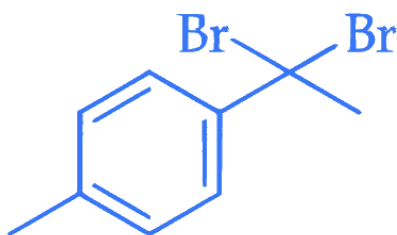
(A)



(B)



(C)



(D)

62. Which of the following mixture gives a buffer solution with $pH = 9.25$?

Given : $pK_b(NH_4OH) = 4.75$

- (A) $0.5MNH_4OH(0.2\text{ L}) + 0.2MHCl(0.5\text{ L})$
 (B) $0.2MNH_4OH(0.5\text{ L}) + 0.1MHCl(0.5\text{ L})$
 (C) $0.2MNH_4OH(0.4\text{ L}) + 0.1MHCl(1\text{ L})$
 (D) $0.4MNH_4OH(1\text{ L}) + 0.1MHCl(1\text{ L})$

63. When 1 g of compound (X) is subjected to Kjeldahl's method for estimation of nitrogen, 15 mL 1 M H_2SO_4 was neutralized by ammonia evolved. The percentage of nitrogen in compound (X) is :

- (A) 0.21 (B) 21
 (C) 42 (D) 0.42

64. The energy of first (lowest) Balmer line of H atom is x J. The energy (in J) of second Balmer line of H atom is :

- (A) $\frac{x}{1.35}$ (B) $1.35x$
 (C) x^2 (D) $2x$

65. Given below are two statements :

Statement I : The first ionization enthalpy of Cr is lower than that of Mn .

Statement II : The second and third ionization enthalpies of Cr are higher than those of Mn .

In the light of the above statements, choose the correct answer from the options given below :

- (A) Statement I is true but Statement II is false
 (B) Both Statement I and Statement II are true
 (C) Both Statement I and Statement II are false
 (D) Statement I is false but Statement II is true

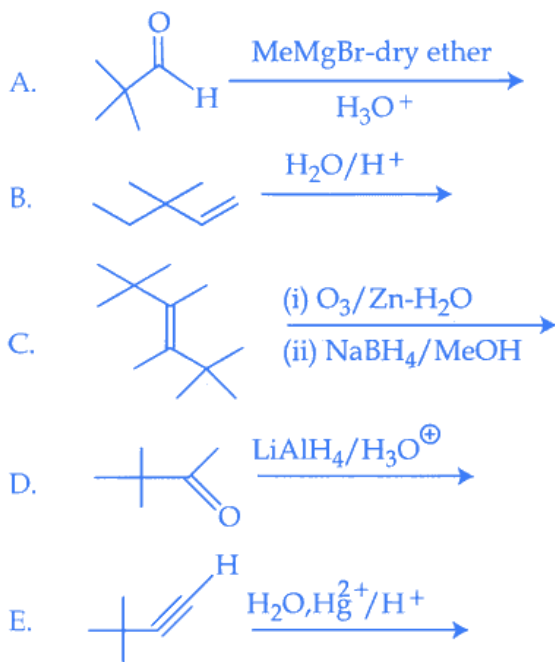
66. At T(K), 100 g of 98% $H_2SO_4(w/w)$ aqueous solution is mixed with 100 g of 49% $H_2SO_4(w/w)$ aqueous solution. What is the mole fraction of H_2SO_4 in the resultant solution?

(Given : Atomic mass H = 1u; S = 32u; O = 16u).
 (Assume that temperature after mixing remains constant)

- (A) 0.9 (B) 0.663
 (C) 0.1 (D) 0.337

67.

3, 3-Dimethyl-2-butanol cannot be prepared by :



Choose the correct answer from the options given below :

- (A) B and E Only (B) B, C and E Only
(C) B and C Only (D) B Only

68. $A + 2B \longrightarrow AB_2$

36.0 g of 'A' (Molar mass : 60 g mol^{-1}) and 56.0 g of 'B' (Molar mass : 80 g mol^{-1}) are allowed to react. Which of the following statements are correct ?

- A. 'A' is the limiting reagent.
B. 77.0 g of AB_2 is formed.
C. Molar mass of AB_2 is 140 g mol^{-1} .
D. 15.0 g of A is left unreacted after the completion of reaction.

Choose the correct answer from the options given below :

- (A) A and C Only (B) A and B Only
(C) C and D Only (D) B and D Only

69. Given below are two statements :

Statement I : Elements 'X' and 'Y' are the most and least electronegative elements, respectively among N, As, Sb and P. The nature of the oxides X_2O_3 and Y_2O_3 is acidic and amphoteric, respectively.

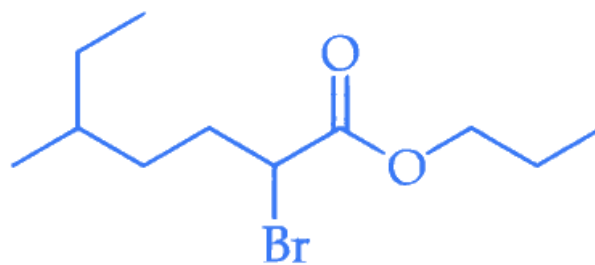
Statement II : BCl_3 is covalent in nature and gets hydrolysed in water. It produces $[B(OH)_4]^-$ and $[B(H_2O)_6]^{3+}$ in aqueous medium.

In the light of the above statements, choose the correct answer from the options given below :

- (A) Statement I is false but Statement II is true (B) Statement I is true but Statement II is false
(C) Both Statement I and Statement II are false (D) Both Statement I and Statement II are true

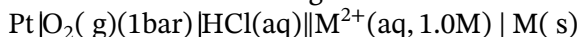
70.

The IUPAC name of the following compound is :



- (A) 2-bromo-5-methylhexylpropanoate
(B) n-propyl-1-bromo-4-methylhexanoate
(C) n-propyl-2-bromo-5-methylheptanoate
(D) 2-bromo-5-methylpropanoate

71. Consider the following electrochemical cell :



The pH above which, oxygen gas would start to evolve at anode is ____ (nearest integer).

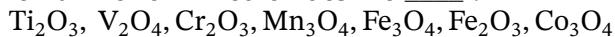
$$\left[\begin{array}{l} \text{Given : } E_{\text{M}^{2+}/\text{M}}^{\circ} = 0.994 \text{ V} \\ E_{\text{O}_2/\text{H}_2\text{O}}^{\circ} = 1.23 \text{ V} \end{array} \right]$$

$$\text{standard reduction potential and } \frac{RT}{F}(2.303) \\ = 0.059 \text{ V at the given condition]}$$

72. The mass of benzanilide obtained from the benzoylation reaction of 5.8 g of aniline, if yield of product is 82%, is ____ g (nearest integer).

(Given molar mass in g mol^{-1} : H : 1, C : 12, N : 14, O : 16)

73. Among the following oxides of 3d elements, the number of mixed oxides are ____ .



74. If the enthalpy of sublimation of Li is 155 kJ mol^{-1} , enthalpy of dissociation of F_2 is 150 kJ mol^{-1} , ionization enthalpy of Li is 520 kJ mol^{-1} , electron gain enthalpy of F is -313 kJ mol^{-1} , standard enthalpy of formation of LiF is -594 kJ mol^{-1} . The magnitude of lattice enthalpy of LiF is ____ kJ mol^{-1} . (Nearest Integer)

75. Consider $\text{A} \xrightarrow{k_1} \text{B}$ and $\text{C} \xrightarrow{k_2} \text{D}$ are two reactions. If the rate constant (k_1) of the $\text{A} \rightarrow \text{B}$ reaction can be expressed by the following equation $\log_{10} k = 14.34 - \frac{1.5 \times 10^4}{T/\text{K}}$ and activation energy of $\text{C} \rightarrow \text{D}$ reaction (E_{a2}) is $\frac{1}{5}$ th of the $\text{A} \rightarrow \text{B}$ reaction (E_{a1}), then the value of (E_{a2}) is ____ kJ mol^{-1} . (Nearest Integer)

- End of Question Paper -