MHT CET 2017
Paper II - Physics and Chemistry

Question Paper

11 May 2017

Time Allowed: 90 Minutes **Maximum Marks: 100**

General Instructions

1. This paper contains 100 objective questions, each with four answer choices.

2. The paper contains 50 Physics questions and 50 Chemistry questions.

3. Choice and sequence for attempting the questions are at the convenience of the candidate.

4. Each correct response earns 1 mark. There is no negative marking.

5. Read each question carefully and select the correct answer.

6. Use a black ink ball point pen to mark answers on the OMR answer sheet.

7. Two or more marked answers, scratching, overwriting or inappropriate marking do not receive marks. Do not use whitener.

8. Use the blank space in the question booklet for rough work, not the answer sheet.

PHYSICS

1. The frequencies for series limit of Balmer and Paschen series respectively are 'v_1' and 'v_3'. If frequency of first line of Balmer series is 'v_2' then the relation between 'v_1', 'v_2' and 'v_3' is

(a) $v_1 - v_2 = v_3$ (b) $v_1 + v_3 = v_2$
(c) $v_1 + v_2 = v_3$ (d) $v_1 - v_3 = 2v_1$

2. When three capacitors of equal capacities are connected in parallel and one of the same capacity is connected in series with its combination. The resultant capacity is 3.75μF. The capacity of each capacitor is

(a) 5 μF (b) 6 μF (c) 7 μF (d) 8 μF

3. Sensitivity of moving coil galvanometer is '*S*'. If shunt of $\frac{1}{8}$ th of the resistance of galvanometer is connected to moving coil galvanometer, its sensitivity becomes

(a) $\frac{S}{3}$ (b) $\frac{S}{6}$ (c) $\frac{S}{9}$ (d) $\frac{S}{12}$

4. Two unknown resistances are connected in two gaps of a meter-bridge. The null point is obtained at 40 cm from left end. A 30 Ω resistance is connected in series with the smaller of the two resistances, the null point shifts by 20 cm to the right end. The value of smaller resistance in Ω is

(a) 12 (b) 24 (c) 36 (d) 48

5. In Fraunhofer diffraction pattern, slit width is 0.2 mm and screen is at 2 m away from the lens. If wavelength of light used is 5000Å, then the distance between the first minimum on either side of the central maximum is (θ is small and measured in radian)

(a) 10^{-1}m (b) 10^{-2}m
(c) 2×10^{-2}m (d) 2×10^{-1}m

6. In series *LCR* circuit R= 18 Ω and impedance is 33 Ω An rms voltage 220V is applied across the circuit. The ture power consumed in AC circuit is

(a) 220w (b) 400w (c) 600w (d) 800w

7. Two parallel plate air capacitors of same capacity C are connected in series to a battery of emf *E*. Then one of the capacitors is completely filled with dielectric material of constant *K*. The change in the effective capacity of the series combination is

(a) $\frac{C}{2}\left[\frac{K-1}{K+1}\right]$ (b) $\frac{2}{C}\left[\frac{K-1}{K+1}\right]$

(c) $\frac{C}{2}\left[\frac{K+1}{K-1}\right]$ (d) $\frac{C}{2}\left[\frac{K-1}{K+1}\right]^2$

8. The polarising angle for transparent medium is 'θ' and 'v' is the speed of light in that medium. Then relation between 'θ' and 'v' is (c = velocity if light in air)

(a) $\theta = \tan^{-1}\left(\frac{v}{c}\right)$ (b) $\theta = \cot^{-1}\left(\frac{v}{c}\right)$

(c) $\theta = \sin^{-1}\left(\frac{v}{c}\right)$ (d) $\theta = \cos^{-1}\left(\frac{v}{c}\right)$

9. Two identical light waves having phase difference 'ϕ' propagate in same direction.When they superpose, the intensity of resultant wave is proportional to

(a) $\cos^2\phi$ (b) $\cos^2\left(\frac{\phi}{2}\right)$

(c) $\cos^2\left(\frac{\phi}{3}\right)$ (d) $\cos^2\left(\frac{\phi}{4}\right)$

10. For a transistor, α_{dc} and β_{dc} are the current ratios, then the value of $\frac{\beta_{dc}-\alpha_{dc}}{\alpha_{dc}\cdot\beta_{dc}}$

(a) 1 (b) 1.5 (c) 2 (d) 2.5

11. A radioactive element has rate of disintegration 10,000 disintegrations per minute at a particular instant. After four minutes it become 2500 disintegrations per minute. The decay constant per minute is

(a) $0.2\log_e 2$ (b) $0.5\log_e 2$
(c) $0.6\log_e 2$ (d) $0.8\log_e 2$

12. When the same monochromatic ray of light travels through glass slab and through water, the number of waves in glass slab of thickness 6cm is same as in water column of height 7cm. If refractive index of glass is 1.5, then refractive index of water is

(a) 1.258 (b) 1.269 (c) 1.286 (d) 1.310

13. If the electron in hydrogen atom jumps from second Bohr orbit to ground state and difference between energies of the two states is radiated in the form of photons. If the work function of the material is 4.2 eV, then stopping potential is

[Energy of electron in nth orbit $= -\frac{13.6}{n^2}$ eV]

(a) 2V (b) 4V (c) 6V (d) 8V

14. The magnetic moment of electron due to orbital motion is proportional to
(n = principal quantum numbers)

(a) $\frac{1}{n^2}$ (b) $\frac{1}{n}$ (c) n^2 (d) n

15. Photodiode is a device

(a) which is always operated in reverse bias
(b) which is always operated in forward bias
(c) in which photo curent is independent of intensity of incident radiation
(d) which may be operated in forwad or revese bias.

16. A wheel of moment of inertia 2 kg m^2 is rotating about an axis passing through centre and perpendicular to its plane at a speed 60 rad/s. Due to friction, it comes to rest in 5 minutes. The angular momentum of the wheel three minutes before it stops rotating is

(a) 24 kg m^2/s (b) 48 kg m^2/s
(c) 72 kg m^2/s (d) 96 kg m^2/s

17. The equation of the progressive wave is $y = 3\sin\left[\pi\left(\frac{t}{3}-\frac{x}{5}\right)+\frac{\pi}{4}\right]$, where x and y are in metre and time in second. Which of the following is correct?

(a) Velocity v = 1.5 m/s
(b) Amplitude A = 3cm
(c) Frequency f = 0.2Hz
(d) Wavelength λ = 10m

18. Two spherical black bodies have radii 'r_1' and 'r_2'. Their surface temperature are 'T_1' and 'T_2'. If they radiate same power, then $\frac{r_2}{r_1}$ is

(a) $\frac{T_1}{T_2}$ (b) $\frac{T_2}{T_1}$

(c) $\left(\frac{T_1}{T_2}\right)^2$ (d) $\left(\frac{T_2}{T_1}\right)^2$

19. The closed and open organ pipes have same length. When they are vibrating simultaneously in first overtone, produce three beats. The lenght of open pipe is made $\frac{1}{3}$rd and closed pipe is made three times the original, the number of beats produced will be

(a) 8 (b) 14 (c) 17 (d) 20

20. A lift of mass 'm' is connected to a rope which is moving upward with maximum acceleration 'a'. For maximum safe stress, the elastic limit of the rope is 'T'. The minimum diameter of the rope is (g = gravitational acceleration)

(a) $\left[\frac{2m(g+a)}{\pi T}\right]^{\frac{1}{2}}$ (b) $\left[\frac{4m(g+a)}{\pi T}\right]^{\frac{1}{2}}$

(c) $\left[\frac{m(g+a)}{\pi T}\right]^{\frac{1}{2}}$ (d) $\left[\frac{m(g+a)}{2\pi T}\right]^{\frac{1}{2}}$

21. A solid sphere of mass 2 kg is rolling on a frictionless horizontal surface with velocity 6m/s. It collides on the free end of an ideal spring whose other end is fixed. The maximum compression produced in the spring will be (Force constant of the spring = 36 N/m)

(a) $\sqrt{14}$ m (b) $\sqrt{2.8}$ m
(c) $\sqrt{1.4}$ m (d) $\sqrt{0.7}$ m

22. A flywheel at rest is to reach an angular velocity of 24rad/s in 8 second with constant angular acceleration. The total angle turned through during this interval is

(a) 24 rad (b) 48 rad (c) 72 rad (d) 96 rad

23. Two uniform wires of the same material are vibrating under the same tension. If the first overtone of the first wire is equal to the second overtone of the second wire and radius of the first wire is the twice the radius of the second wire, then the ratio of the lengths of the first wire to second wire is

(a) $\frac{1}{3}$ (b) $\frac{1}{4}$ (c) $\frac{1}{5}$ (d) $\frac{1}{6}$

24. When one end of the capillary is dipped in water, the height of water column is 'h'. The upward force of 105 dyne due to surface tension is balanced by the force due to the weight of water column. The inner circumference of capillary is (Surface tension of water = 7×10^{-2} N/m)

(a) 1.5 cm (b) 2 cm (c) 2.5 cm (d) 3 cm

25. For a rigid diatomic molecule, universal gas constant $R = nC_p$, where 'C_p' is the molar specific heat at constant pressure and 'n' is a number. Hence 'n' is equal to

(a) 0.2257 (b) 0.4 (c) 0.2857 (d) 0.3557

26. An ideal gas has pressure 'p' volume 'V' and absolute temperature 'T'. If 'm' is the mass of each molecules and 'K' is the Boltzmann constant, then density of the gas is

(a) $\frac{pm}{KT}$ (b) $\frac{KT}{pm}$ (c) $\frac{Km}{pT}$ (d) $\frac{pK}{Tm}$

27. A big water drop is formed by the combination of 'n' small water drops of equal radii. The ratio of the surface energy of 'n'drops to the surface energy of big drop is

(a) $n^2 : 1$ (b) $n : 1$ (c) $\sqrt{n} : 1$ (d) $\sqrt[3]{n} : 1$

28. The ratio of binding energy of a satellite at rest on earth's surface to the binding energy of a satellite of same mass revolving around the earth at a height h above the earth's surface is (R = radius of the earth).

(a) $\frac{2(R+h)}{R}$ (b) $\frac{R+h}{2R}$
(c) $\frac{R+h}{R}$ (d) $\frac{R}{R+h}$

29. A particle performing SHM starts equilibrium position and its time period is 16 seconds. After 2 seconds its velocity is π m/s. Amplitude of oscillation is $\left(\cos 45° = \frac{1}{\sqrt{2}}\right)$

(a) $2\sqrt{2}$m (b) $4\sqrt{2}$m
(c) $6\sqrt{2}$m (d) $8\sqrt{2}$m

30. In sonometer experiment, the string of length 'L' under tension vibrates in second overtone between two bridges. The amplitude of vibration is maximum at

(a) $\frac{L}{3}, \frac{2L}{3}, \frac{5L}{6}$ (b) $\frac{L}{8}, \frac{L}{4}, \frac{L}{2}$
(c) $\frac{L}{2}, \frac{L}{4}, \frac{L}{6}$ (d) $\frac{L}{6}, \frac{L}{2}, \frac{5L}{6}$

31. The depth 'd' at which the value of acceleration due to gravity becomes $\frac{1}{n}$ times the value at the earth's surface is (R = radius of earth)

(a) $d = R\left(\frac{n}{n-1}\right)$ (b) $d = R\left(\frac{n-1}{2n}\right)$
(c) $d = R\left(\frac{n-1}{n}\right)$ (d) $d = R^2\left(\frac{n-1}{n}\right)$

32. A particle is performing SHM starting extreme position, graphical representation shows that between displacement and acceleration there is a phase difference of

(a) 0 rad (b) $\frac{\pi}{4}$ rad
(c) $\frac{\pi}{2}$ rad (d) π rad

33. The fundamental frequency of an air column is a pipe closed at one end is 100 Hz. If the same pipe is open at both the ends, the frequencies produced in Hz are

(a) 100, 200, 300, 400...
(b) 100, 300, 500, 700...
(c) 200, 300, 400, 500...
(d) 200, 400, 600, 800

34. For a particle moving in vertical circle, the total energy at different positions along the path
(a) is conserved
(b) increases
(c) decreases
(d) may increase or decrease

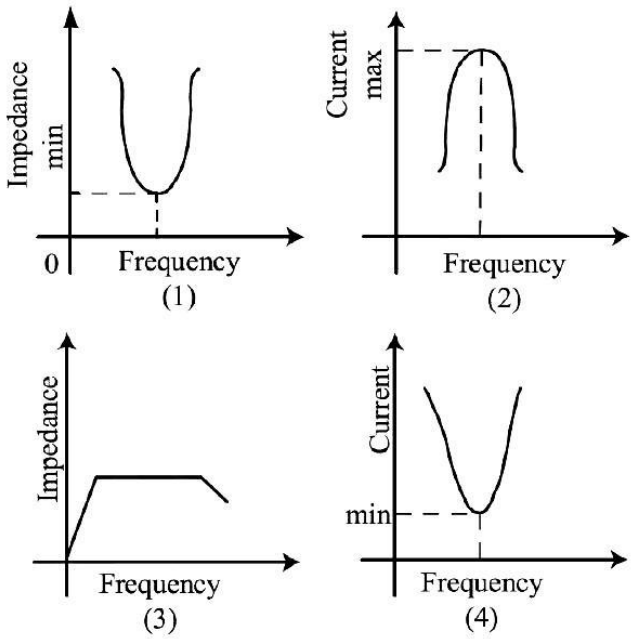
35. A simple pendulum of length 'L' has mass 'M' and it oscillates freely with amplitude 'A'. At extreme position, its potential energy is (g=acceleration due to gravity)

(a) $\frac{MgA^2}{2L}$ (b) $\frac{MgA}{2L}$

(c) $\frac{MgA^2}{L}$ (d) $\frac{2MgA^2}{L}$

36. On a photosensitive material, when frequency of incident radiation is increased by 30%, kinetic energy of emitted photoelectrons increases from 0.4 eV to 0.9 eV. The work function of the surface is
(a) 1 eV (b) 1.267 eV
(c) 1.4 eV (d) 1.8 eV

37. Out of the following graphs, which graphs shows the correct relation (graphical representation) for LC parallel resonant circuit



(a) 1 (b) 2 (c) 3 (d) 4

38. According to de-Broglie hypothesis, the wavelength associated with moving electron of mass 'm' is 'λ_e'. Using mass energy relation and Planck's quantum theory, the wavelength associated with photon is 'λ_p'. If the energy (E) of electron and photon is same, then relation between 'λ_e' and 'λ_p' is

(a) $\lambda_p \propto \lambda_e$ (b) $\lambda_p \propto \lambda_e^2$

(c) $\lambda_p \propto \sqrt{\lambda_e}$ (d) $\lambda_p \propto \frac{1}{\lambda_e}$

39. A parallel plate air capacity 'C' farad, potential 'V' volt and energy 'E' joule. When the gap between the plastes is completely filled with dielectric
(a) both V and E increase
(b) both V and E decrease
(c) V decrease, E increases
(d) V increases, E decrease

40. The resistivity of potentiometer wire is 40×10^{-8} ohm-metre and its area of cross- section is $8\times10^{-6}m^2$. If 0.2 ampere current is flowing through the wire, then the potential gradient of the wire is
(a) 10^{-1} V/m (b) 10^{-2} V/m
(c) 10^{-3} V/m (d) 10^{-4} V/m

41. A ceiling fan rotates about its own axis with some angular velocity. When the fan is switched off the angular velocity becomes $\left(\frac{1}{4}\right)$th of the original in time 't' and 'n' revolutions are made in that time. The number of revolutions made by the fan during the time interval between switch off and rest are (Angular retardation is uniform)

(a) $\frac{4n}{15}$ (b) $\frac{8n}{15}$ (c) $\frac{16n}{15}$ (d) $\frac{32n}{15}$

42. A disc of moment of inertia 'I_1' is rotating in horizontal plane about an axis passing through a centre and perpendicular to its plane with constant angular speed 'ω_1'. Another disc of moment of inertia 'I_2' having zero angular speed is placed co-axially on a rotating disc. Now, both the discs are rotating with constant angular speed 'ω_2'. The energy lost by the initial rotating disc is

(a) $\frac{1}{2}\left[\frac{I_1+I_2}{I_1I_2}\right]\omega_1^2$ (b) $\frac{1}{2}\left[\frac{I_1I_2}{I_1-I_2}\right]\omega_1^2$

(c) $\frac{1}{2}\left[\frac{I_1-I_2}{I_1I_2}\right]\omega_1^2$ (d) $\frac{1}{2}\left[\frac{I_1I_2}{I_1+I_2}\right]\omega_1^2$

43. A particle performs linear SHM at a particular instant, velocity of the particle is 'u' and acceleration is α while at another instant velocity is 'v' and acceleration is 'β' ($0 < \alpha < \beta$). The distance between the two position is

(a) $\frac{u^2-v^2}{\alpha+\beta}$ (b) $\frac{u^2+v^2}{\alpha+\beta}$

(c) $\frac{u^2-v^2}{\alpha-\beta}$ (d) $\frac{u^2+v^2}{\alpha-\beta}$

44. The observer is moving with velocity 'v_0' towards the stationary source of sound and then after crossing moves away from the source with velocity 'v_0'. Assume that the medium through which the sound waves travel is at rest. If v is the velocity of sound and n is the frequency emitted by the source, then the difference between apparent frequencies heard by the observer is

(a) $\frac{2nv_0}{v}$ (b) $\frac{nv_0}{v}$ (c) $\frac{v}{2nv_0}$ (d) $\frac{v}{nv_0}$

45. A metal rod of length L and cross-sectional area A is heated through T °C. What is the force required to prevent the expansion of the rod lengthwise?
[Y = Young's modulus of the material of rod, α = coefficient of linear expansion]

(a) $\frac{YA\alpha T}{1-\alpha T}$ (b) $\frac{YA\alpha T}{1+\alpha T}$
(c) $\frac{1-\alpha T}{YA\alpha T}$ (d) $\frac{1+\alpha T}{YA\alpha T}$

46. Two coils P and Q are kept near each other. When no current flows through coil P and current increase in coil Q at the rate 10A/s, the emf in coil P is 15mV. When coil Q carries no current and current of 1.8A flows through coil P, the magnetic flux linked with the coil Q is
(a) 1.4 m Wb (b) 2.2 mWb
(c) 2.7 m Wb (d) 2.9 m Wb

47. In Young's double experiment, in air interference pattern second minimum is observed exactly in front of one slit. The distance between the two coherent source is 'd' and the distance between source and screen is 'D'. The wavelength of light source used is

(a) $\frac{d^2}{D}$ (b) $\frac{d^2}{2D}$ (c) $\frac{d^2}{3D}$ (d) $\frac{d^2}{4D}$

48. In communication system, the process of superimposing a low frequency signal on a high frequency wave is know n as
(a) repeater (b) attenuation
(c) modulation (d) demodulation

49. A bar magnet has length 3 cm, cross-sectional area 2 cm^2 and magnetic moment 3 A m^2. The intensity of magnetisation of bar magnet is
(a) 2×10^5 A/m (b) 3×10^5 A/m
(c) 4×10^5 A/m (d) 5×10^5 A/m

50. The magnetic flux near the axis and inside the air core solenoid of length 60cm carrying current 'I' is 1.57×10^{-6} Wb. Its magnetic moment will be (cross-sectional area of a solenoid is very small as compared to its length, $\mu_0 = 4\pi \times 10^{-7}$ SI unit)
(a) 0.25A (b) 0.50A (c) 0.75A (d) 1 A

CHEMISTRY

51. The work done during combustion of 9×10^{-2} kg of ethane, C_2H_6 (g) at 300 K is
(Given R = 8.314 J deg^{-1}, atomic mass C = 12, H = 1
(a) 6.236 kJ (b) −6.236 kJ
(c) 18.71 kJ (d) −18.71 kJ

52. What type of sugar molecule is present is DNA?
(a) D-3-deoxyribose (b) D-ribose
(c) D-2-deoxyribose (d) D-glucopyranose

53. The molality of solution containing 15.20 g of urea, (molar mass = 60) dissolved in 150 g or water is
(a) 1.689 mol kg^{-1} (b) 0.1689 mol kg^{-1}
(c) 0.5922 mol kg^{-1} (d) 0.2533 mol kg^{-1}

54. The acid, which contains both –OH and –COOH groups is
(a) phthalic acid (b) adipic acid
(c) glutaric acid (d) salicylic acid

55. Identify the compound, in which phosphorus exists in the oxidation state of +1.
(a) Phosphonic acid (H_3PO_3)
(b) Phosphinic acid (H_3PO_2)
(c) Pyrophosphorus acid ($H_4P_2O_5$)
(d) Orthophosphoric acid (H_3PO_4)

56. Identify the weakest oxidising agent among the following.
(a) Li^+ (b) Na^+ (c) Cd^{2+} (d) I_2

57. The monomers used in preparation of dextron are
(a) lactic acid and glycolic acid
(b) 3-hydroxy butanoic acid and 3-hydroxy pentanoic acid
(c) styrene and 1, 3-butadiene
(d) hexamethylenediamine and adipic acid

58. Which among the following compounds does not act as reducing agent?
(a) H_2O (b) H_2S (c) H_2Se (d) H_2Te

59. Which of the following processes is not used to preserve the food?
(a) Irradiation (b) Addition of salts
(c) Addition of heat (d) Hydration

60. In case of substituted aniline the group which decreases the basic strength is
(a) $-OCH_3$ (b) $-CH_3$
(c) $-NH_2$ (d) $-C_6H_5$

61. (+) 2-methylbutan-1-ol and (−) 2-methylbutan-1-ol have different values for which property?
(a) Boiling point (b) Relative density
(c) Refractive index (d) Specific rotation

62. Which among the following is not is mineral of iron?
(a) Haematite (b) Magnesite
(c) Magnetite (d) Siderite

63. Nitration of which among the following compounds yields cyclonite?
(a) Formaldehyde
(b) Benzaldehyde
(c) Urotropine
(d) Acetaldehyde ammonia

64. Calculate the work done during compression of 2 mol of an ideal gas from a volume of $1m^3$ to $10\ dm^3$ 300K against a pressure of 100 KPa.
(a) –99 kJ (b) +99 kJ
(c) +22.98 kJ (d) –22.98 kJ

65. Which element among the following does form $p\pi - p\pi$ multiple bonds?
(a) Arsenic (b) Nitrogen
(c) Phosphorus (d) Antimony

66. Which of the following statement(s) is/are incorrect in case of Hofmann bromamide degradation?
(a) Reaction is useful for decreasing length of carbon chain by one carbon atom
(b) It gives tertiary amine
(c) It gives primary amine
(d) Aqueous or alco. KOH is used with bromine

67. Which of the following statements is INCORRECT for pair of elements Zr–Hf?
(a) Both possess same number of valence electrons.
(b) Both have identical atomic sizes.
(c) Both have almost identical ionic radii.
(d) Both of these belong to same period of periodic table.

68. Aldehyde or ketones when treated with $C_6H_5 - NH–NH_2$, the product formed is
(a) semicarbazone (b) phenlhydrazone
(c) hydrazone (d) oxime

69. Solubility of which among the following solids in water changes slightly with temperature?
(a) KNO_3 (b) $NaNO_3$
(c) KBr (d) NaBr

70. What is the quantity of hydrogen gas liberated when 46 g of sodium reacts with excess ethanol? (Given atomic mass of Na = 23)
(a) 2.4×10^{-3} kg (b) 2.0×10^{-3} kg
(c) 4.0×10^{-3} kg (d) 2.4×10^{-2} kg

71. *Tert*-butyl methyl ether on treatment with hydrogen iodide in cold gives
(a) *tert*-butyl iodide and methyl iodide
(b) *tert*-butyl alcohol and methyl alcohol
(c) *tert*-butyl alcohol and methyl iodide
(d) *tert*-butyl iodide and methyl alcohol

72. Name the process that is employed to refine aluminium.
(a) Hall's process (b) Mond process
(c) Hoope's process (d) Serperck's process

73. The colour and magnetic nature of manganate ion (MnO_4^{2-}) is
(a) green, paramagnetic
(b) purple, diamagnetic
(c) green, diamagnetic
(d) purple, paramagnetic

74. The osmotic pressure of solution containing 34.2 g of cane sugar (molar mass = 342 g mol^{-1}) in 1L of solution at 20°C is (Given R = 0.082 L atm K^{-1} mol^{-1})
(a) 2.40 atm (b) 3.6 atm
(c) 24 atm (d) 0.0024 atm

75. In assigning R-S configuration, which among the following groups has highest priority?
(a) $–SO_3H$ (b) –COOH
(c) –CHO (d) $–C_6H_5$

76. Which of the following is used as antiseptic?
(a) Chloramphenicol (b) Bithional
(c) Cimetidine (d) Chlordiazepoxide

77. In preparation of sulphuric acid from sulphur dioxide in lead chamber process. What substance is used as a catalyst?
(a) Manganese dioxide
(b) Vanadium pentoxide
(c) Nitric oxide
(d) Raney nickel

78. The correct charge on and co-ordination number of 'Fe' in $K_3[Fe(CN)_6]$ is
(a) +2, 4 (b) +3, 6 (c) +2, 6 (d) +3, 3

79. Which among the following reactions is an example of pseudo first order reaction?
(a) Inversion of cane sugar
(b) Decomposition of H_2O_2
(c) Conversion of cyclopropane to propene
(d) Decomposition of N_2O_5

80. The amine, which reacts with *p*-toluenesulphonyl chloride to give a clear solution, which on acidification gives insoluble compound is
(a) $C_2H_5NH_2$ (b) $(C_2H_5)_2NH$
(c) $(C_2H_5)_3N$ (d) $CH_3NHC_2H_5$

81. Which among the following equation represents Arrhenius equation?
(a) $k = A_e^{\frac{E_a}{RT}}$ (b) $k = A.e^{\frac{RT}{E_a}}$
(c) $k = \frac{A}{e^{E_a/RT}}$ (d) $k = \frac{A}{e^{RT/E_a}}$

82. Which of the following compound will give positive iodoform test?
(a) Isopropyl alcohol
(b) Propionaldehyde
(c) Ethylphenyl ketone
(d) Benzyl alcohol

83. The first law of thermodynamics for isothermal process is
(a) $q = -W$ (b) $\Delta U = W$
(c) $\Delta U = q_v$ (d) $\Delta U = -q_v$

84. The conversion of ethyl bromide using sodium iodide and dry acetone, this reaction is know as
(a) Swarts reaction
(b) Finkelstein reaction
(c) Sandmeyer reaction
(d) Stephen reaction

85. What is the hybridisation of carbon atoms in fullerene?
(a) sp^3 (b) sp (c) sp^2 (d) dsp^3

86. What is the SI unit of conductivity?
(a) Sm (b) Sm^{-1} (c) Sm^2 (d) Sm^{-2}

87. Which of the following in Baeyer's reagent?
(a) Alkaline $KMnO_4$ (b) Acidic $K_2Cr_2O_7$
(c) Alkaline $Na_2Cr_2O_7$ (d) MnO_2

88. What is the chief constituent of pyrex glass?
(a) B_2O_3 (b) SiO_2 (c) Al_2O_3 (d) Na_2O

89. Which of the following compounds has lowest boiling point?
(a) *n*-butyl alcohol (b) Iso-butyl alcohol
(c) *Tert*-butyl alcohol (d) *Sec*-butyl alcohol

90. Identify the invalid equation
(a) $\Delta H = \Sigma H_{products} - \Sigma H_{reactants}$
(b) $\Delta H = \Delta U + p\Delta V$
(c) $\Delta H^\circ_{(reaction)} = \Sigma H^\circ_{(product\ bonds)} - \Sigma H^\circ_{(reactant\ bonds)}$
(d) $\Delta H = \Delta U + \Delta nRT$

91. The rate constant for a first order reaction is $7.0 \times 10^{-4}\ s^{-1}$. If initial concentration of reactant is 0.080 M, what is the half life of reaction?
(a) 990 s (b) 79.2 s
(c) 12375 s (d) 10.10×10^{-4} s

92. The polymer used in making handles of cookers and frying pans is
(a) bakelite (b) nylon-2-nylon-6
(c) orlon (d) Polyvinyl chloride

93. Which halogen has the highest value of negative electron gain enthalpy?
(a) Fluorine (b) Chlorine
(c) Bromine (d) Iodine

94. What is the actual volume occupied by water molecules present in 20 cm^3 of water?
(a) 20 cm^3 (b) 10 cm^3
(c) 40 cm^3 (d) 24.89 dm^3

95. Which of the following co-ordinate complexes if an exception to EAN rule? (Given atomic number Pt = 78, Fe = 26, Zn = 30, Cu = 29)
(a) $[Pt(NH_3)_6]^{4+}$ (b) $[Fe(CN)_6]^{4-}$
(c) $[Zn(NH_3)_4]^{2+}$ (d) $[Cu(NH_3)_4]^{2+}$

96. Which among the following equations represents the reduction reaction taking place in lead accumulator at positive electrode, while it is being used as a source of electrical energy?
(a) $Pb \rightarrow Pb^{2+}$ (b) $Pb^{4+} \rightarrow Pb$
(c) $Pb^{2+} \rightarrow Pb$ (d) $Pb^{4+} \rightarrow Pb^{2+}$

97. For which among the following equimolar aqueous solutions Van't Hoff factor has the lowest value?
(a) Aluminium chloride
(b) Potassium sulphate
(c) Ammonium chloride
(d) Urea

98. The amino acid, which is basic in nature is
(a) histidine (b) tyrosine
(c) proline (d) valine

99. Which element among the following does not form diatomic molecules?
(a) Argon (b) Oxygen
(c) Nitrogen (d) Bromine

100. A molecule of stachyose contains how many carbon atoms?
(a) 6 (b) 12 (c) 18 (d) 24